

MODELLING OF EMULSION POLYMERIZATION IN A CONTINUOUS STIRRED TANK REACTOR

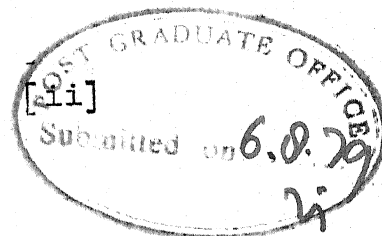
A Thesis Submitted
In Partial Fulfilment of the Requirements
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By

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to the

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INDIAN INSTITUTE OF TECHNOLOGY, KANPUR
AUGUST, 1979**



CERTIFICATE

This is to certify that the present work 'MODELLING OF EMULSION POLYMERIZATION IN A CONTINUOUS STIRRED TANK REACTOR' has been carried out under my supervision and has not been submitted elsewhere for a degree.

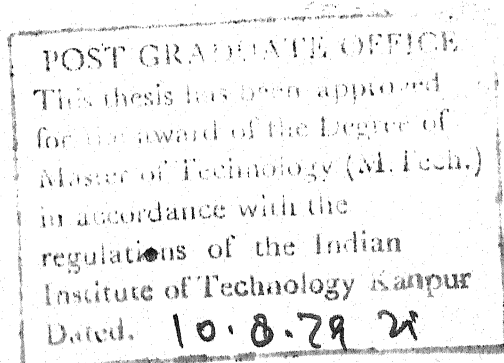
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ABSTRACT

A general modelling framework was suggested by Min and Ray [17] for emulsion polymerization reactions. A simplified model based on the framework has been proposed to simulate the emulsion polymerization reaction in a continuous stirred tank reactor. The model uses the population balance equation approach. The method of moments has been used to solve the population balance equations. The model is capable of calculating conversion, rate of polymerization, number of particles, particle size distribution and the molecular weight distribution of resultant polymer latex. The variables studied were residence time and initiator concentration. The results obtained using this model were compared with the experimental data and the theoretical results of DeGraff and Poehlein [21]. The proposed model is found to predict the aforesaid quantities well at low residence times. The predictions of the model are in moderate agreement with data at high residence times. The theoretical approach by DeGraff and Poehlein behaves in the opposite way.

CHAPTER 1

INTRODUCTION

The 50 year old technique of emulsion polymerization is still being widely used in polymer industry. It often has significant advantages over homogeneous polymerization. Its final product is a latex and hence possesses the obvious advantage of direct application in cases like paints and coatings. Apart from this, there are distinct advantages in operating the reactor. The viscosity of a polymer latex in an emulsion polymerization reactor is low and hence the temperature control is relatively easy. In addition, a considerably high molecular weight product can be obtained at much higher reaction rates. For these reasons, emulsion polymerization has become commercially important in polymer industry. At present it is a major processing step in the manufacture of polymer products such as synthetic rubbers, paints, inks, coatings, adhesives, high impact strength co-polymers etc.

The commercial technique of emulsion polymerization can be divided into two classes: (1) batch and (2) continuous.

Batch systems are relatively simple and were developed first. These systems, although less efficient, are used for low production rates and for speciality applications. The advantages of this technique are the relatively simple equipment to be handled and the fact that a small amount

of material can be easily processed. This type of polymerization has been extensively studied in literature. Numerous articles are available in literature regarding the experimental and theoretical aspects of emulsion polymerization in batch. The more important of these works are by Smith and Ewart [1], Stockmayer [2], O'Toole [3], Gardon [4-10], Ugelstad [11], Medvedev [12], Min and Ray [13]. These and many other works have been comprehensively reviewed by Alexander and Nappel [14], Vander Hoff [15,16], Min and Ray [17].

As the annual output of the polymeric materials increases, more and more attention is being given to the use of continuous reactors. Although continuous stirred tank reactor (CSTR) is widely used commercially for emulsion polymerization, very little published information is available about these reactors. Experimental data is limited since the equipments are more costly and less flexible, experimental run times are longer and larger amounts of raw materials are consumed. Similarly the modelling of these reactors involves some conceptual problems which are not present in batch systems.

The earliest of the models developed for emulsion polymerization in a CSTR is by Greshberg and Longfield [18]. They have employed a residence time distribution model coupled with the Smith-Ewart kinetics. Omi and co-workers [19] and Nomura and co-workers [20] have employed Smith-Ewart kinetics for different reaction systems. They have also used

different models for initiation of particles. DeGraff and Poehlein [21] have employed Stockmayer's [2] theory coupled with residence time distribution for modelling a single stirred tank reactor. Recently population balance approach has been employed for modelling of continuous reactors.

Stevens and Funderburk [22] have employed a simple population balance approach to model a CSTR. Further work using this approach was done by Thompson and Stevens [23], Thompson and co-workers [24] and Cauley and Thompson [25]. These authors have used population balance equations to generate radical number distribution. They have used size independent particle growth rate. Cauley and Thompson [25] have further assumed instantaneous termination of radicals inside the particles. They have not developed material balance equations for various reactive species. The conversion and molecular weights of the polymer could not be calculated.

Min and Ray [17] have developed a comprehensive model for emulsion polymerization in batch and CSTR. They have taken into consideration a number of mechanisms like desorption of radicals from particles, chain transfer, initiation in aqueous phase, end-group stabilization of polymer particles, initiation by precipitation etc. They have, however, not presented any numerical result from this comprehensive model for CSTR.

The latest work on modelling of CSTR is by Kirilov and Ray [26]. They have used the general model of Min and Ray [17] for a specific case by making a number of simplifications. They have employed simple population balance equation for particle size distribution and have used Stockmayer's [2] relation for calculating the average number of radicals per particle. They have calculated average particle size and conversion obtained in the reactor. They did not calculate either the particle size distribution or the molecular weights of the polymer product.

In the present work, an attempt has been made to develop and solve modelling equations for emulsion polymerization in a CSTR. The population balance approach has been used. The population balance equations employed are more comprehensive. As a result, the molecular weight distribution of the resultant product can also be generated in addition to conversion and the particle size distribution.

CHAPTER 2

PHYSICAL MODEL AND POPULATION BALANCE EQUATIONS

Before starting the actual formulation of modelling equations, it is necessary to understand the physical picture and details of various mechanisms operating in an emulsion polymerization reactor. The currently accepted [14-17] picture is as follows:

A typical emulsion polymerization recipe consists of 100 parts of water, 50 parts of monomer, 2 parts of emulsifier, 0.1 part of initiator and small amounts of chain transfer agents and other additives. The soap is present in micelles, on the surface of monomer droplets and in the aqueous phase. The monomer is present mainly as droplets and a small amount (1 per cent) is solubilized in the micellar soap. A small portion of the monomer may also be dissolved in the aqueous phase. Generally, the initiator is present in the aqueous phase. In early stages of the reaction, the system consists of 3 types of particles dispersed in continuous aqueous phase: (1) emulsified monomer droplets of size 10000 \AA and at a concentration of $10^{12}/\text{ml}$, (2) soap micelles with size $50-100 \text{ \AA}$ and at a concentration of $10^{18}/\text{ml}$ and (3) emulsifier stabilized monomer swollen polymer particles of size $500-1000 \text{ \AA}$, and at a concentration of $10^{14}-10^{15}/\text{ml}$ in later stages.

As the reaction proceeds the polymer particles grow in size, consuming the monomer present in the whole system, thus requiring more and more amount of soap to stabilize them. The soap is supplied by micelles which decrease in number and finally disappear. Similarly, the monomer droplets shrink in size as they supply monomer to the polymer particles and finally disappear, leaving the reaction system consisting of polymer particles dispersed in the aqueous phase.

Now some of the mechanisms operating in the reaction mixture will be discussed.

Nucleation of polymer particles is one of the least understood phenomena. It can occur in two different ways. The initiator present in the aqueous phase thermally decomposes into constituent radicals. (reaction temperature approximately $50-70^{\circ}\text{C}$) and these radicals grow by adding the monomer units from the monomer dissolved in the aqueous phase. During this process the radicals and the active polymer chains exhibit Brownian motion. During their motion they can collide with micelles, get absorbed into them and start the polymerization reaction thereby converting the micelles into polymer particles. They can also collide with and get absorbed in existing polymer particles. The radicals which are not captured by micelles or particles continue to add on monomer units and in the process decrease their

aqueous phase solubility. When a critical limit of chain size is reached they will precipitate and initiate new particles. Both these mechanisms of particle nucleation operate simultaneously. The former is more important for monomer scarcely soluble in water, while the latter is predominant for monomers which are relatively more soluble in water.

The particles thus initiated consist of monomer, polymer chains and the live polymer radicals. Polymerization continues in the particles, the monomer being supplied by monomer droplets. Radicals absorb into and desorb from the particles. Propagation, termination as well as chain transfer can take place inside particles. Soap required for stabilization is supplied by the micelles until they disappear. As more and more amount of monomer is consumed the monomer droplets decrease in size and number. Ultimately they disappear leaving only the polymer particles containing live radicals and dissolved monomer. As the reaction proceeds to higher range of conversion the monomer concentration in a particle decreases, viscosity of reaction medium increases and the reaction can become diffusion controlled. Gel effect may play an important role at higher conversion. Apart from these mechanisms, particles can also coagulate with each other to a limited extent. The extent of coagulation depends upon the particle size, amount of soap adsorbed on each particle, reaction temperature etc.

The polymer particles consist of monomer, polymer chains and live radicals as stated earlier. However, the exact structure of these particles is not yet known. The monomer and polymer may solubilize each other to form a homogeneous mixture. If that is the case, reaction will occur at a constant rate throughout the homogeneous particle. On the other hand, Williams [27] has postulated a non-uniform morphology for the particles, comprising of a polymer rich core and a monomer rich shell. The reaction is assumed to take place on the growing monomer polymer interface. Naper[28] has suggested exactly the opposite structure consisting of a monomer with core and a polymer rich shell.

This, in short, is a physical picture of the emulsion polymerization process. The aforesaid mechanisms operate simultaneously in a batch or continuous emulsion polymerization reactor. The objective of the present work is to use the knowledge of these basic mechanisms to model the characteristics of a polymer latex produced in a continuous stirred tank reactor.

The problem can be approached in two different ways:

In the residence time distribution approach the segregated flow model of a CSTR is employed. In this model the reactor is assumed to consist of a number of small independent reaction units. Each reaction unit is assumed to reside for different times inside the reactor. The reaction

units are assumed to operate as isolated batch reactors. The residence time distribution of different units combined with the knowledge of batch behaviour is believed to represent the CSTR characteristics. This method was most recently used by DeGraff and Poehlein [21] for the modelling of emulsion polymerization CSTR. Here a particle was identified as a reaction unit. They assumed the usual exponential residence time distribution. Another approach for modelling is the use of population balance equations which will be employed in this work. A short discussion about the theory of population balance equations is available in ref. 29. A brief introduction to this method is given below.

For any particulate system, a particle phase space consisting of a minimum number, say p , of independent co-ordinates, attached to a particle distribution, which completely describes the properties of the distribution, is postulated. The particle phase space may be divided into two subregions consisting of m internal and three external co-ordinates. External co-ordinates refer to the three spatial co-ordinates of the particle. Internal co-ordinates quantitatively represent the state of the particle independent of its spatial position.

Let us consider a $(m+3)$ dimensional particle distribution function $n(Z,t)$ defined over region Z consisting of three external co-ordinates and m independent internal

co-ordinates. In a dynamic particulate system, individual particles change their position in the particle phase space. If the change of particle co-ordinates is gradual and continuous in any direction the rate of such 'motion' is called as convective particle velocity along that co-ordinate axis. Thus the particle phase-space velocity vector is defined as

$$\begin{aligned}\vec{v} &= v_x \delta_x + v_y \delta_y + \dots v_m \delta_m + v_1 \delta_1 + v_2 \delta_2 + v_3 \delta_3 \\ &= \vec{v}_{int} + \vec{v}_{ext}\end{aligned}\quad (2.1)$$

where v 's are the components of the particle velocity and δ 's are the unit vectors along the various co-ordinate axes. In general, the external velocity \vec{v}_{ext} can be calculated from the fluid velocity and momentum exchange relationship between fluid and particle. The internal particle velocity \vec{v}_{int} is assumed to be unique function of state of the particle.

There is also a possibility of particles suddenly appearing or disappearing in particle phase space along any co-ordinate. This possibility is represented by the birth and death functions of particle distribution. In general, these birth and death functions can be related to the state of the total system and the position of a particle in the particle phase space.

The population balance equation for particles in some fixed subregion Z_1 of particle phase space will be based on

the usual conservation law.

$$\text{Accumulation} = \text{Input} - \text{Output} + \text{net generation} \quad (2.2)$$

Consider the subregion to move convectively with particle phase space velocity \vec{v} . The population balance equation can be written as

$$\frac{d}{dt} \int_{Z_1} n dZ = \int_{Z_1} (B-D) dZ \quad (2.3)$$

where B and D represent birth and death functions.

Expanding L.H.S.

$$\begin{aligned} \frac{d}{dt} \int_{Z_1} n dZ &= \int_{Z_1} \frac{\partial n}{\partial t} dZ + n \frac{d\vec{x}}{dt} \bigg|_{Z_1} \\ &= \int_{Z_1} \left[\frac{\partial n}{\partial t} + \vec{\nabla} \cdot \left(\frac{d\vec{x}}{dt} n \right) \right] dZ \end{aligned}$$

where \vec{x} are the co-ordinates of system.

Replacing $\frac{d\vec{x}}{dt}$ by $(\vec{v}_{\text{ext}} + \vec{v}_{\text{int}})$ and combining with 2.3 gives the following expression:

$$\int_Z \left[\frac{\partial n}{\partial t} + \vec{\nabla} \cdot (\vec{v}_{\text{ext}} \cdot n) + \vec{\nabla} \cdot (\vec{v}_{\text{int}} \cdot n) + D-B \right] dZ = 0$$

The equation is valid for any arbitrary region Z_1 .

Hence the integrand vanishes for all Z.

$$\text{Therefore } \frac{\partial n}{\partial t} + \vec{\nabla} \cdot (\vec{v}_{\text{ext}} \cdot n) + \vec{\nabla} \cdot (\vec{v}_{\text{int}} \cdot n) + D-B = 0 \quad (2.4)$$

This is a general expression which can be used to describe any particulate system, provided the number of

particles in the system is sufficiently large so that $n(Z,t)$ can retain its statistical significance.

This equation will now be applied to a continuous well stirred particulate system of volume $V(t)$, having a population density function $n(Z,t)$, p input and q output streams. Since in a well stirred system, tracing of external coordinates of particles is not of interest, this equation can be integrated over the range of external co-ordinates V or volume occupied by the system

$$\int_V \left(\frac{\partial n}{\partial t} + \vec{\nabla} \cdot \vec{v}_{int} + D-B + \vec{\nabla} \cdot \vec{v}_{ext} \cdot n \right) dV = 0$$

Therefore
$$V \left[\frac{\partial n}{\partial t} + \vec{\nabla} \cdot \vec{v}_{int} \cdot n + D-B \right] + \int_V \vec{\nabla} \cdot \vec{v}_{ext} \cdot n dV = 0 \quad (2.5)$$

The integral can be transformed into a surface integral of population flux over the surfaces of the system. The surfaces are: the cross-sections of input-output streams, free interface of liquid and total fluid-particle interface. The final expression is

$$\int_V \vec{\nabla} \cdot \vec{v}_{ext} \cdot n dV = n \frac{dV}{dt} - \sum_p Q_i n_i + \sum_q Q_o n \quad (2.6)$$

where Q 's are the flow rates of input-output streams.

Combining 2.5 and 2.6 and simplifying -

$$\frac{\partial n}{\partial t} + \vec{\nabla} \cdot \vec{v}_{int} \cdot n + n \frac{d(\log V)}{dt} = B-D + \sum_p \frac{Q_i n_i}{V} - \sum_q \frac{Q_n}{V} \quad (2.7)$$

This expression represents the macroscopic population balance equation for a particulate, well mixed system. With the knowledge of input-output streams, expressions for velocity \vec{v}_i , birth and death functions this equation can be used to model emulsion polymerization in a CSTR mathematically. The values of population density $n(t)$ can be obtained and used to calculate the characteristics of the resultant polymer latex.

In next chapter, this equation will be used to develop the design equations for an emulsion polymerization CSTR.

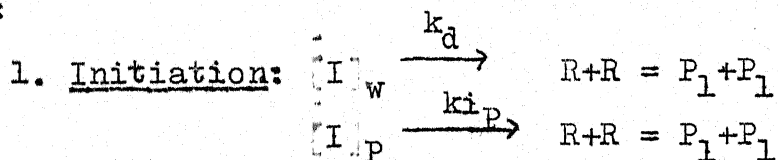
CHAPTER 3

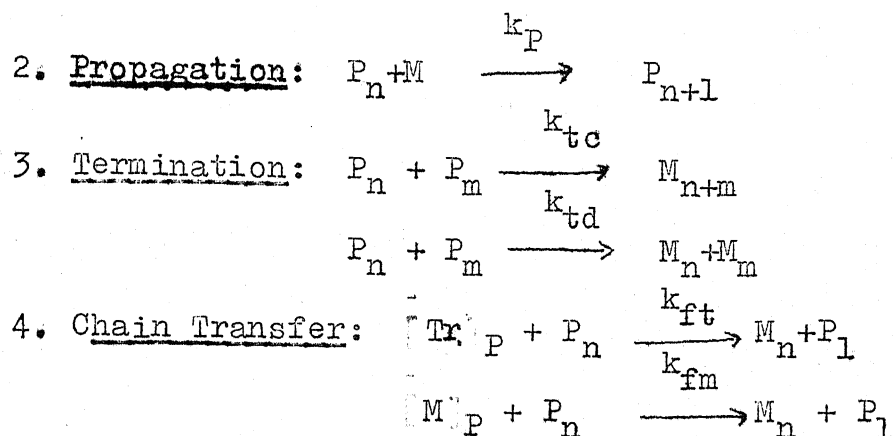
FORMULATION OF MODELLING EQUATIONS

From the previous chapters it can be seen that an emulsion polymerization latex is a particulate system comprising of a continuous aqueous phase and a particulate phase consisting of polymer particles, micelles and monomer droplets. A set of equations, consisting of material balance equations for continuous phase and population balance equations for particulate phase, completes the mathematical representation of physical polymerization system. Energy balance equations for the system are generally unnecessary because of low latex viscosity and high heat transfer rates, which result in an isothermal operation.

In the following pages development of the modelling equations will be given. This presentation closely follows the approach given by Min and Ray [17]. For convenience, nomenclature for this chapter is provided at the end of the chapter.

Kinetic Scheme: The usual kinetic scheme involving initiation, propagation, termination as well as chain transfer to monomer and transfer agents has been used. The scheme is given below:





Particulate Phase:

For characterizing a polymer latex, the polymer particles are of prime importance. Very little, if any, reaction takes place in the monomer droplets or micelles. Hence it is not necessary to develop separate population balance equations for these species. It will be assumed that these distributions can be represented by average micelle size (volume) v_m and average droplet size (volume) v_d respectively.

With the polymer recipe and average residence time of the reactor usually employed in practice, the number density of polymer particles is of the order of $10^{14}/\text{cm}^3$ emulsion. Hence we can attach an appropriate distribution function to the particles and use macroscopic population balance equation as given in the last chapter.

$$\frac{\partial n}{\partial t} + \nabla \cdot \vec{v}_{int} \cdot n + n \frac{d}{dt}(\log V) = B - D + \sum_p \frac{Q_{in,p}}{V} - \sum_q \frac{Q_{out,q}}{V} \quad (2.7)$$

Consider a reactor of volume V , with an input and output stream, both having a volumetric flow rate of q . The

The contents of reactor are assumed to be completely mixed. Let r_i represent the net generation term (B-D) resulting in discrete changes of variables due to the various mechanisms. Let n be the distribution function under consideration. The aforesaid equation, when applied to such system becomes

$$\frac{\partial n}{\partial t} + \vec{\nabla} \cdot \vec{v}_{int} \cdot n = \sum_{i=1}^m r_i + \frac{1}{\theta} n \left(\begin{array}{l} \text{feed} \\ - \frac{1}{\theta} n \end{array} \right) \quad (3.1)$$

where θ represents the mean residence time and m depends upon the number of mechanisms to be included in the model.

Radical Number Distribution:

Consider a distribution function $f(i, v, t)$ which represents number of particles per cm^3 of emulsion, having volume v to $v+dv$ and i radicals, at any time t . The knowledge of this distribution is necessary to calculate the rate of polymerization and \bar{i}_{avg} . The population balance equation corresponding to this function is

$$\frac{\partial f(i, v, t)}{\partial t} + \frac{\partial}{\partial v} \cdot (r_v \cdot f(i, v, t)) = \sum_{i=1}^m r_i + \frac{1}{\theta} f(i, v, t) \left(\begin{array}{l} \text{feed} \\ - \frac{1}{\theta} f(i, v, t) \end{array} \right) \quad (3.2)$$

First term represents rate of change of $f(i, v, t)$ with respect to time. r_v in the second term represents the rate of change of volume of particle with respect to time, [ie $\frac{dv}{dt}$].

The size of a particle varies due to the propagation reaction taking place inside the particle and due to particle swelling. This quantity will depend upon the model assumed for particle structure. For homogeneous particle morphology Gardon [4] has calculated this quantity by correlating the propagation mechanism with the monomer concentration and polymer and monomer densities. He obtained following relationship for the conditions when the monomer droplets are present in the system and the aqueous phase is saturated (with monomer).

$$\frac{dv}{dt} = r_v = \frac{k_P}{N_A} \cdot \frac{d_m}{d_P} \cdot \frac{\bar{\phi}}{(1-\bar{\phi})} \cdot i \quad (3.3)$$

The rate terms r_i in the RHS of 3.2 depend upon the particular mechanisms to be taken into consideration. These terms will be discussed below.

First term, r_1 represents the change in $f(i, v, t)$ due to absorption of radicals into the particles. Radicals can either propagate and precipitate as oligomers in the aqueous phase or can be absorbed into the particles. The relative importance of these two mechanisms depends upon the monomer solubility and the propagation rate constant in the aqueous phase. Gardon [4] has obtained an expression for rate of radical absorption by computing the probability of collision of a radical with micelles and particles. The final expression is

$$R_{ia} = \frac{2 f k_d [I]_w}{A_m + A_p} \quad \text{when } S \gg 4/L$$

$$= \frac{2 f k_d [I]_w}{(4/L)} \quad \text{when } S \ll 4/L$$

where L represents the average length travelled by a growing chain before it reaches the critical length at which it precipitates and is given by

$$L = (2D t_{\text{prec}})^{0.5}$$

Here t_{prec} represents the time required by radical to reach critical chain length for precipitation and is a function of the propagation rate constant and monomer concentration in the aqueous phase. D is the diffusion coefficient of radicals in the aqueous phase.

Rate r_1 in eqn.3.2 is then given by the following expression

$$r_1 = R_{ia} \times a_p N_A (f(i-1,v,t) - f(i,v,t)) \quad (3.4)$$

where the first term represents the formation of $f(i,v,t)$ by absorption of a radical into $f(i-1,v,t)$ type of particles and the second term represents destruction of $f(i,v,t)$ type of particle by absorption of a radical, thus transforming it into a $f(i+1,v,t)$ type particle.

Second rate term (r_2) represents the effect of formation of a radical pair inside the particle by the decomposition of monomer-soluble initiator.

$$r_2 = f k_{ip} [I]_P \cdot N_A \cdot v (f(i+2, v, t) - f(i, v, t)) \quad (3.5)$$

Radicals can also desorb from the particle. This effect can be accounted for by the following expression:

$$r_3 = \frac{k_{op}}{v} [(i+1) f(i+1, v, t) - i f(i, v, t)] \quad (3.6)$$

First term represents the formation of $f(i, v, t)$ type particle by radical desorption from a $f(i+1, v, t)$ type particle and second term represents the destruction of $f(i, v, t)$ by desorption of a radical.

Rate of change of $f(i, v, t)$ by termination of two radicals inside the particle, is given by the following expression:

$$r_4 = \frac{k_{tc} + k_{td}}{2} \left(\frac{1}{N_A v} \right) [(i+2)(i+1) f(i+2, v, t) - (i)(i-1) f(i, v, t)] \quad (3.7)$$

The rate of formation of new particles by radical absorption in micelles is given by

$$r_5 = R_i \times \frac{a_m \cdot m}{A_m + A_p} \delta_{il} \delta(v - v_m) \quad (3.8)$$

δ functions on right hand side of the expression indicate that new particles, thus formed, will have volume v_m and will contain one radical.

Particle Size Distribution:

Let $F(v, t)$ be the particle size distribution function. This distribution is quite important since the particle size is one of the characterizing properties of a polymer latex.

Consider the balance equation for $F(v, t)$, which represents the number of particles having volume between v and $v+dv$ at any time t . It is related to radical number distribution by the following equation:

$$F(v, t) = \sum_{i=0}^{\infty} f(i, v, t) \quad (3.9)$$

By combining 3.2 to 3.9 the population balance equation for $F(v, t)$ can be derived to be

$$\begin{aligned} \frac{\partial F(v, t)}{\partial t} &+ \frac{\partial}{\partial v} (r_v F(v, t)) = \\ \text{Rate of change of } F(v, t) \text{ with respect to time} &+ \text{Rate of change of } F(v, t) \text{ due to volume growth of particles} \\ R_i \frac{A_m}{A_m + A_p} \cdot \delta(v - v_m) &+ \frac{1}{\theta} [F(v, t)_{\text{feed}} - F(v, t)] \\ \text{Rate of change of } F(v, t) \text{ due to generation of new particles of size } v_m &+ \text{Change of } F(v, t) \text{ due to input and output} \end{aligned} \quad (3.10)$$

where

$$r_v = \frac{k_p}{N_A} \cdot \frac{d_m}{d_p} \cdot \frac{\bar{\phi}}{(1 - \bar{\phi})} \cdot \bar{i}_{\text{avg}} \quad (3.11)$$

and \bar{i}_{avg} [i.e. average number of radicals per particle] is given by

$$\bar{i}_{\text{avg}} = \frac{\int_{v_m}^{\infty} \sum_{i=0}^{\infty} i f(i, v, t) dv}{\int_{v_m}^{\infty} \sum_{i=0}^{\infty} f(i, v, t) dv} \quad (3.12)$$

Live Polymer Chain Length Distribution:

In order to calculate the molecular weight distribution the chain length distribution of the polymer radicals present inside the particle is needed. This distribution is used to solve the dead polymer chain length distribution which is in turn used to obtain the m.w.d. of the polymer product.

Consider the particles having volume between v and $v+dv$ and i radicals at any time t . These i radicals will in general have different chain lengths. Amongst these radicals consider the one having chain length n . Let $\bar{L}(n,i,v,t)$ represent the total number of radicals of chain length n present in all the particles having volume between v and $v+dv$ and having i radicals, at any time t . The population balance for such radicals is given by the following expression.

$$\begin{aligned} \frac{\partial \bar{L}(n,i,v,t)}{\partial t} + \frac{\partial}{\partial v}(r_v \bar{L}(n,i,v,t)) &= \sum_{i=1}^k r_i \\ &+ \frac{1}{\theta} [\bar{L}(n,i,v,t) \Big|_{\text{feed}} - \bar{L}(n,i,v,t)] \end{aligned} \quad (3.13)$$

First term represents change in $\bar{L}(n,i,v,t)$ with respect to time. The second term represents change in $\bar{L}(n,i,v,t)$ due to change in size of a particle, where r_v represents the rate of change of volume with respect to time.

The rate terms r_i on right hand side, as before, depend on specific assumptions made. Firstly consider the effect of absorption of radicals. It has been assumed that

only primary radicals can absorb into the particle and that the particle nucleation by precipitation in aqueous phase has been negligible. Let $g(n,i,v,t)$ represent number of radicals of chain length n in a $f(i,v,t)$ type particle. When a radical is absorbed in a $f(i-1,v,t)$ type particle it converts the particle into a $f(i,v,t)$ type, making in the process, a contribution of $g(n,i-1,v,t)$ radicals to $L(n,i,v,t)$. Similarly when a primary radical is absorbed in a $f(i,v,t)$ type particle $g(n,i,v,t)$ number of radicals will be lost from $L(n,i,v,t)$. When a primary radical is absorbed in a particle it adds only to $L(1,i,v,t)$. Thus the total effect, per cm^3 of emulsion can be represented as:

$$\begin{aligned}
 r_1 &= R_i \frac{a_p}{A_p + A_m} \cdot f(i-1,v,t) \cdot g(n,i-1,v,t) \\
 &\quad - R_i \frac{A_p}{A_p + A_m} f(i,v,t) \cdot g(n,i,v,t) \\
 &\quad + R_i \frac{A_p}{A_p + A_m} f(i-1,v,t) \delta n_1 \\
 &= R_i \left[\frac{a_p}{A_p + A_m} \right] [f(n,i-1,v,t) - f(n,i,v,t) + f(i-1,v,t) \delta n_1]
 \end{aligned}
 \tag{3.14}$$

From the definition

$$g(n,i,v,t) = \frac{L(n,i,v,t)}{f(i,v,t)}
 \tag{3.15}$$

The initiator dissolved in a polymer particle can decompose to give a pair of radicals of chain length one each. A contribution of $g(n,1-2,v,t)$ is made to $L(n,i,v,t)$

per generation of a pair due to such an event. In addition to this the radicals formed themselves contribute to addition of radicals with chain length one. The same process occurring in a $f(i,v,t)$ type particle would result in a loss to $L(n,i,v,t)$

$$\begin{aligned}
 r_2 &= k_{ip} [I]_P \cdot v \cdot N_A [(g(n,i-2,v,t) + 2\delta n_1)f(i-2,v,t) \\
 &\quad - g(n,i,v,t) \cdot f(i,v,t)] \\
 &= k_{ip} [I]_P \cdot v \cdot N_A [L(n,i-2,v,t) - L(n,i,v,t) + 2\delta n_1 f(i-2,v,t)] \\
 &\quad (3.16)
 \end{aligned}$$

The effect of radical desorption from particles is given by

$$\begin{aligned}
 r_3 &= k_o \frac{a_p}{v} [(i+1)f(i+1,v,t) \cdot g(n,i+1,v,t)(1 - \frac{1}{i+1}) \\
 &\quad - i f(i,v,t) g(n,i,v,t)] \\
 &= \frac{k_o a_p}{v} [(i) L(n,i+1,v,t) - (i) L(n,i,v,t)] \quad (3.17)
 \end{aligned}$$

Here the first term represents the production of $L(n,i,v,t)$ by desorption of radicals from $f(i+1,v,t)$ except when a radical of length n desorbs. The second term represents loss due to desorption from radicals from $f(i,v,t)$ type particles.

The effect of the kinetic steps inside the polymer particle depends upon the structure of particle. Assuming homogeneous particle morphology, the effect of propagation and transfer reaction can be expressed as

$$\begin{aligned}
r_4 &= k_p[M]_P f(i,v,t) \cdot g(n-1,i,v,t) - k_p[M]_P f(i,v,t) g(n,i,v,t) \\
&\quad - k_{ft}[Tr]_P f(i,v,t) g(n,i,v,t) + k_{ft}[Tr]_P i f(i,v,t) \delta n_1 \\
&\quad - k_{fm}[M]_P f(i,v,t) g(n,i,v,t) + k_{fm}[M]_P i f(i,v,t) \delta n_1 \\
&= k_p[M]_P [L(n-1,i,v,t) - L(n,i,v,t)] + k_{ft}[Tr]_P [-L(n,i,v,t) \\
&\quad + f(i,v,t) \cdot i \cdot \delta n_1] + k_{fm}[M]_P [f(i,v,t) \cdot i \cdot \delta n_1 - L(n,i,v,t)]
\end{aligned}
\tag{3.18}$$

The first two terms represent the effect of propagation; third and fourth term represent chain transfer to transfer agent and the fifth and sixth term represent chain transfer to monomer. The chain transfer reactions involve addition of one chain to $f(l,i,v,t)$. This effect has been accounted for in fourth and sixth term.

The effect of termination reactions can be written as

$$\begin{aligned}
r_5 &= \frac{k_{tc} + k_{td}}{N_A v} \left[\frac{(i+2)(i+1)}{2} - (i+1) \right] \cdot L(n,i+2,v,t) \\
&\quad - \frac{(i)(i-1)}{2} L(n,i,v,t) \\
&= \frac{k_{tc} + k_{td}}{N_A v} \left[\frac{(i+1)(i)}{2} L(n,i+2,v,t) - \frac{(i)(i-1)}{2} L(n,i,v,t) \right]
\end{aligned}
\tag{3.19}$$

First and third terms in the initial expression represent change due to termination between any two radicals. Third term represents the termination between a radical of chain length n and remaining $(i+1)$ radicals. The possibility

of termination between two radicals of same chain length n , which will be a second order term, has been neglected.

The effect due to radical absorption in micelles is given by the following expression:

$$\begin{aligned} r_6 &= R_i \frac{a_m}{(A_p + A_m)} \cdot m \cdot l \cdot \delta n_l \delta(v - v_m) \delta_{il} \\ &= R_i \frac{A_m}{(A_p + A_m)} \delta n_l \delta(v - v_m) \delta_{il} \end{aligned} \quad (3.20)$$

Dead Polymer Chain Length Distribution:

This distribution is related to the chain length of dead polymer molecules inside the particles and is thus closely connected with m.w.d. of the resultant polymer latex. Consider the particles having volume between v and $v+dv$ and possessing i active radicals at any time t . Apart from these i active radicals, each particle contains some dead polymer chains. Let $G(n, i, v, t)$ represent total number of these dead chains contained in all $f(i, v, t)$ type particles at any time t . The population balance equation for this function can be written as

$$\begin{aligned} \frac{\partial G(n, i, v, t)}{\partial t} + \frac{\partial r_v \cdot G(n, i, v, t)}{\partial v} &= \sum_{i=1}^k r_i \\ + \frac{1}{\theta} [G(n, i, v, t) \Big|_{\text{feed}} - G(n, i, v, t)] & \quad (3.21) \end{aligned}$$

First term represents the change in $G(n, i, v, t)$ with respect to time and the second term expresses the change due to change in volume of particle.

Rate terms on the right hand side depend upon the particular mechanisms to be considered in the model. Derivation of these terms is similar to that for corresponding terms in live polymer chain length distribution. These terms will be discussed briefly.

The first rate term r_1 represents the effect of radical absorption on $G(n,i,v,t)$. Let $g'(n,i,v,t)$ represent number of dead polymer chains per particle. Hence $G(n,i,v,t) = g'(n,i,v,t) f(i,v,t)$. When a new radical enters a polymer particle of type $f(i-1,v,t)$, it converts that particle into $f(i,v,t)$ type particle and a contribution of $g'(n,i-1,v,t)$ is made towards $G(n,i,v,t)$. Hence the total contribution of this mechanism towards change in $G(n,i,v,t)$ is given by

$$\begin{aligned} r_1 &= R_i \frac{a_p}{(A_m + A_p)} f(i-1,v,t) g'(n,i-1,v,t) - R_i \frac{a_p}{(A_m + A_p)} \\ &\quad f(i,v,t) g'(n,i,v,t) \\ &= R_i \frac{a_p}{(A_m + A_p)} [G(n-1,i,v,t) - G(n,i,v,t)] \quad (3.22) \end{aligned}$$

Similarly the effect of radical initiation in particles and desorption from particles can be shown to be given by following expression. It is assumed here that dead polymer chains can not desorb from particles.

$$r_2 = f_{kd} [I]_p [G(n,i-2,v,t) - G(n,i,v,t)] \quad (3.23)$$

$$r_3 = \frac{k_o a_p}{v} [(i+1) G(n,i+1,v,t) - i G(n,i,v,t)] \quad (3.24)$$

The effect of the kinetic steps is more important in case of $G(n,i,v,t)$ and can be given by following expression:

$$\begin{aligned}
 r_4 = & k_{ft}[\text{Tr}]_P L(n,i,v,t) + k_{fm}[\text{M}]_P L(n,i,v,t) \\
 & + \frac{k_{tc}+k_{td}}{2} [(i+2)(i+1) G(n,i+2,v,t) - (i)(i-1) G(n,i,v,t)] \\
 & + \frac{k_{tc}}{2v} [f(i+2,v,t) \sum_{m=1}^{n-1} g(m,i+2,v,t) g(n-m,i+2,v,t)] \\
 & + \frac{k_{td}}{2v} [g(n,i+2,v,t) f(i+2,v,t)] \quad (3.25)
 \end{aligned}$$

In this expression the first two terms represent formation of dead polymer chains due to chain transfer with transfer agent and with monomer. Third term expresses the change in $G(n,i,v,t)$ by termination of any two radicals in $f(i+2,v,t)$ type particle (i.e. formation) and in $f(i,v,t)$ type particle (destruction). Fourth term represents the additional contribution when two terminating radicals form a polymer chain of length n . Last term represent the possibility of disproportionation involving a live polymer radical of chain length n .

Apart from these population balance equations, balances which define the environment of particles must also be written. These equations include material balances for various reactive species and balances for number of droplets, micelles etc.

Emulsifier Balances:

The total amount of soap present per cm^3 emulsion is given by following expression:

$$\frac{d(S)_T}{dt} = \frac{1}{\theta} \left[\bar{[S]}_T \text{ feed} - \bar{[S]}_T \right] \quad (3.26)$$

This soap may only be present in aqueous phase as micelles or may be **absorbed** on the particles and droplets. The process of transfer of emulsifier will be assumed to be extremely fast. Under these conditions the balance for number of micelles yields

$$\frac{m a_m}{\theta E} = \bar{[S]}_T - S_{wc} - \frac{A_p}{\theta E} - \frac{m_d v_d}{\theta E} \quad (3.27)$$

The terms on right hand side represent total soap, the soap present in the aqueous phase, the soap adsorbed on particles and the soap adsorbed on the monomer droplet respectively.

Total Monomer Balance:

The balance for the total amount of monomer present in the reactor is given below:

$$\frac{d V_R \bar{[M]}_T}{dt} = \frac{V_R}{\theta} \left[\bar{[M]}_T \text{ feed} - \bar{[M]}_T \right] - V_R k_p \bar{[M]}_p \int_0^{\infty} \sum_{i=0}^{\infty} i f(i, v, t) dv \quad (3.28)$$

The first two terms represent the effect due to input and output terms. The third term represents the loss of monomer due to propagation in particles. Here the small contributions due to chain transfer and propagation in aqueous phase are neglected.

Monomer Droplets Balance:

Monomer droplets balance can be written as follows. As in the previous case the dynamic of transfer of monomers will be assumed to be extremely fast.

$$m_d v_d = [M]_T V_{mol} - \int_0^{\infty} V[M]_P V_{mol} \cdot F(v) \cdot dv - V_{mol} m \cdot [M]_m V_{mol} \quad (3.29)$$

The terms on right hand side of equation represent total monomer content, monomer in particles, and monomer in micelles.

Aqueous Phase Balances:

The balance for concentration of the initiator in aqueous phase is given below

$$\frac{d V_w [I]_w}{dt} = - V_w k_d [I]_w + \frac{V_w}{\theta} [I]_w \Big|_{\text{feed}} - [I]_w \quad (3.30)$$

Lastly an expression for the monomer concentration in polymer (i.e. $[M]_P$ or $\bar{\phi}$) is required. The nature of the equation will, in general, depend upon the model for particle structure considered. For a homogeneous particle morphology the free energy of mixing of polymer can be equated to the surface free energy [4]. The resultant expression for $\bar{\phi}$ is:

$$(1-\bar{\phi}) + \ln(\bar{\phi}) + \frac{2 \gamma V_{mol}}{RT r} = 0 \quad (3.31)$$

So far the population balance equations for particulate phase and the material balance for aqueous phase have been

discussed. These balances are assumed to represent the given system completely. These equations can be applied to a practical system and the results can be used to obtain desired characteristics of the resultant polymer latex.

NOMENCLATURE

CE	surface coverage by the emulsifier, cm^2/mole
a_m	surface area of micelles, cm^2
a_p	surface area of particle of size v , cm^2
A_m	total surface area of micelles, cm^2/cm^3 emulsion
A_p	total surface area of particles, cm^2/cm^3 emulsion
d_p	density of polymer, gms/cm^3
d_m	density of monomer, gms/cm^3
f	efficiency of initiator decomposition, dimensionless
$f(i,v,t)$	number of particles containing i radicals and possessing volume between v to $v+dv$ at time t , moles/cm^3 emulsion
$F(v,t)$	number of particles having volume between v and $v+dv$ at time t , moles/cm^3 emulsion
$g(n,i,v,t)$	number of live chains of chain length n present in a particle containing i radicals and possessing volume between v to $v+dv$ at time t , dimensionless
$g'(n,i,v,t)$	number of dead chains of chain length n present in particle containing i radicals and possessing volume between v to $v+dv$ at time t , dimensionless
$G(n,i,v,t)$	number of dead chains of chain length n present in all the particles having i radicals and possessing volume v to $v+dv$, at time t , moles/cm^3 emulsion
i	number of radicals in a particle, dimensionless
\bar{i}_{avg}	average number of radicals per particle, dimensionless
$[I]_w$	initiator concentration in aqueous phase, moles/cm^3 , water
$[I]_p$	initiator concentration in polymer particles, moles/cm^3

k_{ip}	rate constant for dissociation of initiator in polymer phase, l/cms
k_d	rate constant for dissociation of initiator in aqueous phase, l/cms
k_p	rate constant for propagation, $\text{cm}^3/\text{mole} \cdot \text{sec}$.
k_{tc}	rate constant for termination by combination, $\text{cm}^3/\text{mole} \cdot \text{sec}$.
k_{td}	rate constant for termination by disproportionation, $\text{cm}^3/\text{mole} \cdot \text{sec}$
k_{ft}	rate constant for chain transfer to transfer agent, $\text{cm}^3/\text{mole} \cdot \text{sec}$.
f_{fm}	rate constant for chain transfer to monomer, $\text{cm}^3/\text{mole} \cdot \text{sec}$.
k_o	rate constant for radical desorption, cms/sec
$L(n,i,v,t)$	number of live chains of chain length n in all the particles having i radicals and possessing volume between v and $v+dv$ at time, t , moles/ cm^3 emulsion
m	number of micelles, moles/ cm^3 emulsion
$[M]_p$	concentration of monomer in the particles, moles/ cm^3
$[M]_m$	concentration of monomer in the micelles, moles/ cm^3
$[M]_T$	total concentration of monomer in the reactor, moles/ cm^3 emulsion
n	chain length of polymer, dimensionless
r	radius of particle, cms
R_i	rate of initiation, moles/sec.
R_{ia}	rate of absorption of radicals in particles, and micelles, moles/sec. cm^2

$[S]_T$	total emulsifier concentration, moles/cm ³ emulsion
S_{wc}	critical miscelle concentration, moles/cm ³ water
t	time, sec.
$[Tr]_P$	concentration of chain transfer agent inside the particle, moles/cm ³
v	volume of particle, cm ³
v_m	volume of micelle, cm ³
v_d	volume of monomer droplets, cm ³
V_R	volume of reactor, cm ³
V_w	volume of aqueous phase per cm ³ emulsion, cm ³ /cm ³ emulsion
$\bar{\phi}$	monomer volume fraction in the particles, dimensionless
Ψ	monomer polymer interaction parameter, dimensionless
γ	surface tension, dynes/cm
θ	mean residence time, sec.
δ_{ig}	Kronecker delta function, 1 for $i=j$, 0 for $i \neq j$
$\delta(v-v_m)$	Dirac delta function; 1 for $v=v_m$, 0 for $v \neq v_m$.

CHAPTER 4

METHOD OF SOLUTION OF THE MODELLING EQUATIONS

In the previous chapter, the modelling equations for emulsion polymerization in a CSTR were presented. These equations contain various material balance equations and the population balance equations. The material balance equations are ordinary differential or nonlinear algebraic equations and are relatively easy to handle. However, the various population balance equations involve different distribution functions like $F(v)$, $f(i,v)$, $L(n,i,v)$ and $G(n,i,v)$ and are complex coupled partial differential-difference equations. Special care is required to decide the line of approach to solve these equations. In this chapter normalisation procedures have been introduced and for convenience the additional nomenclature for this chapter is listed at the end of the chapter.

The aim of solving these equations is to calculate the values of the aforesaid distribution functions. However, generation of these functions numerically for different values of attributes may take exorbitant time. To avoid this, the problem can be approached by making use of the concept of moments of a distribution,

Let $f(x)$ represent a continuous distribution for the attribute x . Then the k th moment of $f(x)$ is defined as

$$\bar{f}_k = \int_0^{\infty} x^k f(x) dx \quad (4.1)$$

Similarly if $H(i)$ represents a distribution defined only for discrete values of i , then the k th moment of the discrete distribution is given by

$$\bar{H}_k = \sum_{i=i \min}^{i=i \max} i^k H(i) \quad (4.2)$$

It can be seen from these definition that, with the knowledge of a distribution function, all of its moments can be calculated. Obviously, a continuous distribution function, say $f(x)$, containing infinite information cannot be exactly specified by a finite set of moments. However, it is possible to represent a distribution function, in terms of finite number of moments to an accuracy sufficient for modelling purposes. One of the methods to achieve this is to represent the distribution function in a series expansion of orthogonal polynomials like Laguerre polynomial [17,30]. The Laguerre polynomial $L_n^\alpha(x)$ has the following orthogonal property.

$$\int_0^{\infty} (x)^\alpha \exp(-x) L_n^\alpha(x) L_m^\alpha(x) dx = \frac{\Gamma(\alpha+1) \Gamma(n+1)}{\Gamma(n+1)} \delta_{nm} \quad (4.3)$$

$$\text{where } \delta_{nm} = 1 \quad \text{if } n=m \quad (4.4)$$

$$= 0 \quad \text{if } n \neq m$$

$$\text{where } L_n^\alpha(x) = \sum_{j=0}^n (-1)^j \binom{n+\alpha}{n-j} \frac{x^j}{j!} \quad (4.5)$$

Using the definition and orthogonal property of Laguerre polynomial, as stated above, a normalized distribution function $G(y)$ can be represented in terms of summations involving Laguerre polynomial [40].

$$G(y) = \frac{y^\alpha \exp(-y)}{\Gamma(\alpha+1)} \sum_{n=0}^{\infty} c_n^\alpha L_n^\alpha(y) \quad (4.6)$$

The coefficients C_n^α are related to the moments \bar{G}_k of the normalized distribution function $G(y)$ by following expression:

$$c_n^\alpha = \sum_{j=0}^n (-1)^j \frac{\Gamma(n)}{\Gamma(n-j) \Gamma(j)} \frac{\Gamma(\alpha+1)}{\Gamma(\alpha+j+1)} \bar{G}_j \quad (4.7)$$

Equation 4.7 follows from the orthogonal properties of L_n^α . Following Min [30], the parameter α is chosen so that c_1^α and c_2^α are zero, which results in rapid convergence. The resulting expression for α is the following:

$$\alpha = -1 + \frac{(\bar{G}_1)^2}{\bar{G}_2 - (\bar{G}_1)^2} \quad (4.8)$$

The normalized function $G(y)$, used here is related to the absolute distribution function $f(x)$ as follows:

$$f(x)dx = G(y) dy \frac{f_0}{f_0}$$

and
$$y = \frac{f_1}{f_0(\alpha+1)} \quad (4.9)$$

The moments of the two distributions, $f(x)$ and $G(y)$, are related to each other by the following expression

$$\bar{f}_k = \bar{f}_0 \left[\frac{\bar{f}_1}{\bar{f}_0(\alpha+1)} \right]^{k-1} \bar{G}_k \quad (4.10)$$

It can be seen from the preceding lines that most of the information contained in a continuous distribution function can be obtained from a finite set of moments of that distribution. The number of moments required to specify a differential distribution depends on the complexity of its shape. It has been found that only 3 or 4 moments are often sufficient to represent a unimodal distribution satisfactorily.

Even in practical cases, it is often impossible to measure complete molecular weight distribution or particle size distribution. Generally it is convenient to measure and use the leading moments of these distributions for quality control. This method involves the use of M_n , M_w or polydispersity index to represent m.w.d. and average size to specify the particle size distribution.

For the reason stated above, the modelling equations given in the previous chapter involving differential distribution functions will be transformed to a set of equations involving finite number of moments of those distributions.

It should be noted that the method to represent a distribution in terms of its moments is useful only for continuous distributions. Obviously a discrete distribution containing a large number of closely spaced points can be approximated by a continuous distribution. Consequently, in

the present study the live polymer chain distribution as well as the dead polymer chain length distributions which are actually defined for integer values of n only, are approximated by a continuous distribution in n and the aforesaid method is then made use of.

However, a distribution of polymer particles in i (the number of radicals contained in the particle) is defined only at a few points. It has been found from experiments that the number of live radicals contained in a particle rarely exceeds 10. As a result, the use of moments to represent the discrete distribution in i is not justifiable. Hence this distribution will be generated numerically at the discrete points at which it actually exists (i.e. at $i = 0, 1, 2, \dots$). Since the number of points at which it is defined is small this technique involves the solution of a relatively small number of equations.

To start with, the following assumptions will be made for mathematical simplicity.

1. A distribution in the volume of particles is generated. The fractional moments in volume (i.e. two third and one third moment) are used to calculate the total surface area and the average diameter of the particles. However, when ' v ' is connected with the distribution functions in a more complex way (like the term arising due to termination of radicals in particle), the average value of the volume

of particles is used for mathematical simplicity. This approximation is likely to be more exact when the particle size distribution is relatively narrow.

2. The terms representing the chain transfer to transfer agent and initiation in particles will be maintained while developing the sets of equations involving moments of live and dead polymer chain length distribution. However, the values of $[\bar{\text{Tr}}]_p$ and $[\bar{\text{I}}]_p$ can be evaluated by addition of two material balance equations and thus these mechanisms can be easily accounted for.

3. The value of monomer concentration in particles ($\bar{\phi}$) can be calculated from equation 3.30. This equation shows that ($\bar{\phi}$) is in general a function of particle size. However, from experimental studies it has been found that the concentration of monomer in a particle is almost constant and independent of its volume. The values of the concentrations of monomer in a polymer particle are reported in literature for different monomers. Hence, these values can be directly used to solve remaining modelling equations. Most of the results from this study have been obtained by this method. However, the results obtained without making this assumption are also given for comparison.

The modelling equations given in the earlier chapters are not dimensionless. Hence the numerical values of various variables depend upon the system of units to be used. These

equations should firstly be transformed into sets of nondimensional modelling equations. This approach facilitates the use of these equations, since the variables are not confined to any particular system of units. Similarly it facilitates the numerical solution of these equations, since the range of numerical values of various dimensionless variables is relatively narrow. The dimensionless population balance equations should then be transformed into the sets of equations involving moments of the dimensionless distributions.

This procedure has been explained in the following lines, for the case of live polymer chain length distribution.

Consider the population balance equation (3.10) involving live polymer chain length distribution:

$$\begin{aligned}
 \frac{\partial L(n,i,v,t)}{\partial t} + \frac{\partial}{\partial v} (r_v \cdot L(n,i,v,t)) = & R_{ia} \left[\frac{a_p}{A_p + A_m} \right] \\
 & [L(n,i-1,v,t) - L(n,i,v,t) + f(i-1,v,t) \cdot \delta_{nl}] \\
 & + k_d [I]_P v \cdot N_A [L(n,i-2,v,t) - L(n,i,v,t) + 2\delta_{nl} \cdot f(i-2,v,t)] \\
 & + \frac{k_o a_p}{v} [(i) L(n,i+1,v,t) - (i) L(n,i,v,t)] + k_p [M]_P \\
 & [L(n-1,i,v,t) - L(n,i,v,t)] + k_{ft} [Tr]_P [i f(i,v,t) \delta_{nl} \\
 & - L(n,i,v,t)] + k_{fm} [M]_P [i f(i,v,t) \delta_{nl} - L(n,i,v,t)] \\
 & + \frac{k_{tc} + k_{td}}{N_A v} \left[\frac{(i+1)(i)}{2} L(n,i+2,v,t) - \frac{(i)(i-1)}{2} L(n,i,v,t) \right] \\
 & + \frac{1}{\theta} [L(n,i,v,t) \uparrow_{\text{feed}} - L(n,i,v,t)] + R_{ia} \frac{A_m}{(A_p + A_m)} \delta_{nl} \delta_{il} \delta(v-v_m)
 \end{aligned}
 \tag{3.10}$$

At steady state the derivatives with respect to time can be equated to zero.

Similarly all the particles are assumed to have volume v_{avg} . Hence the derivatives with respect to \bar{v} also reduce to zero, and the average volume of the particles v_{avg} can then be substituted in place of volume, v , in these equations. Using the definitions of various dimensionless variables and dimensionless constants given at the end of this chapter and after further simplifications this equation reduces to

$$\begin{aligned}
 0 = & C_7 [\bar{I}]_w \frac{(\bar{v})^{2/3}}{\bar{F}^{2/3+\bar{m}}} [\bar{L}(n,i-1) - \bar{L}(n,i) + \bar{f}(i-1) \delta_{nl}] \\
 & + C_2 [\bar{I}]_p \cdot v [\bar{L}(n,i-2) - \bar{L}(n,i) + 2\delta_{nl} \bar{f}(i-2)] \\
 & + C_3 (\bar{v})^{1/3} [i \bar{L}(n,i+1) - i \bar{L}(n,i)] \\
 & + C_{101}(\bar{\phi}) [\bar{L}(n-1,i) - \bar{L}(n,i)] \\
 & + C_{63} [\bar{Tr}]_p [\delta_{nl} \cdot (i) \bar{f}(i) - \bar{L}(n,i)] \\
 & + C_{102}(\bar{\phi}) [\delta_{nl} \cdot i \bar{f}(i) - \bar{L}(n,i)] + \frac{C_4}{(\bar{v})} [(i) (i+1) \\
 & \bar{L}(n,i+2) - (i)(i-1) \bar{L}(n,i)] + \frac{1}{\theta} [\bar{L}(n,i) \underset{\text{feed}}{\bar{L}(n,i)}] \\
 & + C_7 [\bar{I}]_w \frac{\bar{m}}{(\bar{F}^{2/3} + \bar{m})} \delta_{il} \delta(v-v_m) \delta_{nl}.
 \end{aligned} \tag{4.11}$$

This dimensionless population balance equation is then converted into a set of equations containing moments of the distribution in n . To do this the equation is multiplied

by n^k and is summed over all possible values of n . After further simplifications the following equation is obtained:

$$\begin{aligned}
 0 = & \frac{c_7 [\bar{I}]_w (\bar{v})^{2/3}}{(\bar{F}^{2/3} + \bar{m})} [\bar{L}_k(i-1) - \bar{L}_k(i) + \bar{F}(i-1)] \\
 & + c_2 [\bar{I}]_p (\bar{v}) [\bar{L}_k(i-2) - \bar{L}_k(i) + 2 \bar{F}(i-2)] \\
 & + c_3 (\bar{v})^{1/3} [i \bar{L}_k(i+1) - (i) \bar{L}_k(i)] \\
 & + c_{101}(\bar{\varphi}) \left[\sum_{m=1}^k \frac{k}{m} \frac{k}{k-m} \bar{L}_{k-m}(i) \right] + c_{63} [\bar{Tr}]_p [i \bar{F}(i) \\
 & - \bar{L}_k(i)] + c_{102}(\bar{\varphi}) [i \bar{F}(i) - \bar{L}_k(i)] + \frac{c_4}{\bar{v}} [(i+1)(i) \bar{L}_k(i+2) \\
 & - (i)(i-1) \bar{L}_k(i)] + \frac{1}{\theta} [\bar{L}_k(i)_{\text{feed}} - \bar{L}_k(i)] \\
 & + c_7 [\bar{I}]_w \frac{\bar{m}}{\bar{F}^{2/3} + \bar{m}} \delta_{il}
 \end{aligned} \tag{4.12}$$

Other modelling equations can be rendered dimensionless and the population balance equations can be transformed to a set of equations involving moments of dimensionless distributions, by using the aforesaid approach. The resultant set of modelling equations is given on the following page.

Particle Size Distribution:

$$0 = \bar{I}_{avg} \cdot n \frac{\bar{\Phi}}{1-\bar{\Phi}} \bar{F}_{k-1} + C_7 \frac{[\bar{I}]_w(\bar{m})}{(\bar{F}_{2/3} + \bar{m})} \\ + \frac{1}{\theta} [\bar{F}_k \text{ feed} - \bar{F}_k] \quad k=0 \text{ to } n \quad (4.13)$$

Radical Number Distribution:

$$0 = C_7 \frac{[\bar{I}]_w(\bar{v})^{2/3}}{(\bar{F}_{2/3} + \bar{m})} [\bar{f}(i-1) - \bar{f}(i)] + C_3(\bar{v})^{-1/3} [(i+1)\bar{f}(i+1) \\ - i\bar{f}(i)] + \frac{C_4}{(\bar{v})} [(i+2)(i+1)\bar{f}(i+2) - (i)(i-1)\bar{f}(i)] \\ + \frac{C_7[\bar{I}]_w(\bar{m})}{(\bar{F}_{2/3} + \bar{m})} \delta_{i1} \delta(v-v_m) + \frac{1}{\theta} [\bar{f}(i) \text{ feed} - \bar{f}(i)] \\ + C_2(\bar{v})[\bar{I}]_P [\bar{f}(i-2) - \bar{f}(i)] \quad i = 0 \text{ to } m \quad (4.14)$$

Live Polymer Chain Length Distribution:

$$0 = C_7 \frac{[\bar{I}]_w(\bar{v})^{2/3}}{(\bar{F}_{2/3} + \bar{m})} [\bar{L}_k(i-1) - \bar{L}_k(i) + \bar{f}(i-1)] \\ + C_2[\bar{I}]_P(\bar{v}) [\bar{L}_k(i-2) - \bar{L}_k(i) + 2\bar{f}(i-2)] \\ + C_3(\bar{v})^{-1/3} [(i)\bar{L}_k(i+1) - (i)\bar{L}_k(i)] + C_{101}(\bar{\Phi}) \\ \left[\sum_{m=1}^k \frac{k}{m} \frac{k}{k-m} \bar{L}_{k-m}(i) \right] + C_{63}[\bar{Tr}]_P [i\bar{f}(i) - \bar{L}_k(i)] \\ + C_{102}(\bar{\Phi}) [(i)\bar{f}(i) - \bar{L}_k(i)] + \frac{C_4}{(\bar{v})} [(i+1)(i)\bar{L}_k(i+2) \\ - (i)(i-1)\bar{L}_k(i)] + \frac{1}{\theta} [\bar{L}_k(i) \text{ feed} - \bar{L}_k(i)] \\ + C_7 [\bar{I}]_w \frac{\bar{m}}{(\bar{F}_{2/3} + \bar{m})} \delta_{i1} \quad k=0 \text{ to } n; i=0 \text{ to } m \quad (4.12)$$

Dead Polymer Chain Length Distribution:

$$\begin{aligned}
 0 = & \frac{c_7 [\bar{I}]_w (\bar{v})^{2/3}}{(\bar{F}_{2/3} + \bar{m})} [\bar{G}_k(i-1) - \bar{G}_k(i)] \\
 & + \frac{c_3}{(\bar{v})^{1/3}} [(i+1) \bar{G}_k(i+1) - (i) \bar{G}_k(i)] + c_2 [\bar{I}]_p (\bar{v}) \\
 & [\bar{G}_k(i-2) - \bar{G}_k(i)] + \frac{c_4}{(\bar{v})} [(i+2)(i+1) \bar{G}_k(i+2) \\
 & - (i)(i-1) \bar{G}_k(i)] + \frac{c_{403}}{(\bar{v})} [\bar{f}(i+2) \sum_{p=0}^{k-1} \frac{l_k}{l_{k-p} l_p} g_p(i+2) g_{k-p}(i+2)] \\
 & + c_{102}(\bar{\varphi}) [\bar{L}_k(i)] + \frac{c_{404}}{(\bar{v})} [(i+1) \bar{L}_k(i+2)] + c_{63} [\bar{Tr}]_p \bar{L}_k(i) \\
 & + \frac{1}{\bar{\theta}} [\bar{G}_k(i) \Big|_{\text{feed}} - \bar{G}_k(i)]
 \end{aligned}$$

(4.15)

k=0 to n
i=0 to m

Emulsifier Balance:

$$0 = \frac{1}{\bar{\theta}} [\bar{S}_T] \Big|_{\text{feed}} - [\bar{S}]_T \quad (4.16)$$

Initiator Balance:

$$0 = -[\bar{I}]_w + \frac{c_{56}}{\bar{\theta}} [[\bar{I}]_w] \Big|_{\text{feed}} - [\bar{I}]_w \quad (4.17)$$

Monomer Balance:

$$0 = \frac{1}{\bar{\theta}} [\bar{M}]_T \Big|_{\text{feed}} - [\bar{M}]_T + c_{53}(\bar{\varphi}) \bar{F}_o \bar{i}_{\text{avg}} \quad (4.18)$$

Micelle Balance:

$$0 = -\bar{m} + c_{50} [\bar{S}]_T - c_{51} - \bar{F}_{2/3} - c_{52} \bar{m}_d \quad (4.19)$$

Monomer Droplets Balance:

$$0 = -\bar{m}_d + c_{54}[\bar{M}]_T - c_{55} \cdot \bar{F}_1 \cdot (\bar{\varphi}) - c_{556}(\bar{m}) \quad (4.20)$$

Expression for $(\bar{\varphi})$:

$$0 = (1-\bar{\varphi}) + \ln(\bar{\varphi}) + \frac{1}{2} (1-\bar{\varphi})^2 + c_{65} \frac{\bar{F}_0}{\bar{F}_{1/3}} \quad (4.21)$$

It should be noted that the expressions 4.12, 4.14, 4.15 are functions of k and i . Similarly the expression 4.13 is a function of k . Thus, these expressions indeed represent sets of equations depending on different values of k and i .

The particle size distribution is continuous and the distribution in n can be assumed to be similar to a continuous distribution. Hence the different moments are used in the aforesaid sets of equations to represent them. However, it can be seen from these expressions that different fractional moments with respect to volume (i.e. $\bar{F}_{1/3}$ and $\bar{F}_{2/3}$) need to be evaluated separately. This can be done by expressing the differential distribution in volume in terms of first 5 integer moments as described previously in equations 4.4 to 4.10. Any fractional moment \bar{F}_t can then be calculated from the differential distribution. The resulting equation expressing the fractional moment in terms of first 5 integer moments is given by

$$\bar{G}_t^x = \frac{|\alpha + t + 1|}{|\alpha + 1|} + \frac{1}{|\alpha + 1|} \sum_{n=3}^m c_n^\alpha \sum_{k=0}^n (-1)^k \times \frac{|\alpha + 1|}{|n - k + 1|} \frac{|\alpha + t + k + 1|}{|\alpha + k + 1|} \frac{1}{|k + 1|} \quad (4.22)$$

The distribution in i , however, cannot be approximated to behave like a continuous distribution, as explained previously. Hence, the equations involving i (i.e. 4.12, 4.14 and 4.15) are solved for different discrete values of i . These equations, however, involve advanced terms calculated at $(i+1)$ and $(i+2)$. Thus the set of n equations will involve $n+2$ variables. However, as stated earlier, the number of radicals in a particle is generally limited to a small value. In other words, after a sufficiently high value of i the terms evaluated at $(i+1)$ and $(i+2)$ tend to become zero. Thus the advanced terms requiring evaluation of $(i+1)$ and $(i+2)$ can be approximated to zero and the set of n equations can be solved for remaining n variables.

Method of Solution:

The modelling equations given earlier can be divided into 3 groups. The first group of equations contains various material balance equations, the equation for $(\bar{\phi})$, the equations involving moments of particle size distribution and the equations involving radical number distribution. Equations 4.13, 4.14, 4.18, 4.19, 4.20 and 4.21 are interdependent and should be solved simultaneously. Equations 4.16 and 4.17 can be solved separately. The solution of this first group of equations gives the values of the various dimensionless concentration terms, radical number distribution and moments of particle size distribution. Using these values the second

group of equations involving moments of live polymer chain length distribution can be solved to obtain $\bar{f}_k(i)$ for different values of i and k . These answers can further be utilized to solve the third group of equations involving dead polymer chain length distribution to obtain $\bar{G}_k(i)$ for different values of i and k .

The basic aim of solving these modelling equations was to evaluate the properties of the output stream from the reactor. These properties include conversion, rate of polymerization in the reactor, number of particles per cm^3 of emulsion, particle size distribution and the molecular weight of the resultant polymer. The conversion, rate and number of particles per cm^3 emulsion can be calculated by knowing the values of $[\bar{M}]_T$ and $[\bar{F}_0]$.

The expressions to obtain these are given below:

$$X = \frac{[\bar{M}]_{\text{feed}} - [\bar{M}]_T}{[\bar{M}]_{\text{feed}}} \quad (4.23)$$

$$\text{Rate} = \frac{[\bar{M}]_{\text{feed}} - [\bar{M}]_T}{\bar{\theta}} \times \frac{[\bar{I}_0]_w k_p d_m}{N_A \cdot d_p \cdot v_m} \quad (4.24)$$

$$N_0 = [\bar{F}_0] \cdot m_0 \cdot N_A \quad (4.25)$$

The particle size distribution of the latex can be generated from its moments \bar{F}_k by using equation 4.22. The average volume and average diameter of these latex particles can also be evaluated from these moments.

$$V_{avg} = \frac{\overline{\overline{F}}_1}{\overline{\overline{F}}_0} \quad (4.26)$$

$$D_{number} = \frac{\overline{\overline{F}}_1^{1/3}}{\overline{\overline{F}}_0} \times \left(\frac{6}{\pi}\right)^{1/3} \quad (4.27)$$

$$D_{vol} = (V_{avg} \times \frac{6}{\pi})^{1/3} \quad (4.28)$$

The number average as well as weight average molecular weight and the polydispersity index can be calculated from moments of dead polymer chain length distribution. These expressions are given below:

$$M_n = \frac{\overline{\overline{GT}}_1}{\overline{\overline{GT}}_0} \quad (4.29)$$

$$M_w = \frac{\overline{\overline{GT}}_2}{\overline{\overline{GT}}_1} \quad (4.30)$$

$$\text{where } \overline{\overline{GT}}_s = \sum_{i=0}^m \overline{\overline{G}}_s(i) \quad (4.31)$$

$$PDI = \frac{M_w}{M_n} \quad (4.32)$$

The entire molecular weight distribution of the resultant polymer can also be generated from these moments by using the procedure outlined earlier in equations 4.4 to 4.10 and 4.22.

Description of Computer Program:

The equations in group 1 are subdivided in two sub-groups, first containing equations 4.16 to 4.21 and 4.13 while the second comprises of equation 4.14. For the solution

of the first subgroup the feed conditions and initial guesses for variables as well as initial guess for \bar{i}_{avg} is needed. This group of equations can then be solved to give values of different dimensionless concentrations and the moments of particle size distribution. These values are then fed to the second subgroup which now becomes a set of linear equations. The starting value of $m=5$ is used. The values of $\bar{f}(m+1)$ and $\bar{f}(m+2)$ are assumed to be zero and the equations are solved to get values of $\bar{f}(i)$; $i=1, m$. An increment of two is then made in m and $(m+2)$ equations are again solved, assuming $\bar{f}(m+3)$ and $\bar{f}(m+4)$ to be equal to zero. The values of $\bar{f}(i)$, $i=1, m+2$ are then obtained. The values of $\bar{f}(m+1)$ and $\bar{f}(m+2)$ are checked to be less than a small constant z (10^{-7} was used). If these values are not less than z an increment of two is again made in m and the set of equations is again solved. The process is repeated until the last two variable values of a set are less than z , thus checking the assumption made in the solution of previous set. The value of \bar{i}_{avg} is then calculated. This improved value of \bar{i}_{avg} is then compared with the previous value of \bar{i}_{avg} and is used together with the values of different variables as the improved guesses to solve the equations in the first subgroup. The process is repeated till the difference between two values of \bar{i}_{avg} is less than a particular convergence criterion.

The equations in the first subgroup are highly nonlinear in nature. The method suggested by Brown has been employed

to solve these equations. The details of this method were taken from reference [31]. It has been found that it is not possible to give good initial guesses for these equations, without which the method does not tend to converge to the solution. Because of this reason an objective function consisting of sums of absolute values of RHS of the aforesaid equations is formed. This objective function is then minimized with respect to the aforesaid variables using the multivariable constrained optimization technique of Box. The details of this technique are given in references [32] and [33]. Relatively worse initial guesses can be used for this optimization procedure. The values of different variables from the optimization procedure are then used as the initial guesses for solving the equations of first group. The solution of linear equations is done by standard matrix inversion procedure. The subprogrammes needed for solution of this group are NONLIN, BACK, AMAX1, AMIN1, FUNX, CONSX, CHECH, CENTR, FUNC, CONST, MATIN, SET1, LINEAR, GOLDEN, CONFR and OBJECT.

The answer from first group of equations are then used for the solution of equations in second group. While solving the radical number distribution (i.e. second subgroup of group 1), the assumption made for solving a set of equations was immediately checked by solving an increased number of equations. Hence when the convergence was ultimately achieved, the assumptions made in the solution of last but one set of

equations (containing say, 'p' equations) was verified by the solution of last set of equations. Hence these 'p' equations should, in principle, be solved for live and dead polymer chain length distribution. However, the solution of live polymer chain length distribution for $f(o)$ type of particles is meaningless. Hence 'p-1' equations are solved for live polymer chain length distribution. However, for dead polymer chain length distribution 'p' equations need to be solved. The equations are linear and are solved by the same method employed earlier. The subprogrammes required for this part are SET2, MATIN and CONFR.

The answer from 2nd group of equations is then used to solve the equations in 3rd group. The method of solution of these equations is similar to that employed for the 2nd group. The answers from this part give the values of $\bar{G}_k(i)$ for different values of i and k. The subprogrammes required for this portion are SET3, MATIN and CONFR.

After the solution of modelling equations various properties of output stream are calculated by using equations 4.23-4.32. Cumulative particle size distribution, live polymer chain length distribution, molecular weight distribution are then generated. The subprogrammes required for this portion include DISTR, DIFF2, CONFR and QG10.

The diagram showing the scheme to solve the modelling equations is given on the next page.

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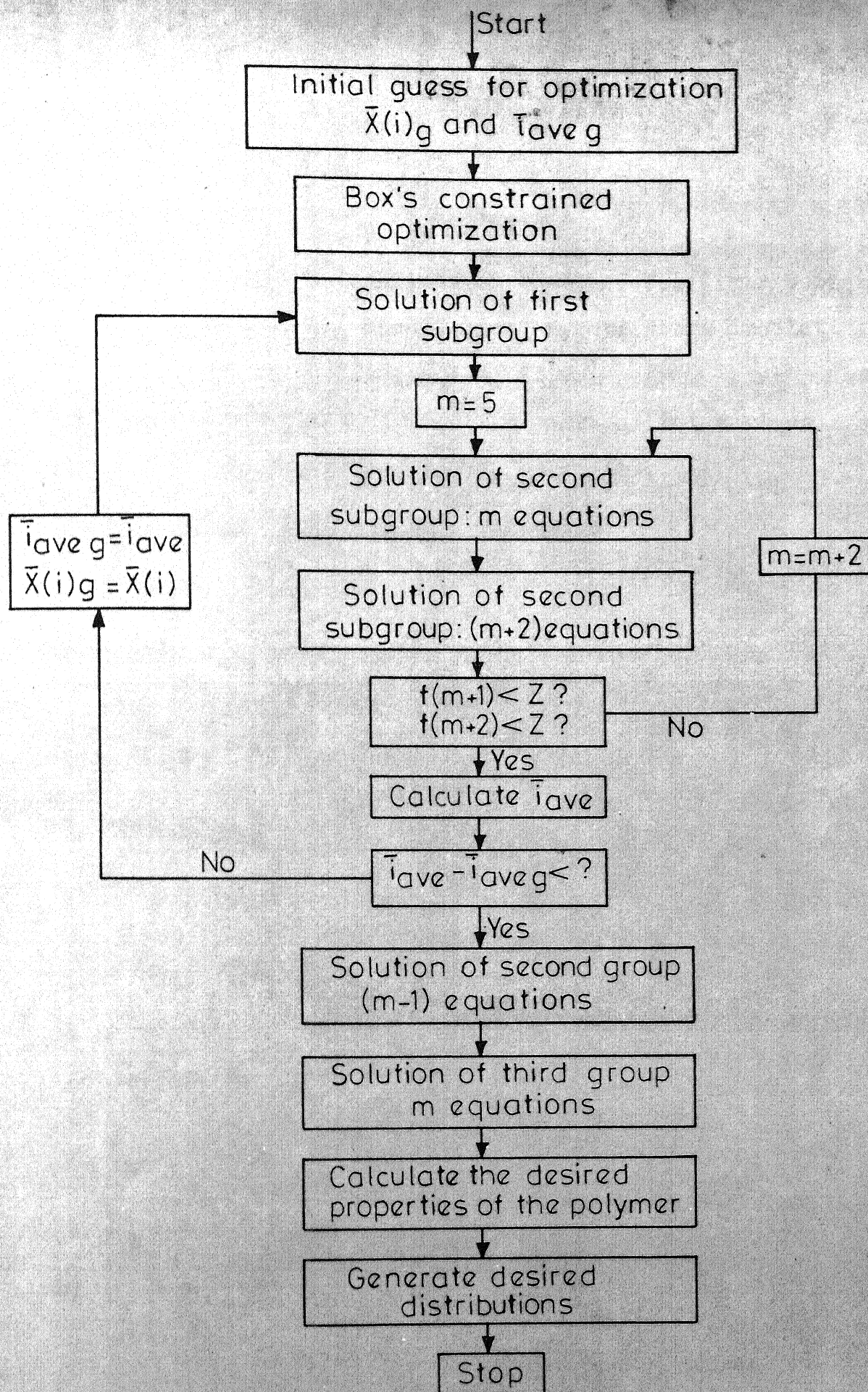


Fig. 1 -Scheme for solution of modelling equations.

NOMENCLATURE

- (1) m_o = nondimensionalization constant for distribution functions, micelle concentration and droplet concentration, moles/lit.
- (2) $[I_o]_w$ = nondimensionalization constant for various concentration terms, moles/lit.

Definitions of Dimensionless Variables:

1. $[\bar{S}]_T = \frac{[S]_T}{[I_o]_w}$
2. $[\bar{I}]_w = \frac{[I]_w}{[I_o]_w}$
3. $[\bar{M}]_T = \frac{[M]_T}{[I_o]_w}$
4. $[\bar{m}] = \frac{m}{m_o}$
5. $(\bar{m}_d) = \frac{m_d}{m_o}$
6. $\bar{f}(i) = \frac{f(i)}{m_o}$
7. $\bar{L}(n,i) = \frac{L(n,i,v_{avg})}{m_o}$
8. $\bar{\bar{L}}_k(i) = \sum_{n=1}^{\infty} n^k \bar{L}(n,i)$
9. $\bar{\bar{F}}_k = \bar{v}^k \bar{F}(\bar{v}) d\bar{v}$
10. $\bar{F}(\bar{v}) = \frac{F(v)}{m_o}$

$$11. \bar{G}(n,i) = \frac{G(n,i,v_{avg})}{m_o}$$

$$12. \bar{G}_k(i) = \sum_{n=1}^{\infty} n^k \bar{G}(n,i)$$

Defintions of Dimensionless Constants:

$$1. C_7 = 2 \cdot k_d \cdot N_A \cdot d_P(I_o)_w \cdot v_m \cdot v_w \cdot f / (K_P \times d_m \times m_o)$$

$$2. C_2 = \frac{k_{iP}}{k_P} N_A^2 v_m^2 [I_o]_w \cdot \frac{d_P}{d_m} \cdot f$$

$$3. C_3 = \frac{k_o}{k_P} \cdot N_A \cdot \frac{d_P}{d_m} \cdot (v_m)^{2/3} (36\pi)^{1/3}$$

$$4. C_4 = \frac{k_{te} + k_{td}}{2k_P} \cdot \frac{d_P}{d_m}$$

$$5. C_{50} = \frac{[I_o]_w}{(36\pi)^{1/3}} \frac{CE}{(36\pi)^{1/3} \cdot v_m^{2/3} m_o \cdot N_A}$$

$$6. C_{51} = \frac{S_{WC} CE}{(36\pi)^{1/3} \cdot v_m^{2/3} m_o} \cdot \frac{1}{N_A}$$

$$7. C_{52} = \frac{v_d^{2/3}}{v_m^{2/3}}$$

$$8. C_{53} = \frac{N_A \cdot v_m \cdot d_P \cdot m_o}{[I_o]_w M_w}$$

$$9. C_{54} = \frac{[I_o]_w V_{mol}}{N_A \cdot v_d \cdot m_o}$$

$$10. C_{556} = \frac{[M]_m \cdot v_m \cdot V_{mol}}{v_d}$$

$$11. C_{55} = \frac{v_m}{v_d}$$

$$12. C_{56} = \frac{k_P \cdot d_m}{k_d \cdot N_A \cdot d_P \cdot v_m}$$

$$13. \quad c_{63} = \frac{k_{ft}}{k_P} \cdot [I_o]_w \cdot N_A \cdot \frac{d_P}{d_m} \cdot v_m$$

$$14. \quad c_{65} = \left(\frac{4\pi}{3v_m} \right)^{1/3} \frac{2 V_{mol} \cdot \gamma}{RT}$$

$$15. \quad c_{101} = \frac{d_P \cdot v_m \cdot N_A}{d_m \cdot V_{mol}}$$

$$16. \quad c_{102} = \frac{k_{fm} N_A d_P v_m}{k_P d_m V_{mol}}$$

$$17. \quad c_{403} = \frac{k_{tc}}{2k_P} \cdot \frac{d_P}{d_m}$$

$$18. \quad c_{404} = \frac{k_{td}}{2k_P} \cdot \frac{d_P}{d_m}$$

CHAPTER 5

RESULTS AND DISCUSSION

The modelling equations developed in the earlier chapters have been solved by using the approach outlined in the previous chapter. The model has been applied for the reaction conditions reported by DeGraff and Poehlein [21]. Their work is one of the most readily available and comprehensive work on emulsion polymerization in a CSTR. The experimental set-up used by these authors consisted of a jacketed stirred tank reactor fitted with a temperature controller. Nitrogen atmosphere was maintained inside the reactor. The reactants were fed in two different streams; first stream comprising of monomer (styrene) and the other comprising of aqueous solution of initiator (Ammonium persulfate) and surfactant (sodium lauryl sulfate). The effluent stream was collected in a stirred vessel containing hydroquinone inhibitor and was then used for experimental measurements. They have also developed a theoretical model, using residence time distribution (RTD) approach, to obtain particle size distribution, rate of polymerization, number of particles and number average molecular weight of the polymer product. They have carried out experiments at various reaction conditions and have tried to compare the results with the values obtained from their model. Computations using the model developed here will be compared with their results.

Various constants used for the solution of the modelling equations are listed on the next page.

As has been stated earlier the chain transfer to transfer agent and the initiation in particles have been neglected in the present work. These mechanisms can, in general, be neglected if the monomer feed is free from transfer agent or trace impurities and if it is insoluble in water. Hence the values of k_{ft} and k_{ip} are assumed to be zero. k_{td} is also assumed to be zero, since for styrene the contribution due to termination by disproportionation is usually negligible compared to that with termination by combination [37]. It has been found from earlier studies on batch systems that desorption from particles play a relatively small part in emulsion polymerization of styrene [15]. Hence the value of k_o is assumed to be zero. It has been reported that the micelles swell and approximately double their size when saturated with monomer [36]. This implies that approximately 75 per cent of the volume of a saturated micelle is filled with monomer which results in the value of $[M]_m$ as 0.0063.

The reaction was carried out by DeGraff and Poehlein [21] at several temperatures and surfactant concentrations. Only the runs at 70°C and with a surfactant concentration of 2.79 gms/100 gms water were selected for simulation. The monomer to water ratio in the reactor feed was 0.54. From this ratio, V_w is calculated to be 0.6494. The value of $(\bar{\varphi})$ was

	Constant	Units	Value	Ref.No.
1.	k_p	$\text{cm}^3/\text{mole}\cdot\text{sec}$	2.650×10^5	[21]
2.	k_o	cms/sec	0.0	assumed
3.	k_d	$1/\text{sec}$	2.33×10^{-5}	[21]
4.	k_{iP}	$1/\text{sec}$	0.0	assumed
5.	k_{td}	$\text{cm}^3/\text{mole sec}$	0.0	assumed
6.	k_{tc}	$\text{cm}^3/\text{mole sec}$	7.2×10^9	[21]
7.	k_{ft}	$\text{cm}^3/\text{mole sec}$	0.0	assumed
8.	k_{fm}	$\text{cm}^3/\text{mole sec}$	21.2	[34]
9.	d_p	gms/cm^3	1.0526	[21]
10.	d_m	gms/cm^3	0.8780	[21]
11.	v_m	cm^3	65.4498×10^{-21}	[13]
12.	V_{mol}	$\text{cm}^3/\text{gm}\cdot\text{mole}$	118.45	[21]
13.	v_d	cm^3	4.1888×10^{-12}	assumed
14.	M_w	gm/mole	10^4	-
15.	$[M]_m$	moles/cm^3	0.0063	[36]
16.	CE	cm^2/mole	3.672×10^9	[21]
17.	γ	dynes/cms	7.50	[10]
18.	ψ	dimensionless	0.256	[10]
19.	S_{wc}	$\text{moles}/\text{cm}^3 \text{ water}$	8.9×10^{-6}	[35]

reported by DeGraff and Poehlein [21] to be 0.615 and has been used for most of the simulations.

Calculations were made using the aforesaid constants at various reaction conditions and the results were compared with the theoretical predictions and experimental results obtained by DeGraff and Poehlein [21].

Particle Size Distribution:

The cumulative particle size distributions calculated at different conditions have been presented in Figures 2 and 3. It can be seen from these plots that the results are practically identical to the theoretical results obtained by DeGraff and Poehlein.

As can be seen from the figures, the breadth of the particle size distribution increases with increasing residence time. With the increase in the mean residence time, the distribution of times spent in the reactor broadens. Further the rate of volume growth (r_v) will also increase due to the increase in the value of \bar{i}_{avg} . Both these effects cause a broader particle size distribution.

In the RTD approach it is assumed that a fraction of the micelles fed to the reactor do not get activated and the rest of the micelles are converted into particles at the same instant. Hence all the particles should have the same birth time. The RTD approach followed by DeGraff and Poehlein then requires the knowledge of only two mechanisms; (1) kinetics of volume growth of particles and (2) RTD of the reactor.

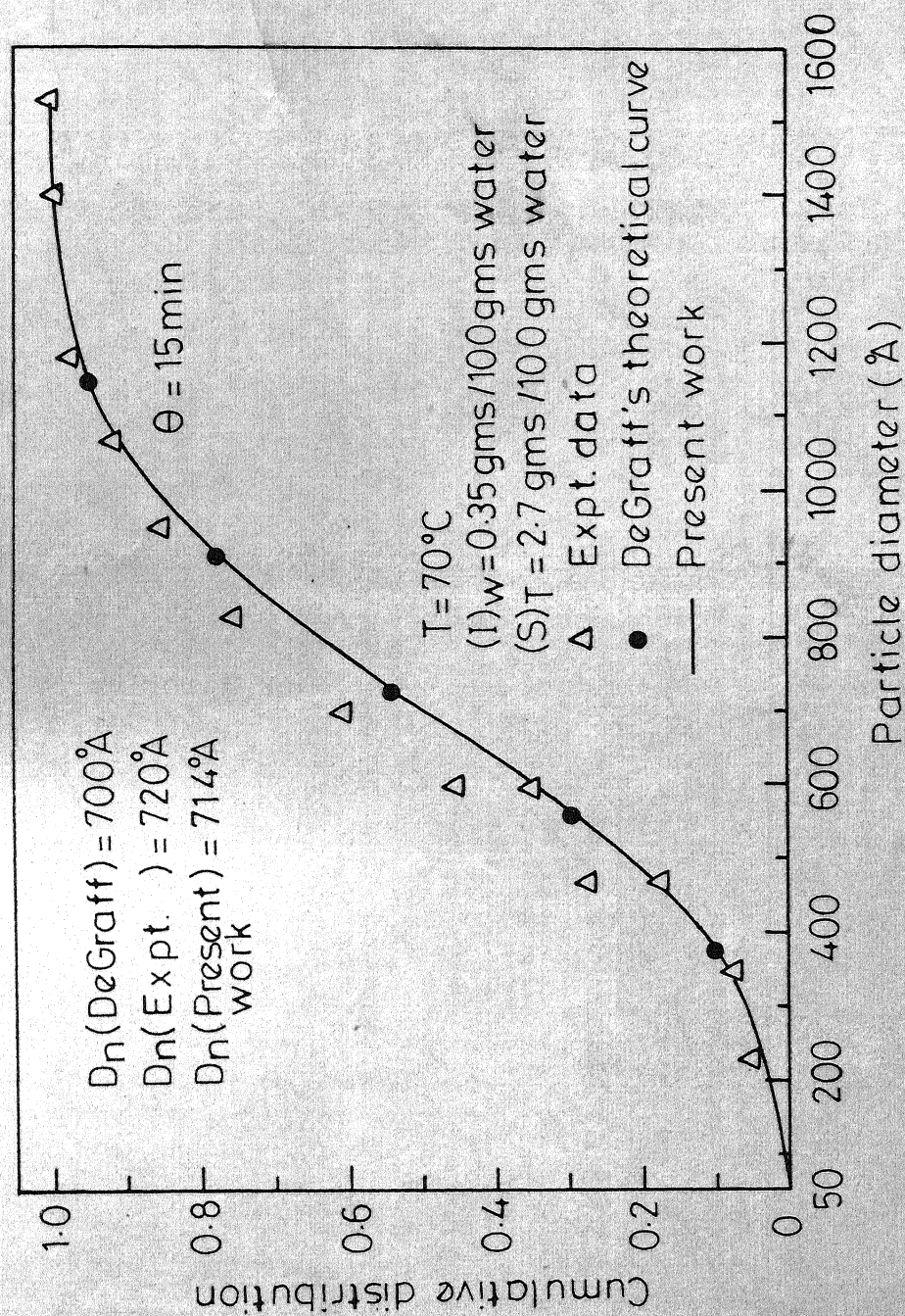


Fig. 2 - Particle size distribution.

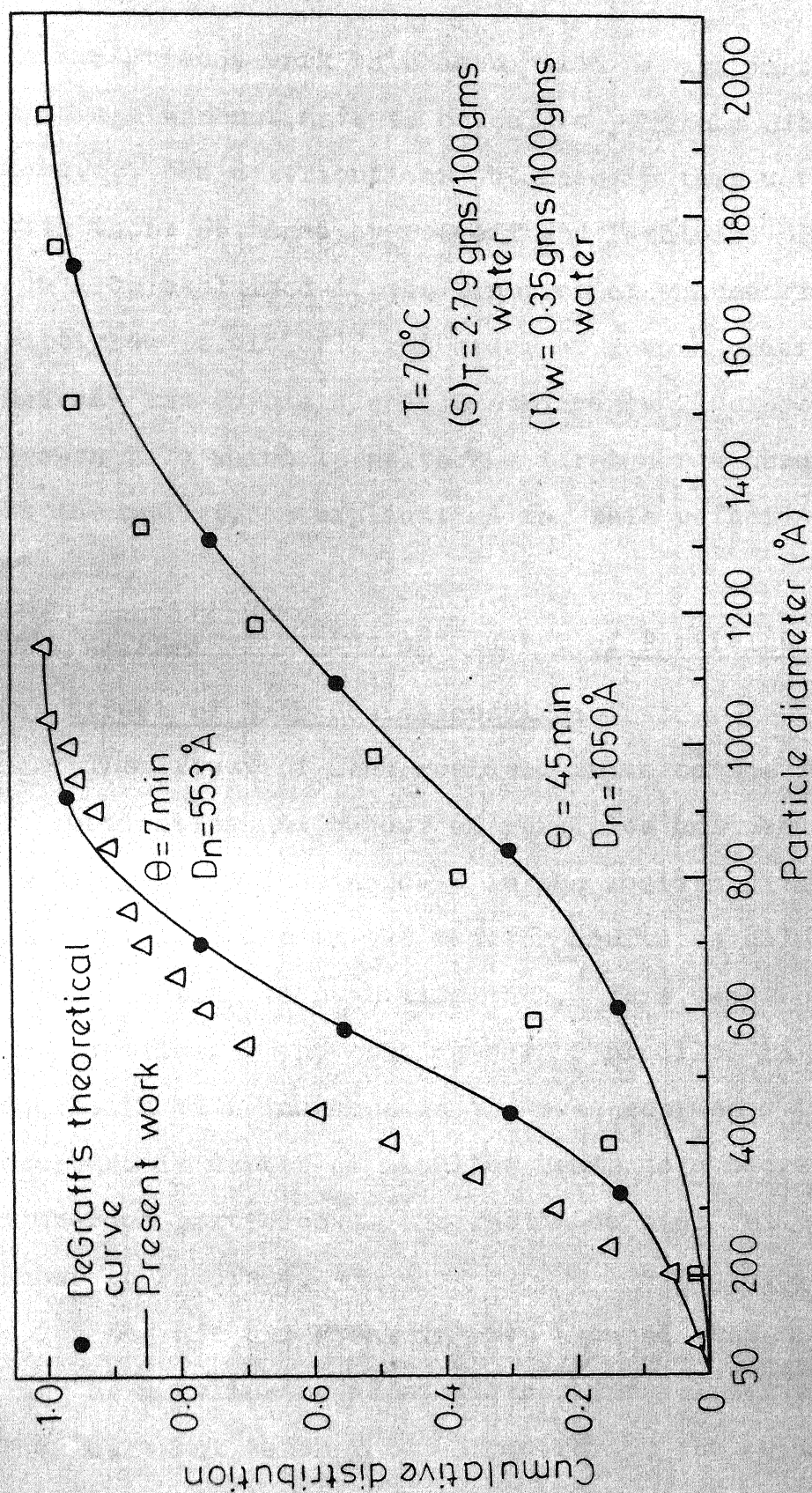


Fig. 3 - Effect of mean residence time on particle size distribution.

In the present work this assumption of all particles being of the same age was not made to calculate particle size distribution. However, the distributions obtained in this work coincide with those obtained by DeGraff and Poehlein. Such coincidence could be explained if the kinetics of volume growth of particles is either first order or zeroth order. However, DeGraff and Poehlein used Stockmayer's [2] expression for growth rate which is neither a first nor a zeroth order process. At the moment, an explanation for this coincidence cannot be offered.

Rate of Polymerization and Number of Particles:

(a) Effect of Mean Residence Time:

The effect of mean residence time on the rate of polymerization and number of particles has been presented in Figure 4. With increase in the residence time, the average size of particles in the reactor increases and requires more amount of soap to stabilize them. This soap is supplied by the micelles. Hence the number of micelles in the reactor decreases with increase in the mean residence time. The decrease in number of micelles leads to a decrease in the number of particles as the residence time is increased as shown in Figure 4.

DeGraff and Poehlein have assumed that soap is present only as micelles or as stabilizer for polymer particles. They have not taken into consideration the emulsifier present

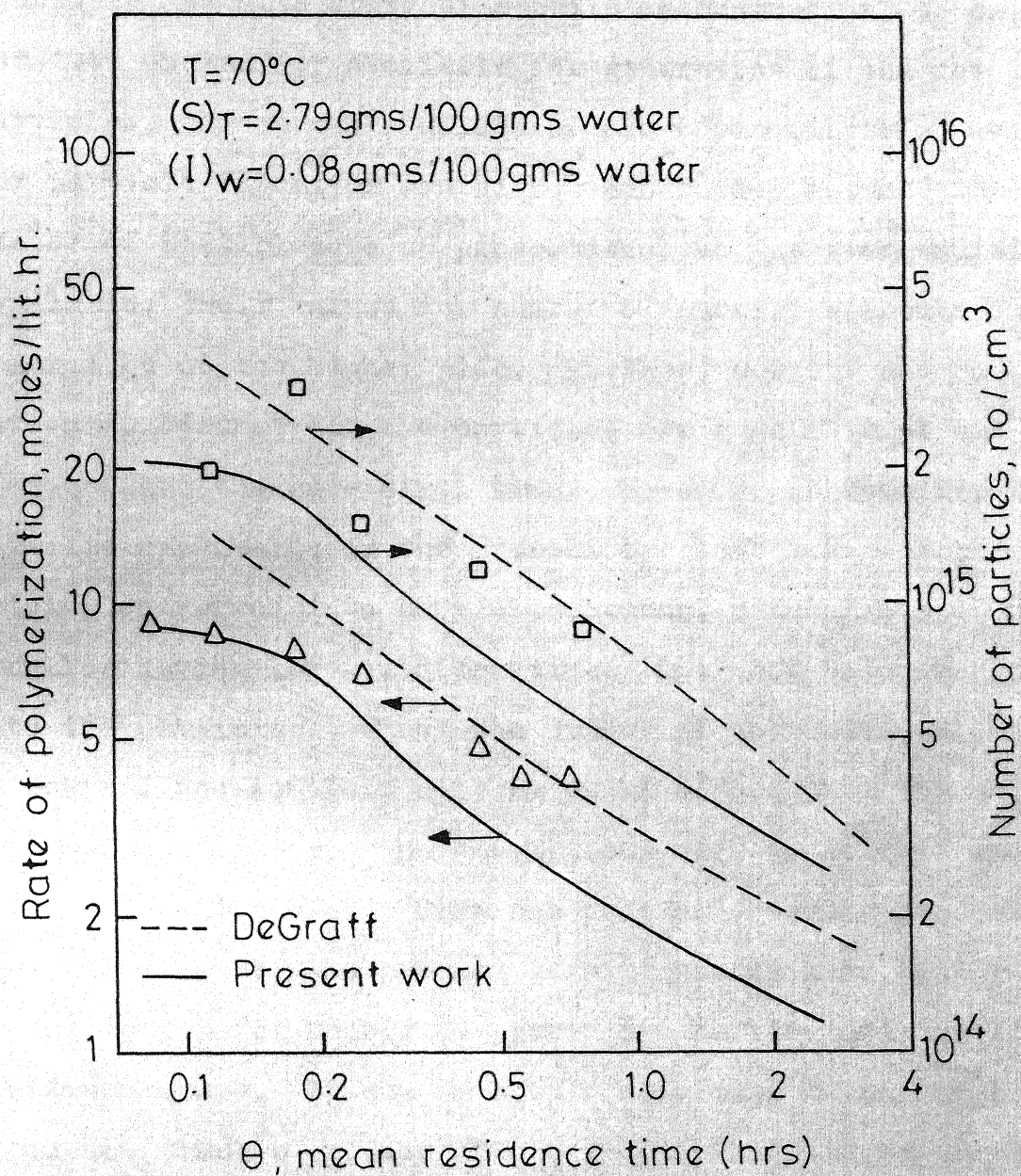


Fig. 4 - Effect of mean residence time on number of particles and rate of polymerization.

on the surface of monomer droplets and that present in aqueous phase upto critical micelle concentration, to calculate the free emulsifier available for generation of the new particles. Hence their values of the free emulsifier available for particle formation are high. Since they assumed the number of particles to be proportional to the free emulsifier available, their values for number of particles should in general be on the higher side. Further, DeGraff and Poehlein have assumed that all the particles are generated at the same time and have the same birth time. However, in reality, the micelles may reside in the reactor for some time before getting converted into particles. Thus, actually, some of the micelles having low residence times, will not be converted into the particles. Hence the number of particles calculated by DeGraff and Poehlein will be still higher than the actual number of particles. The assumptions made by DeGraff and Poehlein, stated above, have not been made in the present work. Therefore, the present results should give better approximation for number of particles, particularly at low residence times. It can be seen from Figure 4 that at low residence times a considerably better fit to the experimental data has been obtained using the present theory, while the results obtained by DeGraff and Poehlein are far from the experimental points. The deviation from theoretical predictions at high residence times is however inexplicable.

The rate of polymerization will be directly proportional to the number of particles times \bar{i}_{avg} as can be seen from 3.28 and 4.24. Hence an explanation similar to that given in the previous paragraph can be given about effect of mean residence time on the rate of polymerization also. The number of particles decreases while the average number of radicals per particle increases with increase in mean residence time. The variation of rate which involves the product of these two terms is given in Figure 4. The almost identical variation of the rate of polymerization and number of particles implies that the average number of radicals in a particle does not vary much with increase in the mean residence time.

(b) Effect of Initiator Concentration:

Figure 5 shows the effect of variation of initiator concentration on M_n and M_w . When the usual values of mean residence time and concentrations of initiator and soap are used the rate of initiation of radicals is considerably high compared to the free emulsifier available. Hence under these reaction conditions the number of particles in the reactor is considerably larger than the number of micelles. When higher concentrations of initiator are used higher rates of generation of radicals are obtained. However, most of these radicals will be captured by the particles since the total area of the particles is considerably larger than the total area of

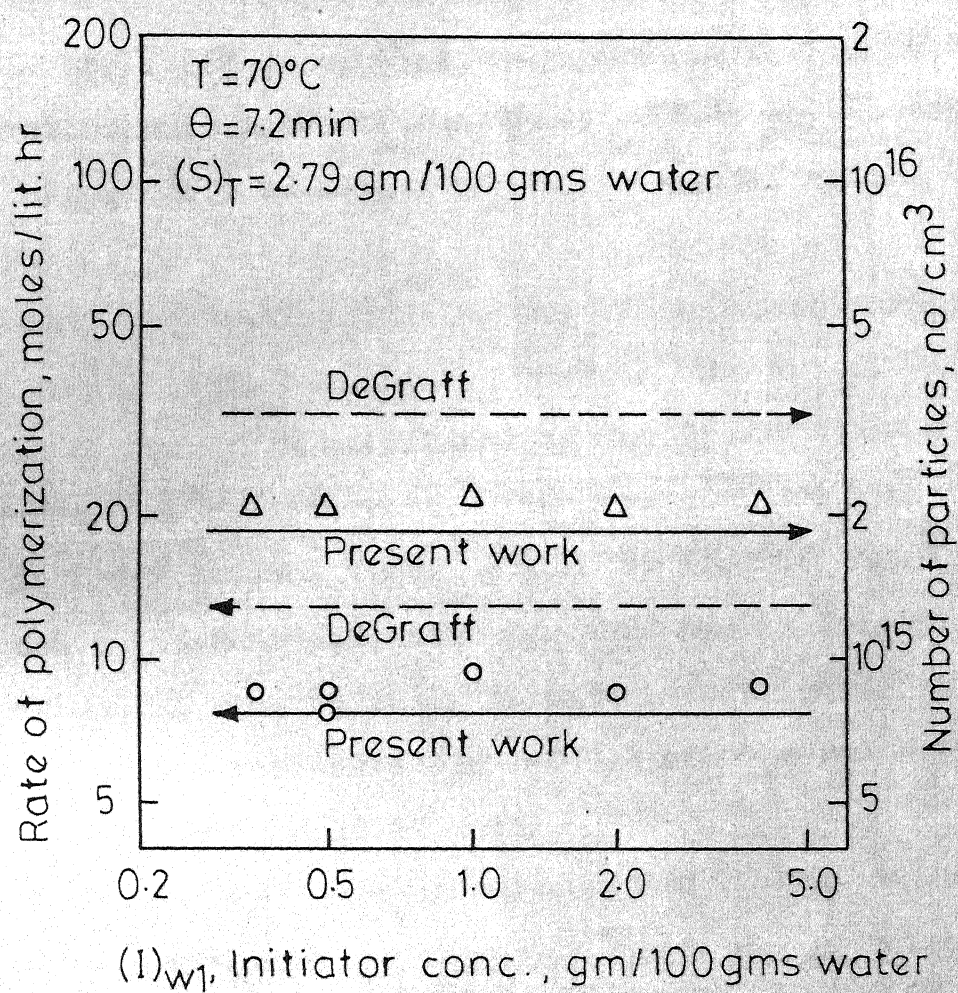


Fig. 5 - Effect of initiator conc. on rate of polymerization and number of particles.

the previous paragraph the value of number of particles is almost constant with respect to initiator concentration. Hence the rate of polymerization, which is proportional to the product of number of particles and the average number of radicals per particle, should not vary significantly with changes in initiator concentration. This behaviour can be seen from Figure 5. The present theory follows the experimental data reasonably well.

Molecular Weights:

(1) Effect of Mean Residence Time:

The effect of mean residence time on M_n and M_w is shown in Figure 6. With increase in mean residence time there is an increase in the particle size and hence the total rate of entry of radicals into particles increases. This decreases the time of growth of radicals inside the particles. Moreover, the average number of radicals present in a particle increases with size leading to the faster termination and lower times of chain growth. Due to these reasons the live chains present inside the particles grow for shorter times resulting in lower molecular weights. This effect can be seen from Figure 6.

DeGraff and Poehlein [21] did not consider the effect due to chain transfer to monomer for calculating M_n . However, this phenomena in general will always be operative and will tend to bring down the molecular weights of the polymer. This

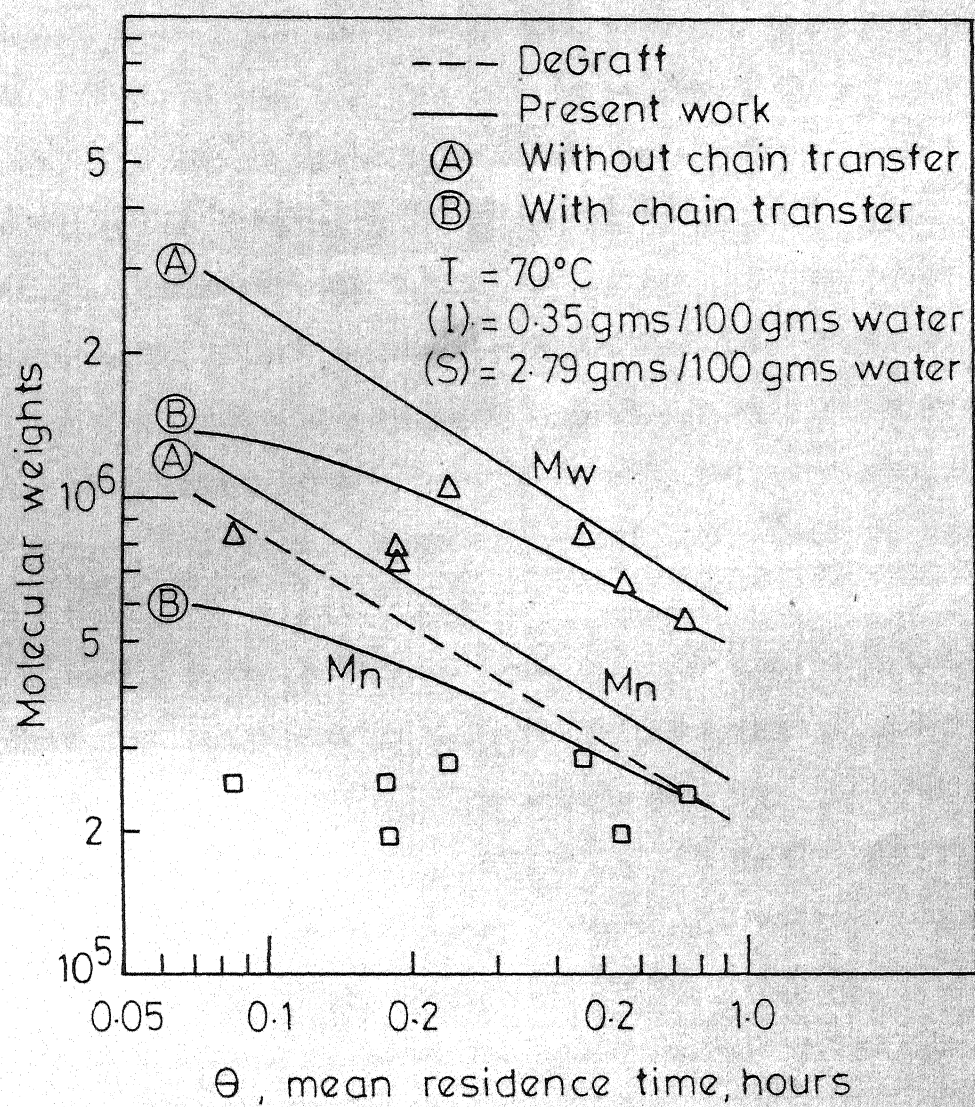


Fig. 6 - Effect of mean residence time on molecular weights.

can be seen from Figure 6 where the variation of molecular weights have been shown with and without considering chain transfer to monomer. The values obtained by the present theory are found to be better approximations to the experimental results than those obtained by DeGraff and Poehlein's theory. However, the values still do not agree with the experimental results satisfactorily at low residence times. Better fit could be obtained by taking into consideration the chain transfer to impurities and the termination by disproportionation. These processes will tend to bring down the molecular weight further.

DeGraff and Poehlein from the RTD approach could not obtain the weight average molecular weight of the product. However, the present theory allows the calculation of M_w from moments of dead polymer chain length distribution. These values are plotted in Figure 6 with and without considering chain transfer to monomer. It can be seen that these values compare fairly well with the experimental results.

(2) Effect of Initiator Concentration:

The effect of initiator concentration on molecular weights has been presented in Figure 7. As the initiator concentration increases the rate of radical absorption in the particles will increase. With the increase in the frequency of radical absorption in a particle, the live chains inside the particles are terminated faster, resulting

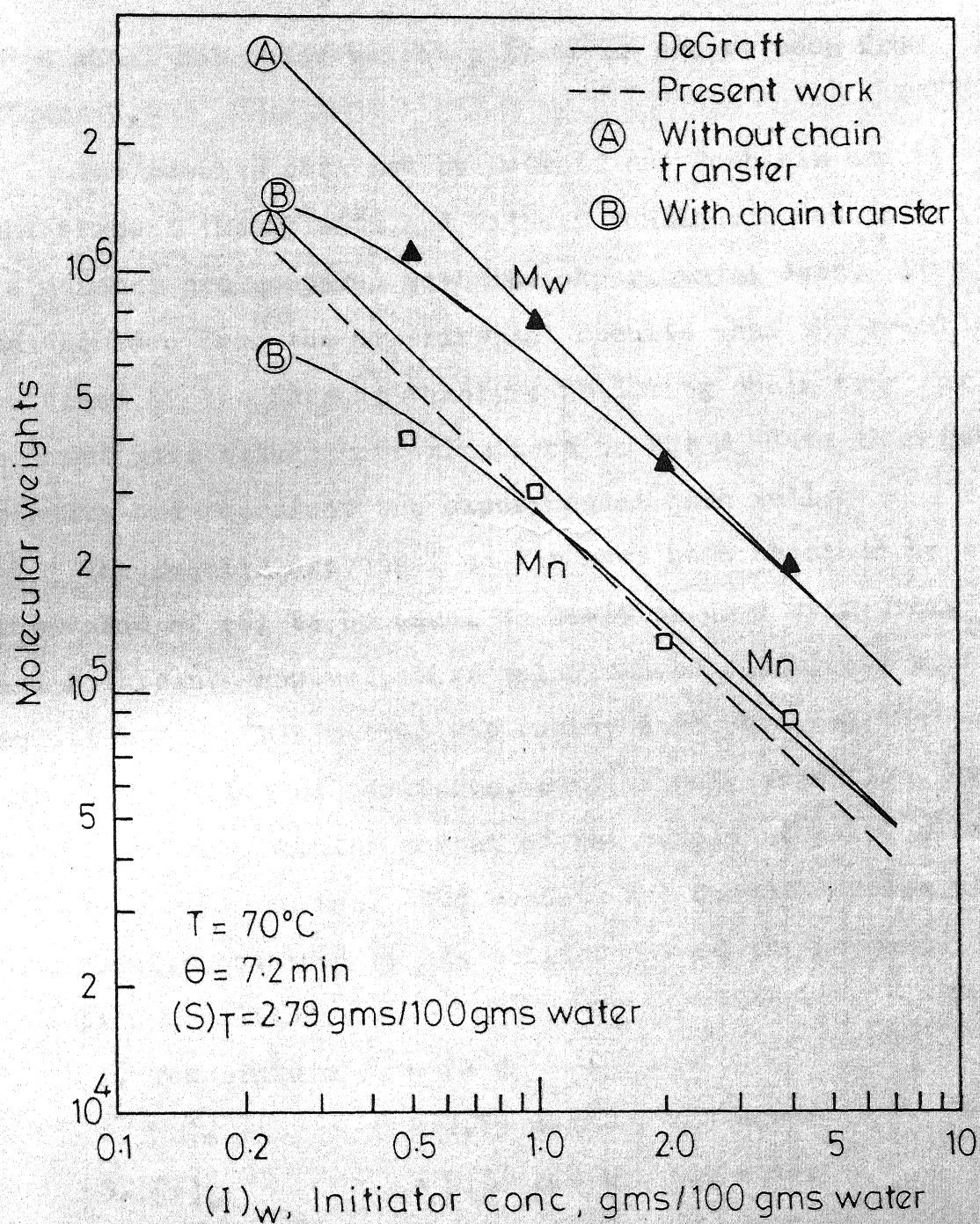


Fig. 7 - Effect of initiator concentration on molecular weights.

in a small molecular weight polymer as can be seen from Figure 7.

The results obtained by DeGraff and Poehlein and by the present theory with and without considering chain transfer to monomer are compared with the experimental data. It can be seen from the experimental results that the results obtained by the present theory considering chain transfer to monomer give better approximations to the M_n data than DeGraff's results and represent the experimental data well.

The results discussed so far have been obtained by using the value of $(\bar{\phi})$ to be equal to 0.615 as used by DeGraff and Poehlein. However, this value can be calculated by using equation 3.31. This method was employed to obtain the values of number of particles, rate of polymerization, average size and the molecular weight of the resultant polymer latex in simulating one run. The results are compared below with the results obtained by the earlier method at the same reaction conditions:

1. Temperature = 70°C
2. $[S]_T$ = 2.79 gms/100 gms water
3. $[I]_w$ = 0.35 gms/100 gms water
4. θ = 15 minutes

	First Approach	Second Approach
1. $\bar{\phi}$	0.615	0.801 (calculated from 4.21)
2. Average diameter (number basis \bar{A})	729	986
3. Average diameter (volume basis \bar{A})	817	1104
4. Number of particles per cm^3 of emulsion	1.09×10^{15}	6.12×10^{14}
5. Rate of polymerization moles/hr	4.7	3.47
6. M_n	5.46×10^5	4.006×10^5
7. M_w	1.32×10^6	9.392×10^5
8. Polydispersity index	2.43	2.344

It can be seen from the table given above that the value of $\bar{\phi}$ obtained by the latex method (i.e. $\bar{\phi} = 0.801$) is different from the one used in earlier method (i.e. $\bar{\phi} = 0.615$). Because of the higher value of $\bar{\phi}$ obtained in the latter method the rate of volume growth would be higher and hence the average particle size obtained is higher. Because of the higher size of particles, they require more soap for stabilization **and hence** less amount of soap is present as micelles. This results in lower value for number of particles by the latter method. Similarly, because of the larger size of the particles the frequency of radical entry in a particle is increased, leading to shorter growth times for the radicals

present in the particle. This results in the lower values of molecular weights as reported in the table.

The aforesaid comparison is made without considering chain transfer to monomer. The comparison was also made by taking chain transfer into consideration. The results are similar to the case where chain transfer is not considered. The molecular weights obtained by both the approaches are, however, smaller than the first case.

	First Approach	Second Approach
1. $\bar{\phi}$	0.615	0.819 (Calculated from 4.21)
2. Average diameter (number basis \AA)	714	986
3. Average diameter _o (volume basis \AA)	799	1104
4. Number of particles per cm^3 emulsion	1.14×10^{15}	6.12×10^{14}
5. Rate of polymerization moles/hr	4.78	3.47
6. M_n	3.925×10^5	3.092×10^5
7. M_w	9.215×10^5	7.074×10^5
8. Polydispersity index	2.347	2.287

Apart from the results stated earlier, the present theory has the merit of ability to predict the entire molecular weight distribution of the polymer product. The molecular weight distribution on weight basis for a particular set of the reaction conditions is plotted in Figure 8.

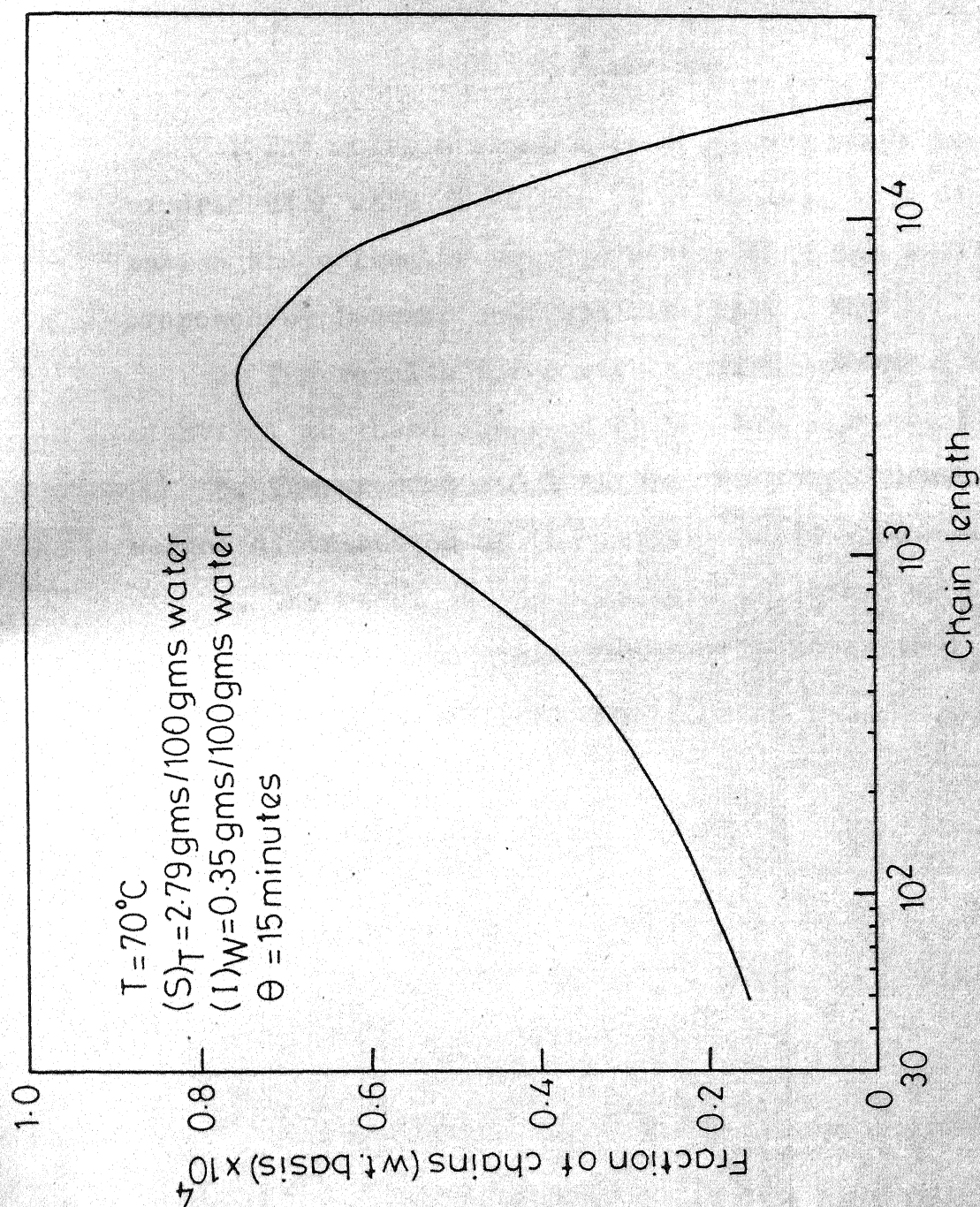


Fig. 8 - Molecular weight distribution of the polymer (wt. basis).

CHAPTER 6

CONCLUSIONS

1. The model presented in this work predicts the experimental data on number of particles, rate of polymerization and molecular weights better than the earlier model proposed by DeGraff and Poehlein [21].

2. The results for particle size distribution are identical to those obtained by the RTD approach [21].

3. The present model allows the complete molecular weight distribution of the polymer product to be evaluated.

4. The model can be made more effective by including the phenomena like chain transfer to impurities and termination by disproportionation.

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00100 C*****
00200 C
00300 C   THIS PROGRAM IS USED TO SIMULATE A CONTINUOUS STIRRED TANK
00400 C   REACTOR FOR EMULSION POLYMERIZATION,THE INPUT DATA FOR THE
00500 C   PROGRAM CONSISTS OF VARIOUS CONSTANTS,FEED CONDITIONS AND
00600 C   VARIOUS REACTOR PARAMETERS,THIS PROGRAM IS ABLE TO CALCULATE
00700 C   THE CONVERSION,RATE OF POLYMERIZATION,AVERAGE PARTICLE SIZE,
00800 C   NUMBER OF PARTICLES,PARTICLE SIZE DISTRIBUTION AND MOLECULAR
00900 C   WT,DISTIBUTION OF THE RESULTANT POLYMER LATEX.
01000 C   INITIALLY THE BOX'S OPTIMIZATION PROCEDURE IS USED TO
01100 C   CALCULATE INITIAL GUESSES FOR THE SOLUTION OF FIRST SUBGROUP,THE
01200 C   EQUATIONS OF FIRST SUBGROUP (I.E.MATERIAL BALANCE EQUATIONS,
01300 C   Particle size distribution and radical number distribution)are
01400 C   THEN SOLVED.THE EQUATIONS OF SECOND GROUP ARE THEN SOLVED TO
01500 C   OBTAIN MOMENTS OF LIVE POLYMER CHAIN LENGTH DISTR.THESE RESULTS
01600 C   ARE THEN USED TO SOLVE THIRD GROUP TO OBTAIN MOMENTS OF DEAD
01700 C   POLYMER CHAIN LENGTH DISTR.VARIOUS OUTPUT VARIABLES ARE THEN
01800 C   CALCULATED AND THE REQUIRED DISTRIBUTIONS ARE GENERATED.
01900 C   THE DETAILS OF THE METHOD OF SOLUTION OF THE MODELLING
02000 C   EQUATIONS IS GIVEN IN THESIS[PAGE      ].
02100 C
02200 C   NOMENCLATURE-----
02300 C   VARIABLES -----
02400 C   XSTART(1,1)-----TOTAL SOAP CONCENTRATION
02500 C   XSTART(1,2)-----INITIATOR CONCENTRATION
02600 C   XSTART(1,3)-----TOTAL MONOMER CONCENTRATION
02700 C   XSTART(1,4)-----MICELLE CONCENTRATION
02800 C   XSTART(1,5)-----MONOMER DROPLETS CONCENTRATION
02900 C   XSTART(1,6)-----MONOMER VOLUME FRACTION IN
03000 C   XSTART(1,7)-----ZEROTH MOMENT W,R,T. VOLUME
03100 C   XSTART(1,8)-----FIRST MOMENT W,R,T. VOLUME
03200 C   XSTART(1,9)-----SECOND MOMENT W,R,T. VOLUME
03300 C   XSTART(1,10)-----THIRD MOMENT W,R,T. VOLUME
03400 C   XSTART(1,11)-----FOURTH MOMENT W,R,T. VOLUME
03500 C
03600 C   XRAD(I)-----RADICAL NUMBER DISTRIBUTION ;NUMBER OF
03700 C   PARTICLES HAVING I RADICALS.
03800 C   XLIVE(K,I)-----MOMENTS OF LIVE CHAIN LENGTH DISTRIBUTION.
03900 C   [K TH MOMENT ; I RADICALS IN PARTICLE.]
04000 C   XDEAD(K,I)-----MOMENTS OF DEAD CHAIN LENGTH DISTRIBUTION.
04100 C   [K TH MOMENT ; I RADICALS IN PARTICLE.]
04200 C   XFD(I)-----FEED CONDITIONS FOR VARIABLES OF FIRST SUB-
04300 C   GROUP.
04400 C   FFD(I)
04500 C   FDLIV(K,I)-----FEED CONDITIONS OF XLIVE(K,I)
04600 C   FDDDED(K,I)-----FEED CONDITIONS OF XDEAD(K,I)
04700 C
04800 C   CONSTANTS-----
04900 C   AKP-----PROPATION RATE CONSTANT.
04900 C   AKD-----RATE CONSTANT FOR DISSOCIATION OF INITIATOR

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5000 C                               IN AQUEOUS PHASE.
5100 C AKIP-----RATE CONSTANT FOR DISSOCIATION OF INITIATOR
5200 C                               IN POLYMER PHASE.
5300 C AKZERO-----DESORPTION RATE CONSTANT.
5400 C AKTC-----RATE CONSTANT FOR TERMINATION BY COMBINATIO
5500 C AKTD-----RATE CONSTANT FOR TERMINATION BY DISPROPOR-
5600 C TIONATION.
5700 C AKFT-----RATE CONSTANT FOR CHAIN TRANSFER TO T
5800 C TRANSFER AGENT.
5900 C AKFM-----RATE CONSTANT FOR CHAIN TRANSFER TO MONOMER
6000 C DP-----DENSITY OF POLYMER.
6100 C DM-----DENSITY OF MONOMER.
6200 C VMOL-----MOLAR VOLUME OF MONOMER.
6300 C SAI-----MONOMER-POLYMER INTERACTION PARAMETER.
6400 C YETA-----SURFACE TENSION.
6500 C AMW-----MOLECULAR WT. OF MONOMER.
6600 C THETA-----MEAN RESIDENCE TIME.
6700 C TEMP-----TEMPERATURE.
6800 C VW-----WATER:EMULSION RATIO.
6900 C AIDW----- (I/O)W ;SEE PAGE 53
7000 C AMZERO----- (M)O ;SEE PAGE 53
7100 C VD-----VOLUME OF DROPLETS.
7200 C VM-----VOLUME OF MICELLES.
7300 C CE-----AREA OCCUPIED BY SOAP.
7400 C AMM-----MONOMER CONCENTRATION IN MICELLES.
7500 C FEE-----EFFICIENCY OF DECOMPOSITION OF THE
7600 C INITIATOR.
7700 C SWC-----CRITICAL MICELLE CONCENTRATION.
7800 C
7900 C
8000 C IPRIN-----PRINTING OPTION : IPRIN=1---INTERMEDIATE
8100 C ITERATIONS ARE PRINTED ;IPRIN=0---ONLY
8200 C DATA AND ANSWER FROM SIMULATION ARE PRINTE
8300 C IPT-----PRINTING OPTION :IPT=0---CONCISE OUTPUT
8400 C IPT=1---DETAILED OUTPUT[ ENTIRE DISTR,
8500 C ARE CALCULATED/PRINTED]
8600 C MX-----INPUT UNIT NUMBER TO BE TYPED BY THE USER
8700 C NX-----RESULT UNIT NUMBER TO BE TYPED BY THE USER
8800 C NX=1,20,21,22,23,24---RESULT ON DISK
8900 C NX=5---RESULT AT THE TERMINAL
9000 C
9100 C
9200 C*****
9300 C DOUBLE PRECISION XSTART(14,16),R(14,11),F(14),G(16),H(16)
9400 C DOUBLE PRECISION XC(11),X(20),XTEMP(6),B(25,1),XRAD(0:20)
9500 C DOUBLE PRECISION T(25,25)
9600 C DOUBLE PRECISION FFD(0:20),FDLIV(0:5,20),FDDED(0:5,0:20),XFD(11)
9700 C DOUBLE PRECISION C50,C51,C52,C53,C54,C556,C56,C7,CT1,C22,C33,C44
9800 C DOUBLE PRECISION XLIVE(0:5,20),XDEAD(0:5,0:20),XCORD(0:50)
9900 C DOUBLE PRECISION YCORD(0:50),GSMAL(0:5,20),XLIVET(0:5)

```

```

10000 DOUBLE PRECISION C79,C78,C65,YETA,XZER,DASHI,DASHI1,DASHI2
10100 DOUBLE PRECISION DIFF,ONETH,TWOTH,TRP,PHI,AM,AIP,SUM1,SUM2
10200 DOUBLE PRECISION BRACE1,BRACE2,BRACE3,SAI
10300 DOUBLE PRECISION THETR,DUMMY,THETVE,V,ANUMBR,AIW,XL1,PART1
10400 DOUBLE PRECISION ZCORD(0:50),XVOL(0:5),XDEADT(0:5),XCUM(0:50)
10500 DOUBLE PRECISION X01,X02,C,DAK,X66,X667,XMAX
10600 INTEGER GAMMA
10700 ACCEPT *,MX
10800 ACCEPT *,NX
10900 C TO READ INPUT PARAMETERS FOR CONSTRAINED OPTIMIZATION
11000 READ(MX,*) NOPT,M,K,ITMAX,IPRIN
11100 READ(MX,*) ALPHA,BETA,DELTA,GAMMA,DASHI
11200 READ(MX,*) (XSTART(1,J),J=1,NOPT)
11300 DO 100 II=2,K
11400 100 READ(MX,*) (R(II,J),J=1,NOPT)
11500 C TO READ INPUT PARAMETERS FOR LINEAR EQUATIONS AND
11600 C CONVERGENCE CRITERION.
11700 READ(MX,*) MAXIT,NUMSIG,IPT,NUMEQ
11800 READ(MX,*) EPS,DELNON,PREC,ZETA
11900 C TO READ INPUT CONSTANTS FOR THE REACTOR.
12000 C
12100 C RATE CONSTANTS-----
12200 READ(MX,*) AKP,AKD,AKZERO,AKIP,AKTD,AKTC,AKFT,AKFM
12300 C MONOMER-POLYMER CHARECTARISTICS----
12400 READ(MX,*) AMW,DP,DM,VMOL,SAI
12500 C REACTOR CHARECTARISTICS----
12600 READ(MX,*) THETA,TEMP,VW
12700 C OTHER CONSTANTS----
12800 READ(MX,*) AIOW,AMZERO,VD,VM,CE,AMM,FEE,SWC,YETA
12900 C TO READ FEED CONDITIONS----
13000 READ(MX,*) (FFD(KI),KI=1,10)
13100 DO 181 KI=0,5
13200 READ(MX,*) (FDLIV(KI,IK),IK=1,10)
13300 181 CONTINUE
13400 DO 182 KI=0,5
13500 READ(MX,*) (FDDED(KI,IK),IK=0,10)
13600 182 CONTINUE
13700 READ(MX,*) (XFD(KI),KI=1,NOPT)
13800 ANA=6.02E+23
13900 RR=8.2764E+07
14000 C
14100 ONETH=1.0/3.0
14200 TWOTH=2./3.0
14300 C CALCULATION OF DIMENSIONLESS CONSTANTS.
14400 BRACE1=(36.0*3.142)**ONETH
14500 BRACE2=VD**TWOTH
14600 BRACE3=VM**TWOTH
14700 C
14800 C2=AKIP*ANA*ANA*VM*VM*AIOW*DP/(AKP*DM)
14900 C3=AKZERO*ANA*DP*BRACE3*BRACE1/(AKP*DM)

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15000 C4=(AKTC+AKTD)*DP/(2.0*AKP*DM)
15100 C50=AIOW*CE/(BRACE1*BRACE3*AMZERO*ANA)
15200 C51=(VW*SWC*CE)/(BRACE1*BRACE3*AMZERO*ANA)
15300 C52=BRACE2/BRACE3
15400 C53=ANA*VM*DP*AMZERO/(AIOW*AMW)
15500 C54=AIOW*VMOL/(ANA*VD*AMZERO)
15600 C55=VM /VD
15700 C556=AMM*VM*VMOL/VD
15800 C56=AKP*DM/(AKD*ANA*DP*VM)
15900 CT1=1.0
16000 C7=2.0*AKD*ANA*DP*AIOW*VM*FEE*VW/(AKP*DM*AMZERO)
16100 THETR=THETA*AKP*DM/(ANA*DP*VM)
16200 DUMMY=1.0E+03
16300 C403=AKTC*DP/(2.0*AKP*DM)
16400 C404=AKTD*DP/(2.0*AKP*DM)
16500 THETVE=1.0/THETR
16600 C63=AKFT*AIOW*ANA*DP*VM/(AKP*DM)
16700 C101=DP*VM*ANA/(DM*VMOL)
16800 C102=AKFM*ANA*DP*VM/(AKP*DM*VMOL)
16900 DUMMY2=1.0E-02
17000 C65=((4.0*3.142/(3.0*VM))*ONETH)*(2.*VMOL*YETA/(RR*TEMP))
17100 DUMMY3=1.0E-03
17200 DUMMY4=4.2985E+03
17300 C*****
17400 C
17500 C THE FOLLOWING LINES SHOULD BE INSERTED FOR SECOND APPROACH.
17600 C X01=0.01
17700 C X02=0.99
17800 C C=0.00001
17900 C WRITE(NX,4189)
18000 C
18100 C*****
18200 C
18300 C PRINTING THE DATA---
18400 WRITE(NX,3185)
18500 WRITE(NX,2001)
18600 WRITE(NX,3185)
18700 WRITE(NX,4181)
18800 WRITE(NX,2002) NOPT,M,K,ITMAX,IPRIN
18900 WRITE(NX,2003) ALPHA,BETA,DELTA,GAMMA,DASHI
19000 WRITE(NX,2004)
19100 WRITE(NX,2005) ((J,XSTART(1,J)),J=1,NOPT)
19200 WRITE(NX,2006)
19300 WRITE(NX,2007) MAXIT,NUMSIG,IPT,NUMEQ
19400 WRITE(NX,2008) EPS,DELNON,PREC,ZETA
19500 WRITE(NX,2009) AKP,AKD,AKZERO,AKIP,AKTD,AKTC,AKFT,AKFM
19600 WRITE(NX,2010)
19700 WRITE(NX,2011) AMW,DP,DM,VMOL,SAI
19800 WRITE(NX,2012)
19900 WRITE(NX,2013) THETA,TEMP,VW

```



```

20000      WRITE(NX,2014)
20100      WRITE(NX,2015) AIOW,AMZERO,VD,VM,CE,AMM,FEE,SWC,YETA
20200      WRITE(NX,3015)
20300      WRITE(NX,3016)((KI,FFD(KI)),KI=1,10)
20400      WRITE(NX,2113)
20500      DO 281 KI=0,5
20600      WRITE(NX,2114) (KI,IK,FDLIV(KI,IK),IK=1,10)
20700  281   CONTINUE
20800      WRITE(NX,2115)
20900      DO 282 KI=0,5
21000      WRITE(NX,2116) (KI,IK,FDDDED(KI,IK),IK=1,10)
21100  282   CONTINUE
21200      WRITE(NX,2487)
21300      WRITE(NX,2488) (KI,XFD(KI),KI=1,NOPT)
21400      C
21500      C
21600  2001   FORMAT(/10X,"DATA FOR THE PROBLEM"/)
21700  2002   FORMAT(/5X,"NOPT = ",I2,5X,"M = ",I2,5X,"K = ",I2,5X,
21800          1"ITMAX = ",I5,5X,"IPRIN = ",I2)
21900  2003   FORMAT(/5X,"ALPHA = ",F10.6,5X,"BETA = ",F10.6,5X,"DELTA = ",
22000          1F10.6//5X,"GAMMA = ",I2,13X,"DASHI = ",F10.6)
22100  2004   FORMAT(/5X,"INITIAL GUESSES-----")
22200  2005   FORMAT(/2(5X,"XSTART(' ,I2,' ) = ",E15.8))
22300  2006   FORMAT(/5X,"INPUT PARAMETERS FOR LINEAR EQUATIONS AND
22400          1THE CONVERGENCE CRITERION")
22500  2007   FORMAT(/5X,"MAXIT = ",I5,5X,"NUMSIG = ",I2,5X,"IPT = ",
22600          1I2/5X,"NUMEQ = ",I2)
22700  2008   FORMAT(/5X,"EPS = ",E15.8,3X,"DELNON = ",E15.8,3X,
22800          1"PREC = ",E15.8/5X,"ZETA = ",E15.8)
22900  2009   FORMAT(/5X,"RATE CONSTANTS-----"/5X,"KP = ",E15.8,5X,
23000          1"KD = ",E15.8,5X,"KZERO = ",E15.8/5X,"KIP = ",E15.8,5X,
23100          1"KTD = ",E15.8,5X,"KTC = ",E15.8/5X,"KFT = ",E15.8,5X,
23200          1"KFM = ",E15.8 )
23300  2010   FORMAT(/5X,"MONOMER POLYMER CHARECTARISTICS---")
23400  2011   FORMAT(/5X,"MW = ",E15.8,5X,"DP = ",E15.8,5X,"DM = ",E15.8/
23500          15X,"VMOL = ",E15.8,5X,"SAI = ",E15.8)
23600  2012   FORMAT(/5X,"REACTOR CHARECTARISTICS---")
23700  2013   FORMAT(/5X,"THETA = ",E15.8,5X,"TEMP = ",E15.8,5X,
23800          1"VW = ",E15.8)
23900  2014   FORMAT(/5X,"OTHER CONSTANTS---")
24000  2015   FORMAT(/5X,"AIOW = ",E15.8,4X,"AMZERO = ",E15.8,4X,"VD = ",
24100          1E15.8/5X,"VM = ",E15.8,4X,"CE = ",E15.8,4X,"AMM = ",
24200          1E15.8 /5X,"FEE = ",E15.8,4X,"SWC = ",E15.8,4X,"YETA = ",E15.8)
24300  3015   FORMAT(/5X,"FEED CONDITIONS-----"/5X,"RADICAL NUMBER DISTRIBUTI
24400          1ON-----")
24500  3016   FORMAT(/2(5X,"FFD(' ,I2,' ) = ",F10.6))
24600  2113   FORMAT(/5X,"LIVE POLYMER CHAINS-----",/)
24700  2114   FORMAT(/3(4X,"FDLIV(' ,I2,1H,,I2,' ) = ",F8.4))
24800  2115   FORMAT(/5X,"DEAD POLYMER CHAINS-----",/)
24900  2116   FORMAT(/3(4X,"FDDDED(' ,I2,1H,,I2,' ) = ",F8.4))

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25000 2487   FORMAT(/5X,'FEED COND. FOR VARIABLES OF INITIAL SET')
25100 2488   FORMAT(/3(5X,I2,4X,E11.4))
25200 4181   FORMAT(/7X,'DATA FOR CONSTRAINED OPTIMIZATION-----')
25300 C*****
25400 C
25500 C       THE FOLLOWING LINE SHOULD BE INSERTED FOR SECOND APPROCH.
25600 C4189   FORMAT(5X,'OUTPUT OBTAINED BY EVALUATING PHI FROM EQUATION
25700 C       4.21)
25800 C
25900 C*****
26000 C
26100 C
26200       CALL CONSX(NOPT,M,K,ITMAX,ALPHA,BETA,GAMMA,DELTA,XSTART,R,F,IT,
26300       1IEV2,NX,G,H,XC,IPRIN,DASHI,ONETH,TWOTH,BRACE1,BRACE2,
26400       1BRACE3,C50,C51,C52,C53,C54,C55,C556,C56,CT1,C7,THETR,DUMMY,XFD,
26500       1C65,SAI)
26600       IF(IT=ITMAX) 20,20,30
26700 20      IF(IPRIN.EQ.0) GO TO 4501
26800       WRITE(NX,2117) F(IEV2)
26900 2117    FORMAT(/5X,'FINAL FUNCTION VALUE =',D18.10,/)
27000       DO 300 J=1,NOPT
27100       WRITE(NX,16)J,XSTART(IEV2, J)
27200 16      FORMAT(2X,'X('I2,') = 'E20.8)
27300 300     CONTINUE
27400 4501    GO TO 899
27500 30      WRITE(NX,17)ITMAX
27600 17      FORMAT(2X,'WARNING MAIN;THE NUMBER OF ITERATIONS HAS EXCEEDED'
27700       12X,I4)
27800 C
27900 C*****
28000 C
28100 899     DO 568 KI=1,NOPT
28200       X(KI)=XSTART(IEV2,KI)
28300 568     CONTINUE
28400 999     DASHI1=DASHI
28500       XZER=X(7)
28600       DO 167 KI=1,4
28700       XTEMP(KI)=X(KI+7)
28800 167     CONTINUE
28900       NNON=NOPT+1
29000       X(NNON)=CONFR(TWOTH,XTEMP,XZER)
29100       CALL NONLIN(NNON,X,NUMSIG,MAXIT,IPRIN,EPS,DASHI1,NX,DELNON,PREC,
29200       1ONETH,TWOTH,BRACE1,BRACE2,BRACE3,C50,C51,C52,C53,C54,C55,C556,
29300       1C56,C7,CT1,C65,DUMMY,THETR,SAI,XFD,X66)
29400       DO 1111 IK=1,NNON
29500       IF(X(IK).LT.0.0) X(IK)=0.0
29600 1111     CONTINUE
29700       V=X(8)*DUMMY4/X(7)
29800       AIW=X(2)
29900       AM=X(4)

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30000      AIP=0.0
30100      ANUMBR=X(7)/DUMMY3
30200      F23=X(12)*(DUMMY4**TWOETH)/DUMMY3
30300      C22=C2*V*AIP
30400      C33=C3/(V**ONETH)
30500      C44=C4/V
30600      C78=C7*(V**TWOETH)*AIW/(AM+F23)
30700      C79=C7*AM*AIW/(AM+F23)
30800      C
30900      C*****
31000      C
31100      C      INSERT THE FOLLOWING LINES IN PLACE OF THE ENCLOSED SET OF LINES
31200      C      GIVEN ABOVE FOR THE SECOND APPROACH.
31300      C
31400      C899      DO 568 KI=1,5
31500      C      X(KI)=XSTART(IEV2,KI)
31600      C568      CONTINUE
31700      C      X667=XSTART(IEV2,6)
31800      C      DO 569 KI=7,NOPT
31900      C      X(KI-1)=XSTART(IEV2,KI)
32000      C      XFD(KI-1)=XFD(KI)
32100      C569      CONTINUE
32200      C999      DASHI1=DASHI
32300      C      XZER=X(6)
32400      C      DO 167 KI=1,4
32500      C      XTEMP(KI)=X(KI+6)
32600      C167      CONTINUE
32700      C      DAK=CONFR(ONETH,XTEMP,XZER)
32800      C      CALL GOLD(X01,X02,C,X667,DAK,XZER,SAI,C65,DUMMY4,ONETH)
32900      C      NNON=NOPT
33000      C      X(NNON)=CONFR(TWOETH,XTEMP,XZER)
33100      C998      X66=X667
33200      C      CALL NONLIN(NNON,X,NUMSIG,MAXIT,IPRIN,EPS,DASHI1,NX,DELNON,PREC,
33300      C      1ONETH,TWOETH,BRACE1,BRACE2,BRACE3,C50,C51,C52,C53,C54,C55,C556,
33400      C      1C56,C7,CT1,C65,DUMMY,THETR,SAI,XFD,X66)
33500      C      DO 1111 KI=1,NNON
33600      C      IF(X(KI).LT.0.0) X(KI)=0.0
33700      C1111      CONTINUE
33800      C      XZER=X(6)
33900      C      DO 168 KI=1,4
34000      C      XTEMP(KI)=X(KI+6)
34100      C168      CONTINUE
34200      C      DAK=CONFR(ONETH,XTEMP,XZER)
34300      C      CALL GOLD(X01,X02,C,XMAX,DAK,XZER,SAI,C65,DUMMY4,ONETH)
34400      C      X667=XMAX
34500      C      DIFF=X667-X66
34600      C      IF(IPRIN.EQ.0) GO TO 4938
34700      C      WRITE(NX,4086) DIFF,X667
34800      C4086      FORMAT(/5X,"DIFFERENCE IN X66 = ",D15.8/5X,"X66 = ",D15.8/)
34900      C4938      IF(ABS(DIFF).LT.0.0001) GO TO 987

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35000 C      GO TO 998
35100 C987    V=X(7)*DUMMY4/X(6)
35200 C      AIW=X(2)
35300 C      AM=X(4)
35400 C      AIP=0.0
35500 C      ANUMBR=X(6)/DUMMY3
35600 C      F23=X(11)*(DUMMY4**TWOHT)/DUMMY3
35700 C      C22=C2*V*IP
35800 C      C33=C3/(V**ONETH)
35900 C      C44=C4/V
36000 C      C78=C7*(V**TWOHT)*AIW/(AM+F23)
36100 C      C79=C7*AM*AIW/(AM+F23)
36200 C
36300 C*****
36400      CALL LIN(NUMEQ,FFD,ANUMBR,THETR,C22,C33,C78,C79,C44,NX,DASHI2,
36500      1NTEMP3,B,IPRIN)
36600      DIFF=DASHI2-DASHI1
36700      IF(IPRIN.EQ.0) GO TO 4502
36800      WRITE(NX,4002) DIFF
36900 4002    FORMAT(/2X,'DIFF ',E15,8/)
37000 4502    IF(ABS(DIFF).LT.ZETA) GO TO 98
37100      DASHI=DASHI2
37200      GO TO 999
37300 98      IF(IPRIN.EQ.0) GO TO 4503
37400      WRITE(NX,2017)
37500 2017    FORMAT(/' FUNCTION VALUES AFTER FIRST SET')
37600      WRITE(NX,2018)((I,X(I)),I=1,NNON)
37700 2018    FORMAT(1(5X,'X(',I2,') = ',E15,8))
37800 4503    DO 187 IK=1,NTEMP3
37900      KI=IK
38000      XRAD(KI-1)=B(KI,1)
38100 187     CONTINUE
38200      IF(IPRIN.EQ.0) GO TO 4504
38300      WRITE(NX,2019)((KI,XRAD(KI)),KI=0,NTEMP3)
38400 2019    FORMAT(/,1(5X'XRAD(',I2,') = ',E15,8))
38500 C      LIVE POLYMER CHAIN LENGTH DISTRIBUTION STARTS.
38600 4504    NTEMP2=NTEMP3-1
38700      PHI=X(6)
38800 C
38900 C*****
39000 C
39100 C      INSERT THE FOLLOWING LINE INSTEAD OF THE PREVIOUS LINE FOR
39200 C      SECOND APPROACH.
39300 C      PHI=X66
39400 C
39500 C*****
39600 C
39700      DO 878 KI=1,NTEMP2
39800      XLIVE(0,KI)=KI*XRAD(KI)
39900 878      CONTINUE

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40000      DO 857 MONT=1,4
40100      MOMENT=MONT
40200      CALL SET2(NTEMP3,C22,C33,C44,C78,C79,C63,C102,C101,TRP,
40300      1MOMENT,PHI,THETVE,FDLIV,XRAD,XLIVE,T,B)
40400      M=1
40500      CALL MATIN(T,NTEMP2,B,M,DETER)
40600      DO 857 KI=1,NTEMP2
40700      XLIVE(MOMENT,KI)=B(KI,1)
40800      857      CONTINUE
40900      DO 862 MONT=0,4
41000      IF(IPRIN.EQ.0) GO TO 4505
41100      WRITE(NX,4250)
41200      4250      FORMAT(/5X,'LIVE POLYMER CHAIN LENGTH DISTRIBUTION-----'/)
41300      WRITE(NX,2499) MONT
41400      2499      FORMAT(/5X,'MOMENT  =',I2)
41500      WRITE(NX,2500)(KI,XLIVE(MONT,KI),KI=1,NTEMP2)
41600      2500      FORMAT(2(5X,I2,5X,E15.8))
41700      862      CONTINUE
41800      C      DEAD POLYMER CHAIN LENGTH DISTRIBUTION STARTS.
41900      4505      DO 911 MOMENT=0,4
42000      DO 911 IRAD=1,NTEMP3
42100      GSMAL(MOMENT,IRAD)=XLIVE(MOMENT,IRAD)/XRAD(IRAD)
42200      911      CONTINUE
42300      DO 912 MONT=0,4
42400      MOMENT=MONT
42500      CALL SET3(NTEMP3,C22,C33,C78,C44,C102,C63,C403,C404,V,
42600      1TRP,PHI,THETVE,XRAD,FDDDED,XLIVE,GSMAL,MOMENT,T,B)
42700      M=1
42800      CALL MATIN(T,NTEMP3,B,M,DETER)
42900      DO 912 KI=1,NTEMP3
43000      XDEAD(MOMENT,KI-1)=B(KI,1)
43100      912      CONTINUE
43200      C
43300      C
43400      IF(IPRIN.EQ.0) GO TO 4506
43500      WRITE(NX,4251)
43600      4251      FORMAT(/5X,'DEAD POLYMER CHAIN LENGTH DISTRIBUTION-----'/)
43700      DO 963 MONT=0,4
43800      WRITE(NX,2499) MONT
43900      WRITE(NX,2500) (KI,XDEAD(MONT,KI),KI=0,NTEMP2)
44000      963      CONTINUE
44100      4506      DO 872 MOMENT=0,4
44200      SUM1=0.0
44300      DO 871 KI=1,NTEMP2
44400      SUM1=SUM1+XLIVE(MOMENT,KI)
44500      871      CONTINUE
44600      XLIVET(MOMENT)=SUM1
44700      872      CONTINUE
44800      DO 874 MOMENT=0,4
44900      SUM2=0.0

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45000      DO 873 KI=0,NTEMP2
45100      SUM2=SUM2+XDEAD(MOMENT,KI)
45200 873   CONTINUE
45300      XDEADT(MOMENT)=SUM2
45400      TYPE *,SUM2
45500 874   CONTINUE
45600      AMNWT=AMW*XDEADT(1)/XDEADT(0)
45700
45800      AMWWT= AMW*XDEADT(2)/XDEADT(1)
45900      PDI=AMWWT/AMNWT
46000      VAVE=X(8)*DUMMY4*VM/X(7)
46100  C
46200  C*****
46300  C
46400      DAVE2=((VAVE*6.0/3.142)**ONETH)*(1.0D+08)
46500      DAVE1=(CONFR(ONETH,XTEMP,XZER)*(DUMMY4**ONETH)*((6.0*VM/3.142)
46600      1**ONETH)/X(7))*1.00D+08
46700      PATNUM=X(7)*AMZERO*ANA/DUMMY3
46800  C
46900  C*****
47000  C
47100  C      INSERT THE FOLLOWING LINES IN PLACE OF THE SET OF LINES
47200  C      ENCLOSED ABOVE FOR THE SECOND APPROACH.
47300  C      VAVE=X(7)*DUMMY4*VM/X(6)
47400  C      DAVE2=((VAVE*6.0/3.142)**ONETH)*(1.0D+08)
47500  C      DAVE1=(CONFR(ONETH,XTEMP,XZER)*(DUMMY4**ONETH)*((6.0*VM/3.142)
47600  C      1**ONETH)/X(6))*1.00D+08
47700  C      PATNUM=X(6)*AMZERO*ANA/DUMMY3
47800  C
47900  C*****
48000  C
48100      CONVER=(XFD(3)-X(3))/XFD(3)
48200      RATE=(XFD(3)-X(3))*AIOW*3.6E+06/(THETA*DUMMY2)
48300  C
48400  C
48500  C
48600  C
48700  C      OUTPUT SECTION BEGINS_____
48800      WRITE(NX,3186)
48900 3186  FORMAT(1H1)
49000      WRITE(NX,3185)
49100 3185  FORMAT(1X,'*****')
49200      WRITE(NX,2147)
49300      WRITE(NX,3185)
49400 2147  FORMAT(/10X,'OUTPUT FROM THE PROGRAM_____/')
49500      WRITE(NX,3185)
49600      WRITE(NX,2148) PATNUM,VAVE,DAVE2,DAVE1,CONVER,RATE
49700 2148  FORMAT(/5X,'(1) NUMBER OF PARTICLES (/CC EMULSION)_____',E15.8/
49800      15X,'(2) AVERAGE VOLUME OF PARTICLES (CC)_____',E15.8/
49900      15X,'(3) AVERAGE DIA ;VOLUME BASIS ("A)_____',E15.8/

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50000      15X,'(4)  AVERAGE DIA ;NUMBER BASIS ("A)_____,E15.8//
50100      15X,'(5)  CONVERSION OBTAINED_____,E15.8/
50200      15X,'(6)  RATE (MOLES/(LIT.HOUR))_____,E15.8//
50300      WRITE(NX,2149) AMNWT,AMWWT,PDI
50400  2149  FORMAT(5X,'(7)  NUMBER AVERAGE MOLECULAR WEIGHT_____,E15.8/
50500      15X,'(8)  WEIGHT AVERAGE MOLECULAR WEIGHT_____,E15.8/
50600      15X,'(9)  POLYDISPERSITY INDEX_____,E15.8//)
50700      WRITE(NX,3185)
50800      IF(IPT.EQ.0) GO TO 4510
50900      WRITE(NX,3186)
51000      WRITE(NX,3187)
51100  3187  FORMAT( /5X,'DETAILS OF THE DISTRIBUTIONS FOLLOW_____'//)
51200      WRITE(NX,3185)
51300  C
51400  C
51500      IWHICH=0
51600      XL1=1.0
51700      PART1=5000.0
51800      CALL DISTR(XL1,PART1,XLIVET,IWHICH,XCORD,YCORD,ZCORD,XCUM)
51900      WRITE(NX,2150)
52000  2150  FORMAT(/5X,'LIVE POLYMER CHAIN LENGTH DISTRIBUTION'//)
52100      WRITE(NX,3185)
52200      WRITE(NX,2151)
52300  2151  FORMAT(/5X,'NO',5X,'MOL. WT.',5X,'NUMB.BASIS',5X,'WT.BASIS
52400      1(NORMAL)'//)
52500      SUM1=0.0
52600      PART1=XCORD(2)-XCORD(1)
52700      DO 870 KI=0,50
52800      SUM1=SUM1+ZCORD(KI)*PART1
52900  870   CONTINUE
53000      DO 863 KI=0,50
53100      ZCORD(KI)=ZCORD(KI)/SUM1
53200  863   CONTINUE
53300      DO 879 KI=0,50
53400      WRITE(NX,2152) KI,XCORD(KI),YCORD(KI),ZCORD(KI)
53500  2152  FORMAT(5X,I2,3X,E11.4,3X,E11.4,3X,E11.4)
53600  879   CONTINUE
53700  C
53800  C
53900  C
54000      CALL DISTR(XL1,PART1,XDEADT,IWHICH,XCORD,YCORD,ZCORD,XCUM)
54100      SUM1=0.0
54200      PART1=XCORD(2)-XCORD(1)
54300      DO 4851 KI=0,50
54400      SUM1=SUM1+ZCORD(KI)*PART1
54500  4851  CONTINUE
54600      DO 4853 KI=0,50
54700      ZCORD(KI)=ZCORD(KI)/SUM1
54800  4853  CONTINUE
54900      WRITE(NX,3186)

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55000      WRITE(NX,3188)
55100      WRITE(NX,3185)
55200  3188  FORMAT( //)
55300      WRITE(NX,2153)
55400  2153  FORMAT(/5X,'DEAD POLYMER CHAIN LENGTH DISTRIBUTION'/)
55500      WRITE(NX,3185)
55600      WRITE(NX,2154)
55700  2154  FORMAT(/5X,'NO',5X,'MOL.WT.',6X,'NUMB.BASIS',5X,'WT.BASIS
55800          1(NORMAL)'/)
55900      DO 891 KI=0,50
56000      WRITE(NX,2152) KI,XCORD(KI),YCORD(KI),ZCORD(KI)
56100  891   CONTINUE
56200  C
56300  C
56400      XL1=1.0
56500      PART1=100.0
56600      IWHICH=1
56700      DO 892 KI=0,4
56800      XVOL(KI)=X(KI+7)*(DUMMY4**KI)/DUMMY3
56900  892   CONTINUE
57000      CALL DISTR(XL1,PART1,XVOL,IWHICH,XCORD,YCORD,ZCORD,XCUM)
57100      DO 2176 KI=0,50
57200      XCORD(KI)=((XCORD(KI)*VM*6.0/3.142)**ONETH)*(1.0D+08)
57300  2176  CONTINUE
57400      WRITE(NX,3186)
57500      WRITE(NX,3188)
57600      WRITE(NX,3185)
57700      WRITE(NX,2155)
57800  2155  FORMAT(/5X,'PARTICLE SIZE DISTRIBUTION [CUMULATIVE]'/)
57900      WRITE(NX,3185)
58000      WRITE(NX,2156)
58100  2156  FORMAT(/5X,'NO',7X,'DIA("A")',9X,'FRACTION' /)
58200      DO 890 KI=0,50
58300      WRITE(NX,2157) KI,XCORD(KI),XCUM(KI)
58400  890   CONTINUE
58500  2157  FORMAT(5X,I2,5X,E11.4,6X,E11.4)
58600  C
58700  4510  WRITE(NX,3186)
58800      STOP
58900      END
59000  C
59100  C
59200      SUBROUTINE NONLIN(N,X,NUMSIG,MAXIT,IPRIN,EPS,DASHI,NX,DELNON,
59300          1PREC,ONETH,TWOTH,BRACE1,BRACE2,BRACE3,C50,C51,C52,C53,C54,C55,
59400          1C556,C56,C7,CT1,C65,DUMMY,THETR,SAI,XFD,X66)
59500  C*****
59600  C
59700  C      THIS SUBROUTINE SOLVES A SET OF NONLINEAR ALGEBRAIC
59800  C      EQUATIONS.IT IS USED TO SOLVE THE EQUATIONS OF FIRST SUBGROUP
59900  C      OF GROUP 1.OTHER SUBROUTINES DIRECTLY CONNECTED WITH THIS

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60000 C SUBROUTINE ARE BACK,AMAX1,AMIN1,FUNX.
60100 C SEE REF 31 FOR DETAILS.
60200 C
60300 C DISCRIPTION OF PRINCIPAL ARGUMENTS____
60400 C N_____NUMBER OF EQUATIONS.
60500 C X(I)_____THE VECTOR OF VARIABLES.
60600 C NUMSIG_____CONVERGENCE CRITERION.
60700 C EPS_____CONVERGENCE CRITERION.
60800 C MAXIT_____MAXIMUM NUMBER OF ITERATIONS ALLOWED BY
60900 C THIS PROCEDURE.
61000 C PREC_____MINIMUM ALLOWABLE STEP SIZE FOR CALCULATING
61100 C PARTIAL DERIVATIVES.
61200 C DELNON_____MINIMUM ALLOWABLE ABSOLUTE VALUE OF PARTIAL
61300 C DERIVATIVES.
61400 C
61500 C*****
61600 DOUBLE PRECISION X(12),PART(12),TEMP(12),COE(12,13),XFD(11)
61700 DOUBLE PRECISION FMAX,FPLUS,F,ABSFF,HOLD,FACTOR,ETA,H
61800 DOUBLE PRECISION ABB,DERMAX,TEST,AA,CC,X66,DASHI,ONETH,TWOTH
61900 DOUBLE PRECISION BRACE1,BRACE2,BRACE3,THETR,DUMMY,SAI
62000 DOUBLE PRECISION C50,C51,C52,C53,C54,C55,C556,C56,C7,CT1,C65
62100 DIMENSION ISUB(12),LOKP(12,12)
62200 RELCON=10.0E+00**(-NUMSIG)
62300 JTEST=1
62400 DO 700 M=1,MAXIT
62500 DO 1111 KI=1,N
62600 IK=KI
62700 IF(X(IK).LE.0.0) X(IK)=0.0
62800 1111 CONTINUE
62900 LQUIT=0
63000 FMAX=0.0
63100 M1=M-1
63200 IF(1PRIN.NE.1) GO TO 9
63300 WRITE(NX,49)M1,(X(I),I=1,N)
63400 49 FORMAT(5X,'M1= ',I5/5(5X,E11.4))
63500 NW=1
63600 9 DO 10 J=1,N
63700 10 LOKP(1,J)=J
63800 DO 555 K=1,N
63900 IF(K-1) 134,134,131
64000 131 KMIN=K-1
64100 KTEMP=K
64200 CALL BACK (KMIN,N,X,ISUB,COE,LOKP,KTEMP)
64300
64400 134 JJJ=K
64500 CALL FUNX(X,F,JJJ,DASHI,
64600 1ONETH,TWOTH,BRACE1,BRACE2,BRACE3,C50,C51,C52,C53,C54,C55,C556,
64700 1C56,C7,CT1,C65,DUMMY,THETR,SAI,XFD,X66)
64800 ABSFF=ABS(F)
64900 FMAX=AMAX1(FMAX,ABSFF)

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65000 C
65100 C
65200 IF(ABS(F).GE.EPS) GO TO 1345
65300 LQUIT=LQUIT+1
65400 IF(LQUIT.NE.N) GO TO 1345
65500 GO TO 725
65600 1345 FACTOR=0.001E+00
65700 135 ITALLY=0
65800 DO 200 I=K,N
65900 ITEMP=LOKP(K,I)
66000 HOLD=X(ITEMP)
66100 ETA=FACTOR*ABS(HOLD)
66200 H=AMIN1(FMAX,ETA)
66300 CCC H=ETA
66400 IF(H.LT.PREC) H=PREC
66500 X(ITEMP)=HOLD+H
66600 IF(K-1) 161,161,151
66700 KTEMP=K
66800 151 CALL BACK(KMIN,N,X,ISUB,COE,LOKP,KTEMP)
66900 161 JJJ=K
67000 CALL FUNX(X,FPLUS,JJJ,DASHI,
67100 1ONETH,TWOTH,BRACE1,BRACE2,BRACE3,C50,C51,C52,C53,C54,C55,C556,
67200 1C56,C7,CT1,C65,DUMMY,THETR,SAI,XFD,X66)
67300 PART(ITEMP)=(FPLUS-F)/H
67400 X(ITEMP)=HOLD
67500 IF(ABS(PART(ITEMP)).LT.DELTA) GO TO 190
67600 C
67700 ABB=ABS(F/PART(ITEMP))
67800 IF(ABB.LE.1.E+15) GO TO 200
67900
68000 190 ITALLY=ITALLY+1
68100 200 CONTINUE
68200 C
68300 NK=N-K
68400 IF(ITALLY.LE.NK) GO TO 202
68500 C
68600 FACTOR=FACTOR*10.0E+00
68700 IF(FACTOR.GT.11.) GO TO 775
68800 GO TO 135
68900 202 IF(K.LT.N) GO TO 203
69000 IF(ABS(PART(ITEMP)).LT.DELTA) GO TO 775
69100 NPL1=N+1
69200 COE(K,NPL1)=0.0
69300 KMAX=ITEMP
69400 GO TO 500
69500 203 KMAX=LOKP(K,K)
69600 DERMAL=ABS(PART(KMAX))
69700 KPLUS=K+1
69800 DO 210 I=KPLUS,N
69900 JSUB=LOKP(K,I)

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70000      TEST=ABS(PART(JSUB))
70100      IF(TEST.LT.DERMAX) GO TO 209
70200      DERMAX=TEST
70300      LOKP(KPLUS,I)=KMAX
70400      KMAX=JSUB
70500      GO TO 210
70600 209   LOKP(KPLUS,I)=JSUB
70700 210   CONTINUE
70800      IF(ABS(PART(KMAX)).EQ.0.0) GO TO 775
70900      ISUB(K)=KMAX
71000      COE (K,N+1)=0.0
71100      DO 220 J=KPLUS,N
71200      JSUB=LOKP(KPLUS,J)
71300      COE(K,JSUB)=-PART(JSUB)/PART(KMAX)
71400  C
71500      NPL1=N+1
71600      COE(K,NPL1)=COE(K,NPL1)+PART(JSUB)*X(JSUB)
71700 220   CONTINUE
71800      NPL1=N+1
71900 500   COE(K,NPL1)=(COE(K,NPL1)-F)/PART(KMAX)+X(KMAX)
72000  C    WRITE(NX,6666) K,(ISUB(IK),IK=1,12),((LOKP(IJ,JI),IJ=1,12),
72100  C      1JI=1,12)
72200 66666  FORMAT(I2/5(2X,E15.7)/5(2X,E15.7))
72300 5555  CONTINUE
72400      NPL1=N+1
72500      X(KMAX)=COE(N,NPL1)
72600      IF(N.EQ.1) GO TO 610
72700      N1=N-1
72800      CALL BACK(N1,N,X,ISUB,COE,LOKP,K)
72900 610   IF(M-1) 650,650,625
73000  C
73100 625   DO 630 I=1,N
73200      AA=ABS(TEMP(I)-X(I))
73300      CC=ABS(X(I))*RELCON
73400
73500      IF(AA.GT.CC) GO TO 649
73600 630   CONTINUE
73700  C
73800  C
73900      JTEST=JTEST+1
74000      IF(JTEST-3) 650,725,725
74100 649   JTEST=1
74200 650   DO 660 I=1,N
74300 660   TEMP(I)=X(I)
74400 700   CONTINUE
74500      WRITE(NX, 1753)
74600 1753  FORMAT(/2X,'WARNING NONLIN;NO CONVERGENCE MAXIMUM NO OF
74700      1ITERATION USED')
74800      IF(IPRIN.NE.1) GO TO 800
74900      WRITE(NX, 1763)

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```

75000 1763  FORMAT(/'FUNCTION VALUE AT LAST APROXIMATION FOLLOW')
75100      IFLAG=1
75200      GO TO 7777
75300 725   IF(IPRIN.NE.1) GO TO 800
75400      IFLAG=0
75500 7777  DO 750 IK=1,N
75600      KII=IK
75700
75800      CALL FUNX(X,PART(KII),KII,DASHI,
75900      1ONETH,TWOTH,BRACE1,BRACE2,BRACE3,C50,C51,C52,C53,C54,C55,C556,
76000      1C56,C7,CT1,C65,DUMMY,THETR,SAI,XFD,X66)
76100 750   CONTINUE
76200      IF(IFLAG.NE.1) GO TO 8777
76300      WRITE(NX, 7788)(PART(K),K=1,N)
76400 7788  FORMAT(3F15.8)
76500      GO TO 800
76600 8777  WRITE(NX, 751)
76700 751   FORMAT(/5X,'CONVERGENCE HAS BEEN ACHIVED .THE FUNCTI')
76800      WRITE(NX, 7515)(PART(K),K=1,N)
76900 7515  FORMAT(5X,'VALUES AT FINAL APPROXIMATION FOLLOW'//
77000      1(2X,F16.8,3(2X,F16.8)))
77100      WRITE(NX, 8888)(X(KI),KI=1,N)
77200 8888  FORMAT(5X,'ANSWER '/3(2X,F16.8))
77300      GO TO 800
77400 775   WRITE(NX,752)
77500 752   FORMAT(/5X,'WARNING NONLIN;MODIFIED JACOBIAN IS SINGULAR TRY ')
77600      WRITE(NX,7525)
77700 7525  FORMAT(5X,'DIFFERENT INITIAL APPROXIMATIO ')
77800      DO 4005 I=1,N
77900      TYPE 4500,(COE(I,J),J=1,N)
78000 4005  CONTINUE
78100      TYPE *,(ISUB(I),I=1,N)
78200 4500  FORMAT(/,3X,6(2X,E11.4))
78300 C     MAXIT=M+1
78400 800   RETURN
78500      END
78600 C
78700 C
78800      SUBROUTINE BACK(KMIN,N,X,ISUB,COE,LOKP,K)
78900 C*****
79000 C
79100 C           THIS SUBROUTINE IS CALLED FROM NONLIN,IT IS USED FOR
79200 C           BACKSOLVING THE SETS OF EQUATIONS OBTAINED IN NONLIN  AT VARIOUS
79300 C           STAGES  OF ITERATION.
79400 C
79500 C*****
79600      DOUBLE PRECISION X(12),COE(12,13)
79700      DIMENSION ISUB(12),LOKP(12,12)
79800      DO 200 KK=1,KMIN
79900      KM=KMIN-KK+2

```

```

80000 C
80100 KM1=KM-1
80200 KMAX=ISUB(KM1)
80300 C
80400 X(KMAX)=0.0
80500 DO 100 J=KM,N
80600 JSUB=LOKP(KM,J)
80700 KM1=KM-1
80800 X(KMAX)=X(KMAX)+COE(KM1,JSUB)*X(JSUB)
80900 100 CONTINUE
81000 KM1=KM-1
81100 NPL1=N+1
81200 X(KMAX)=X(KMAX)+COE(KM1,NPL1)
81300 200 CONTINUE
81400 RETURN
81500 END
81600 C
81700 C
81800 FUNCTION AMAX1(A,C)
81900 C*****
82000 C
82100 C THIS SUBROUTINE IS CALLED FROM NONLIN.IT REPORTS THE
82200 C GREATER OF THE TWO NUMBERS REPRESENTING THE ARGUMENTS(I.E.A AND
82300 C C).
82400 C
82500 C*****
82600 DOUBLE PRECISION A,C
82700 IF(A.GT.C) C=A
82800 AMAX1=C
82900 RETURN
83000 END
83100 C
83200 C
83300 FUNCTION AMIN1(P,Q)
83400 C*****
83500 C
83600 C THIS SUBROUTINE IS CALLED FROM NONLIN.IT REPORTS THE
83700 C SMALLER OF THE TWO NUMBERS REPRESENTING THE ARGUMENTS(I.E.P AND
83800 C Q).
83900 C
84000 C*****
84100 DOUBLE PRECISION P,Q
84200 IF(P.GT.Q) P=Q
84300 AMIN1=P
84400 RETURN
84500 END
84600 C
84700 C
84800 SUBROUTINE FUNX(X,RES,K,DASH1,
84900 1ONETH,TWOTH,BRACE1,BRACE2,BRACE3,C50,C51,C52,C53,C54,C55,C556,

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85000      1C56,C7,CT1,C65, DUMMY,THETR,SAI,XFD)
85100 C*****
85200 C
85300 C      THIS SUBROUTINE IS CALLED FROM NONLIN.IT REPORTS THE
85400 C      VALUE OF K TH EQUATION OF THE SET OF ALGEBRAIC EQUATIONS TO
85500 C      BE SOLVED BY NONLIN.
85600 C
85700 C*****
85800      DOUBLE PRECISION X(12),XFD(12),F(6),FFD(6),FOLD(6)
85900      DOUBLE PRECISION RES,DASHI,ONETH,TWOTH
86000      DOUBLE PRECISION BRACE1,BRACE2,BRACE3,DUMMY,THETR,SAI
86100      DOUBLE PRECISION C50,C51,C52,C53,C54,C55,C556,C56,C7,CT1,C65
86200 C      DOUBLE PRECISION FZERO,FFDZER
86300 C      COMMON ONETH,TWOTH,BRACE1,BRACE2,BRACE3,C50,C51,C52,C53,C54,C55,
86400 C      1C556,C56,CT1,C7,THETR,DUMMY
86500      NMAX=4
86600 C
86700 C
86800 C
86900      W1=8.08E-04
87000      W2=47.69
87100      W3=3.807E-02
87200      W4=2.937E+06
87300      W5=7.0998E-01
87400      W6=0.63
87500      W7=0.1981
87600      FOLD(1)=8.5111E+02
87700      FOLD(2)=7.3E+06
87800      FOLD(3)=9.432E+10
87900      FOLD(4)=1.622E+15
88000 C
88100      DUMMY1=1.0E+06
88200      DUMMY2=1.0E-02
88300      DUMMY3=1.0E-03
88400      DUMMY4=4.2985E+03
88500 C
88600 C      X ARE THE OUTPUT VARIABLES FROM THE REACTOR
88700 C      XFD ARE THE FEED CHARECTARISTIC OF AFORESAID VARIABLES
88800      N=K-8
88900      FZERO=X(7)
89000      FFDZER=XFD(7)
89100      DO 210 INTEMP=1,NMAX
89200      NTEM=INTEMP
89300      F(NTEM)=X(NTEM+7)
89400      FFD(NTEM)=XFD(NTEM+7)
89500      210 CONTINUE
89600      IF(K.GT.7) GO TO 205
89700      GO TO (1,2,3,4,5,6,7),K
89800      1 CONH=CONFR(TWOTH,F,FZERO)
89900      RES=X(12)-CONH

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90000      RETURN
90100  2    RES=(XFD(1)-X(1))/THETR
90200      RES=RES/W1
90300      RETURN
90400  3    RES=C56*(XFD(2)-X(2))/THETR-X(2)
90500      RES=RES/W2
90600      RETURN
90700  4    RES=(XFD(3)-X(3))/(THETR*DUMMY2)-C53*X(6)*(FZERO/DUMMY3)*DASHI
90800      RES=RES/W3
90900      RETURN
91000  5    RES=C50*X(1)-X(4)-C51-X(12)*(DUMMY4**TWOTH)/
91100      1DUMMY3-C52*X(5)
91200      RES=RES/W4
91300      RETURN
91400  6    RES=C54*X(3)/DUMMY2-(X(5))-C55*(F(1)*DUMMY4/DUMMY3)*
91500      1X(6)-C556*X(4)
91600      RES=RES/W5
91700      RETURN
91800  7    RES=X(6)-0.615
91900      RES=RES/W6
92000      RETURN
92100  205  IF (K.GT.8) GO TO 206
92200      RES=C7*X(2)*X(4)/(X(4)+(X(12)*(DUMMY4**TWOTH)/
92300      1DUMMY3))+(FFDZER-FZERO)/(THETR*DUMMY3)
92400  CCC  RES=RES/W7
92500      RETURN
92600  206  IF(K.GT.9)GO TO 207
92700      RES=C7*X(2)*X(4)/(X(4)+(X(12)*(DUMMY4**TWOTH)/
92800      1DUMMY3))+CT1*DASHI*N*X(6)*FZERO/(DUMMY3*(1-X(6)))+
92900      1(FFD(N)-F(N))*(DUMMY4**N)/(THETR*DUMMY3)
93000  CCC  RES=RES/FOLD(N)
93100      RETURN
93200  207  RES=C7*X(2)*X(4)/(X(4)+(X(12)*(DUMMY4**TWOTH)/
93300      1DUMMY3))+CT1*DASHI*N*X(6)*F(N-1)*(DUMMY4**(N-1))/(DUMMY3*
93400      1(1-X(6)))+(FFD(N)-F(N))*(DUMMY4**N)/(THETR*DUMMY3)
93500  CCC  RES=RES/FOLD(N)
93600      RETURN
93700      END
93800  C
93900  C
94000      FUNCTION CONFR(FF,YY,YY0)
94100  C*****
94200  C
94300  C      THE INPUT FOR THIS SUBROUTINE ARE YY AND YY0.YY(I)
94400  C      REPRESENT THE MOMENTS OF THE DISTRIBUTION(I=1,2,---) .YY0
94500  C      REPRESENTS ZERO TH MOMENT OF THE DISTRIBUTION .THE FUNCTION
94600  C      RETURNS THE FF TH MOMENT OF THE DISTRIBUTION.FOR DETAILS OF
94700  C      THE PROCEDURE EMPLOYED SEE CHAPTER 4.
94800  C
94900  C*****

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95000      DOUBLE PRECISION Y(6),YY(6),CCN(6),SSN(6),CALPHA(6),SALPHA(6)
95100      DOUBLE PRECISION YY0,AA,A,ALPHA,ALPHA1,ALKKA,ALPHN1,ALPHF1
95200      DOUBLE PRECISION ALPHFJ,ALPHJ1,FF,RATIO1
95300      NMAX=4
95400      AA=YY(2)*YY0/(YY(1)*YY(1))
95500      Y(1)=1.0/(AA-1.0)
95600      Y(2)=AA*Y(1)*Y(1)
95700      ALPHA=Y(1)-1.
95800      ALPHA1=Y(1)
95900      IF(ALPHA1.LT.0) GO TO 450
96000      C
96100      C
96200      DO 118 IK=3,NMAX
96300      KI=IK
96400      Y(KI) =((ALPHA1*YY0/YY(1))*KI)*YY(KI)/YY0
96500      118  CONTINUE
96600      ALPHF1=ALPHA+FF+1.0
96700      C
96800      DO 100 NMMM=3,NMAX
96900      N=NMMM
97000      CALPHA(N)=1
97100      DO 100 KKJJ=1,N
97200      KK=KKJJ
97300      NKK=N-KK
97400      ALKKA=ALPHA+KK+1.
97500      CCN(KK)=((-1)**KK)*FACTO(N)*GAMM(ALPHA1)/(FACTO(KK)*
97600      1FACTO(NKK)*GAMM(ALKKA))
97700      CALPHA(N)=CALPHA(N)+CCN(KK)* Y(KK)
97800      100  CONTINUE
97900      DO 201 NMMM=3,NMAX
98000      N=NMMM
98100      SALPHA(N)=0.0
98200      ALPHN1=N+ALPHA+1.
98300      SALPHA(N)=GAMM(ALPHN1)*GAMM(ALPHF1)/(FACTO(N)*GAMM(ALPHA1))
98400      DO 201 JJK=1,N
98500      JJ=JJK
98600      ALPHFJ=ALPHA+FF+JJ+1.0
98700      ALPHJ1=ALPHA+JJ+1.0
98800      ALPHN1=ALPHA+N+1.0
98900      NJJ=N-JJ
00100      SSN(JJ)=((-1.0)**JJ)*GAMM(ALPHN1)*GAMM(ALPHFJ)/(FACTO(NJJ)*
00200      1FACTO(JJ)*GAMM(ALPHJ1))
00300      SALPHA(N)=SALPHA(N)+SSN(JJ)
00400
00500
00600      201  CONTINUE
00700      A=GAMM(ALPHF1)
00800      DO 400 NMMM=3,NMAX
00900      N=NMMM

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01000      A=A+CALPHA(N)*SALPHA(N)
01100  400  CONTINUE
01200      A=A/GAMM(ALPHA1)
01300      CONFR=((YY(1)/(YY0*ALPHA1))**FF)*A*YY0
01400      RETURN
01500  450  RATIO1=YY(1)/YY0
01600      CONFR=YY0*(RATIO1**FF)
01700      RETURN
01800      END
01900  C
02000  C
02100      FUNCTION FACTO(N)
02200  C*****
02300  C
02400  C          THIS FUNCTION REPORTS THE VALUE OF FACTORIAL OF AN
02500  C          INTEGER VARIABLE N.
02600  C
02700  C*****
02800      FACTO=1
02900      IF(N.EQ.0) GO TO 2
03000      DO 1 JK=1,N
03100      FACTO=FACTO*JK
03200  1      CONTINUE
03300  2      RETURN
03400      END
03500  C
03600  C
03700      FUNCTION GAMM(XX)
03800  C*****
03900  C
04000  C          THIS FUNCTION REPORTS THE VALUE OF GAMMA FUNCTION OF
04100  C          THE VARIABLE XX.
04200  C
04300  C*****
04400      DOUBLE PRECISION XX,X,Y
04500      IF(XX-57.0) 6,6,4
04600
04700  4      IER=2
04800      GAMM=1.E+30
04900      RETURN
05000  6      X=XX
05100      ERR=1.0E-06
05200      IER=0
05300      GAMM=1.0
05400      IF(X-2.0)50,50,15
05500  10      IF(X-2.0) 110,110,15
05600  15      X=X-1.
05700      GAMM=GAMM*X
05800      GO TO 10
05900  50      IF(X-1.0) 60,120,110

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06000 C      SEE IF X IS NEAR NEGATIVE INTEGER OR ZERO
06100 60      IF(X=ERR) 62,62,80
06200 62      Y=FLOAT(INT(X))-X
06300      IF (ABS(Y)=ERR) 130,130,64
06400 64      IF(1.0-Y=ERR) 130,130,70
06500 C      X NOT NEAR A NEGATIVE INTEGER OR ZERO
06600 70      IF(X=1.0) 80,80,110
06700 80      GAMM=GAMM/X
06800      X=X+1.0
06900      GO TO 70
07000 110     Y=X-1.0
07100      GY=1.0+Y*(-0.5771017+Y*(+0.9858540+Y*(-0.8764218+Y*
07200      1(+0.8328212+Y*(-0.5684729+Y*(0.2548205+Y*(-.05149930))))))
07300      GAMM=GAMM*GY
07400 120     RETURN
07500 130     IER=1
07600      RETURN
07700      END
07800 C
07900 C
08000      SUBROUTINE SET1 (N,FFD,ANUMBR,THETR,C22,C33,C78,C79,C44,T,B)
08100 C*****
08200 C
08300 C      THIS SUBROUTINE CALCULATES THE ELEMENTS OF COEFFICIENT
08400 C      MATRIX OF THE SET LINEAR EQUATIONS IN SECOND SUBGROUP OF THE
08500 C      GROUP 1. COEFFICIENT MATRIX IS REPRESENTED BY T.B(I,1) IS THE
08600 C      VECTOR OF CONSTANTS .
08700 C
08800 C*****
08900      DOUBLE PRECISION T(25,25),B(25, 1),FFD(0:20)
09000      DOUBLE PRECISION ANUMBR,THETR,C22,C33,C78,C79,C44,THETVE,BRACES
09100      N1=N-1
09200      THETVE=1./THETR
09300      DO 104 IROW=1,N
09400      DO 104 ICOL=1,N
09500      T(IROW,ICOL)=0.0
09600 104      CONTINUE
09700 C
09800      DO 105 IROW=1,N
09900      IF(IROW.EQ.N) GO TO 433
10000      ICOL=IROW
10100      ICOLM2=ICOL-2
10200      ICOLM1=ICOL-1
10300      ICOLP1=ICOL+1
10400      ICOLP2=ICOL+2
10500 C
10600      NEST=ICOLM2
10700      IF(NEST.LE.-1) GO TO 102
10800      IF(NEST.LE.0) GO TO 103
10900      T(IROW,NEST)=C22

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11000 103      NEST=ICOLM1
11100          T(IROW,NEST)=C78
11200 102      NEST=ICOL
11300          T(IROW,NEST)=-C78-C33*(IROW-1)-C44*(IROW-1)*(IROW-2)-THETVE
11400          1-C22
11500          NEST=ICOLP1
11600          IF(NEST.GT.N) GO TO 105
11700          T(IROW,NEST)=C33*ICOL
11800          NEST=ICOLP2
11900          IF(NEST.GT.N) GO TO 105
12000          T(IROW,NEST)=ICOL*ICOLP1*C44
12100          GO TO 105
12200 433      DO 434 KI=1,N
12300          IK=KI
12400          T(IROW,IK)=1.0
12500 434      CONTINUE
12600 105      CONTINUE
12700 C
12800          BRACES=THETR*ANUMBR
12900          DO 439 IROW=1,N
13000          IF(IROW.EQ.N) GO TO 438
13100          IF(IROW.EQ.2) GO TO 437
13200          B(IROW,1)=-FFD(IROW-1)/BRACES
13300          GO TO 439
13400 437      B(IROW,1)=-FFD(IROW-1)/BRACES-C79/ANUMBR
13500          GO TO 439
13600 438      B(IROW,1)=+1.0
13700 439      CONTINUE
13800          RETURN
13900          END
14000 C
14100 C
14200          SUBROUTINE LIN(NUMEQ,FFD,ANUMBR,THETR,C22,C33,C78,C79,C44,NX,
14300          1DASHI2,NTEMP3,B,IPRIN)
14400 C*****
14500 C
14600 C          THIS SUBROUTINE SOLVES THE EQUATIONS OF SECOND SUBGROUP
14700 C          OF THE FIRST GROUP OF EQUATIONS. INITIALLY 'NUMEQ' EQUATIONS ARE
14800 C          SOLVED BY CALLING SET1 TO OBTAIN COEFFICIENT MATRIX AND MATIN
14900 C          TO THIS EQUATIONS. THE ANSWER IS OBTAINED AS B(I,1) I=1,---NUMEQ.
15000 C          THE LAST TWO VALUES OF THE SET ARE TESTED TO BE LESS THAN
15100 C          1.0E-07 AND THE PROCESS IS REPEATED BY INCREASING THE NUMBER OF
15200 C          EQUATIONS BY 2. FOR DETAILS SEE CHAPTER 4.
15300 C
15400 C*****
15500          DOUBLE PRECISION T(25,25),B(25,1),FFD(0:20)
15600          DOUBLE PRECISION ANUMBR,THETR,C22,C33,C78,C79,C44,DASHI2
15700          DOUBLE PRECISION SUM3,SUM4
15800          NTEMP=NUMEQ
15900 334      CALL SET1(NTEMP,FFD,ANUMBR,THETR,C22,C33,C78,C79,C44,T,B)

```

```

16000      M=1
16100      CALL MATIN(T,NTEMP,B,M,DETER)
16200      NTEMP1=NTEMP-1
16300      NTEMP2=NTEMP-2
16400      IF(B(NTEMP1,1).GE.1.0E-07) GO TO 338
16500      IF(B(NTEMP2,1).GE.1.0E-07) GO TO 338
16600      NTEMP3=NTEMP-3
16700      IF(IPRIN.EQ.0) GO TO 339
16800      WRITE(NX,1119) (NTEMP3,(B(IK,1),IK=1,NTEMP3))
16900 1119   FORMAT(/5X,'ANSWER FROM LIN'//5X,'NTEMP3= ',I2//,2(5X,E15.8 ))
17000      GO TO 339
17100 338   NTEMP=NTEMP+2
17200      GO TO 334
17300 339   SUM3=0.0
17400      SUM4=0.0
17500      DO 335 IK=1,NTEMP3
17600      KI=IK
17700      SUM3=SUM3+B(KI,1)*(KI-1)
17800      SUM4=SUM4+B(KI,1)
17900 335   CONTINUE
18000      DASHI2=SUM3/SUM4
18100      RETURN
18200      END
18300  C
18400  C
18500      SUBROUTINE MATIN(A,N,B,M,DETER)
18600  C*****
18700  C
18800  C          THIS SUBROUTINE INVERTS THE COEFFICIENT MATRIX OF A SET
18900  C          OF LINEAR EQUATIONS.A REPRESENTS THE COEFFICIENT MATRIX .N IS
19000  C          THE NUMBER OF EQUATIONS.PUT M=1 IF ONLY SOLUTION OF EQUATIONS IS
19100  C          NEEDED.THE SOLUTION IS REPRESENTED BY B(I,1),I=1,---,N .
19200  C
19300  C*****
19400      DOUBLE PRECISION B(25,1)
19500      DOUBLE PRECISION A(25,25)
19600      DOUBLE PRECISION AMAX,SWAP,PIVOT,DETRM,T
19700      DIMENSION IPIVT(40),INDEX(40,2)
19800      EQUIVALENCE(IROW,JROW),(ICOLM,JCOLM),(AMAX,T,SWAP)
19900      10   DETR=1.0
20000      15   DO 20 J=1,N
00100      20   IPIVT(J)=0
00200      30   DO 550 I=1,N
00300      40   AMAX=0.0
00400      45   DO 105 J=1,N
00500      50   IF(IPIVT(J)-1) 60,105,60
00600      60   DO 100 K=1,N
00700      70   IF(IPIVT(K)-1) 80,100,740
00800      80   IF(AMAX-ABS(A(J,K))) 85,100,100

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00900	85	IROW=J
01000	90	ICOLM=K
01100	100	CONTINUE
01200	105	CONTINUE
01300	110	IPIVT(ICOLM)=IPIVT(ICOLM)+1
01400	130	IF(IROW-ICOLM) 140,260,140
01500	140	DETER=-DETER
01600	150	DO 200 L=1,N
01700	160	SWAP=A(IROW,L)
01800	170	A(IROW,L)=A(ICOLM,L)
01900	200	A(ICOLM,L)=SWAP
02000	205	IF(M) 260,260,210
02100	210	DO 250 L=1,M
02200	220	SWAP=B(IROW,L)
02300	230	B(IROW,L)=B(ICOLM,L)
02400	250	B(ICOLM,L)=SWAP
02500	260	INDEX(1,1)=IROW
02600	270	INDEX(1,2)=ICOLM
02700	310	PIVOT=A(ICOLM,ICOLM)
02800	320	DETRM=DETER*PIVOT
02900	330	A(ICOLM,ICOLM)=1.0
03000	340	DO 350 L=1,N
03100	350	A(ICOLM,L)=A(ICOLM,L)/PIVOT
03200	355	IF(M) 380,380,360
03300	360	DO 370 L=1,M
03400	370	B(ICOLM,L)=B(ICOLM,L)/PIVOT
03500	380	DO 550 L1=1,N
03600	390	IF(L1-ICOLM) 400,550,400
03700	400	T=A(L1,ICOLM)
03800	420	A(L1,ICOLM)=0.0
03900	430	DO 450 L=1,N
04000	450	A(L1,L)=A(L1,L)-A(ICOLM,L)*T
04100	455	IF(M) 550,550,460
04200	460	DO 500 L=1,M
04300	500	B(L1,L)=B(L1,L)-B(ICOLM,L)*T
04400	550	CONTINUE
04500	600	DO 710 I=1,N
04600	610	L=N+1-I
04700	620	IF(INDEX(L,1)-INDEX(L,2)) 630,710,630
04800	630	JROW=INDEX(L,1)
04900	640	JCOLM=INDEX(L,2)
05000	650	DO 705 K=1,N
05100	660	SWAP=A(K,JROW)
05200	670	A(K,JROW)=A(K,JCOLM)
05300	700	A(K,JCOLM)=SWAP
05400	705	CONTINUE
05500	710	CONTINUE
05600		DO 11 K=1,N
05700		IF(IPIVT(K)-1) 12,11,12
05800	11	CONTINUE

```

10900      SUM1=0.0
11000      DO 467 KI1=0,MOMENT
11100      KI=KI1
11200      IMONT=MOMENT-KI
11300      SUM1=SUM1+FACTO(MOMENT)*GSMAL(KI,IRAD)*GSMAL(IMONT,IRAD)/
11400      1(FACTO(IMONT)*FACTO(KI))
11500  467    CONTINUE
11600      TERM(IRAD)=C403*SUM1*XRAD(IRAD)/V
11700  468    CONTINUE
11800      DO 9 IROW=1,NTEMP3
11900      IF(IROW.GT.1) GO TO 7
12000      B(IROW,1)=-TERM(2)=C404*XLIVE(MOMENT,2)/V

00100      1=FDDDED(MOMENT,0)*THETVE
00200      GO TO 9
00300  7      B(IROW,1)=-TERM(IROW+1)=C404*IROW*
00400      1XLIVE(MOMENT,IROW+1)/V-C63*TRP*XLIVE(MOMENT,IROW-1)
00500      1=C102*PHI*XLIVE(MOMENT,IROW-1)-FDDDED(MOMENT,IROW-1)*THETVE
00600  9      CONTINUE
00700      RETURN
00800      END
00900  C
01000  C
01100      SUBROUTINE SET2(NTEMP3,C22,C33,C44,C78,C79,C63,C102,C101,TRP,
01200      1MOMENT,PHI,THETVE,FDLIV,XRAD,XLIVE,T,B)
01300  C*****
01400  C
01500  C          THIS SUBROUTINE CALCULATES THE COEFFICIENT MATRIX OF THE
01600  C          SET OF SET LINEAR EQUATIONS OF GROUP 2.(I.E. LIVE POLYMER CHAIN
01700  C          LENGTH DISTRIBUTION) COEFFICIENT MATRIX REPRESENTED T,VECTOR OF
01800  C          CONSTANTS IS B(I,1) .
01900  C
02000  C*****
02100      DOUBLE PRECISION T(25,25),B(25,1),XRAD(0:20)
02200      DOUBLE PRECISION FDLIV(5,20),XLIVE(0:5,20),TERM(20)
02300      DOUBLE PRECISION C22,C33,C44,C78,C63,C101,C102,PHI,TRP,THETVE
02400      DOUBLE PRECISION SUM1,C79
02500      NTEMP2=NTEMP3-1
02600      DO 4 IROW=1,NTEMP2
02700      DO 4 ICOL=1,NTEMP2
02800      T(IROW,ICOL)=0.0
02900  4      CONTINUE
03000  C
03100      DO 5 IROW=1,NTEMP2
03200      ICOL=IROW
03300      ICOLM2=ICOL-2
03400      ICOLM1=ICOL-1
03500      ICOLP1=ICOL+1
03600      ICOLP2=ICOL+2
03700  C

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03800      NEST=ICOLM2
03900      IF(NEST.LE.-1) GO TO 2
04000      IF(NEST.LE.0) GO TO 3
04100      T(IROW,NEST)=C22
04200  3      NEST=ICOLM1
04300      T(IROW,NEST)=C78
04400  2      NEST=ICOL
04500      T(IROW,NEST)=-C78-C22-C33*ICOL-C63*TRP-C102*PHI-C44*ICOL*ICOLM1
04600      1=THETVE
04700      NEST=ICOLP1
04800      IF(NEST.GT,NTEMP2) GO TO 5
04900      T(IROW,NEST)=C33*ICOL
05000      NEST=ICOLP2
05100      IF(NEST.GT,NTEMP2) GO TO 5
05200      T(IROW,NEST)= C44*ICOLP1*ICOL
05300  5      CONTINUE
05400      DO 468 IRAD1=1,NTEMP2
05500      IRAD=IRAD1
05600      SUM1=0.0
05700      DO 467 KI1=1,MOMENT
05800      KI=KI1
05900      IMONT=MOMENT-KI
06000      SUM1=SUM1+FACTO(MOMENT)*XLIVE(IMONT,IRAD)/(FACTO(KI)*
06100      1FACTO(IMONT ))
06200  467      CONTINUE
06300      TERM(IRAD)=C101*PHI*SUM1
06400  468      CONTINUE
06500  C
06600      DO 9 IROW=1,NTEMP2
06700      IF(IROW.GT.1) GO TO 7
06800      B(IROW,1)=-C78*XRAD(0)-C102*PHI*1.0*XRAD(1)-THETVE*
06900      1FDLIV(MOMENT,1)-TERM(1)-C63*TRP*1.0*XRAD(1)-C79
07000      GO TO 9
07100  7      B(IROW,1)=-C78*XRAD(IROW-1)-2.0*C22*XRAD(IROW-2)-C102*PHI*
07200      1IROW*XRAD(IROW)-THETVE*FDLIV(MOMENT,IROW)-TERM(IROW)
07300      1-C63*TRP*IROW*XRAD(IROW)
07400  9      CONTINUE
07500      RETURN
07600      END
07700  C
07800  C
07900      FUNCTION DIFF2(AMV,V)
08000  C*****
08100  C
08200  C      THE INPUT FOR THIS FUNCTION CONSISTS OF AMV AND V,AMV(I)
08300  C      ARE THE MOMENTS OF A DISTRIBUTION,(I=0,1,2---) THE FUNCTION
08400  C      RETURNS THE VALUE OF DISTRIBUTION FUNCTION EVALUATED AT V,
08500  C      FOR DETAILS SEE CHAPTER 4.
08600  C
08700  C*****

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08800      DOUBLE PRECISION AMV(0:5),GX(6),CALPHA(6),ALPHA(6)
08900      C      AMV ARE FIRST FOUR MOMENTS OF DISTRIBUTION
09000      C      AMVZER IS THE ZEROth ABSOLUTE MOMENT OF DISTRIBUTION
09100      C      X IS THE X CO-ORDINATE OF WHICH
09200      C      THE FUNCTION IS TO BE EVALUATED.
09300      C      RESULT IS THE FUNCTION VALUE CALCULATED AT
09400      C      AFORESAID X CO-ORDINATE @V@.
09500      C
09600      C      INPUT PORTION=====
09700      C      CONVERT MV INTO GX(I,E, MOMENTS OF NORMALIEED
09800      C      DISTRIBUTION.)
09900      C
10000      DOUBLE PRECISION AMVZER,PDI,ALPHA,ALPHA1,ALPIT1,ALPIK1,XMINU,SUM
10100      DOUBLE PRECISION  V
10200      AMVZER=AMV(0)
10300      PDI=AMV(2)*AMVZER/(AMV(1)*AMV(1))
10400      GX(1)=1.0/(PDI-1.0)
10500      GX(2)=PDI*GX(1)*GX(1)
10600      ALPHA=GX(1)-1.0
10700      ALPHA1=GX(1)
10800      NMAX=4
10900      DO 118 KI=3,NMAX
11000      IK=KI
11100      GX(IK)=((ALPHA1*AMVZER/AMV(1))**IK)*AMV(IK)/AMVZER
11200      118  CONTINUE
11300      C      CONVERT ABSOLUTE CO-ORDINATE V TO THE CORRESSPONDING
11400      C      NORMALISED CO-ORDINATE X.
11500      X=(ALPHA1*AMVZER/AMV(1))*V
11600      C      CALCULATE CALPHA.
11700      DO 100 KI=3,NMAX
11800      IK=KI
11900      CALPHA(IK)=1.0
12000      DO 100 ITAM=1,IK
12100      ITEMP=ITAM
12200      IKITAM=IK-ITEMP
12300      ALPIT1=ALPHA1+ITEMP
12400      CALPHA(IK)=CALPHA(IK)+((-1)**ITEMP)*FACTO(IK)*GAMM(ALPHA1)*
12500      1GX(ITEMP)/(FACTO(IKITAM)*FACTO(ITEMP)*GAMM(ALPIT1))
12600      100  CONTINUE
12700      C      CALCULATION OF MODIFIED LAGURE@ POLYNOMIAL
12800      DO 200 KI=3,NMAX
12900      IK=KI
13000      ALPIK1=ALPHA1+IK
13100      ALPHAL(IK)=GAMM(ALPIK1)/(GAMM(ALPHA1)*FACTO(IK))
13200      DO 200 ITAM=1,IK
13300      ITEMP=ITAM
13400      IKITAM=IK-ITEMP
13500      ALPIT1=ALPHA1+ITEMP
13600      ALPHAL(IK)=ALPHAL(IK)+((-1)**ITEMP)*GAMM(ALPIK1)*
13700      1(X**ITEMP)/(FACTO(IKITAM)*GAMM(ALPIT1)*FACTO(ITEMP))

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13800 200 CONTINUE
13900 C CALCULATION OF NORMALISED DISTRIBUTION FUNCTON.
14000 SUM=1.0
14100 DO 300 KI=3,NMAX
14200 IK=KI
14300 SUM=SUM+CALPHA(IK)*ALPHA(IK)
14400 300 CONTINUE
14500 XMINU=-X
14600 GNORM=SUM*EXP(XMINU)*(X**ALPHA)/GAMM(ALPHA1)
14700 C CALCULATION OF VALUE OF THE ABSOLUTE FUNX.
14800 DIFF2=GNORM*ALPHA1*AMVZER*AMVZER/AMV(1)
14900 RETURN
15000 END
15100 C
15200 C
15300 C
15400 C
15500 SUBROUTINE CENTR(N,M,K,IEV1,I,XC,X,K1)
15600 C*****
15700 C
15800 C THIS SUBROUTINE IS REQUIRED FOR OPTIMIZATION,IT
15900 C CALCULATES THE COEFFICIENTS OF THE CENTROID OF THE COMPLEX.
16000 C
16100 C*****
16200 DOUBLE PRECISION X(K,M),XC(N)
16300 DO 20 J=1,N
16400 XC(J)=0.0
16500 DO 10 IL=1,K1
16600 XC(J)=XC(J)+X(IL,J)
16700 10 CONTINUE
16800 RK=K1
16900 XC(J)=(XC(J)-X(IEV1,J))/(RK-1.0)
17000 20 CONTINUE
17100 RETURN
17200 END
17300 C
17400 C
17500 SUBROUTINE CHECK(N,M,K,X,G,H,I,KODE,XC,DELTA,K1)
17600 C*****
17700 C
17800 C THIS SUBROUTINE IS REQUIRED FOR OPTIMIZATION,IT CHECKS
17900 C IF A GIVEN POINT SATISFIES THE GIVEN SET OF CONSTRAINTS AND
18000 C AND CALCULATES A FEASIBLE VERTEX FOR THE COMPLEX.
18100 C
18200 C*****
18300 DOUBLE PRECISION X(K,M),G(M),H(M),XC(N)
18400 10 KT=0
18500 CALL CONST(N,M,K,X,G,H,I)
18600 DO 50 J=1,N
18700 IF(X(I,J)-G(J)) 20,20,30

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18800      20      X(I,J)=G(J)+DELTA
18900      GO TO 50
19000      30      IF(H(J)-X(I,J)) 40,40,50
19100      40      X(I,J)=H(J)-DELTA
19200      50      CONTINUE
19300      NO=5
19400      C      WRITE (NO,*) ((J,X(1,J),XC(J),X(2,J)) ,J=1,N)
19500      IF(KODE) 110,110,60
19600      60      NN=N+1
19700      DO 100 J=NN,M
19800      CALL CONST(N,M,K,X,G,H,I)
19900      IF(X(I,J)-G(J)) 80,70,70
20000      70      IF(H(J)-X(I,J)) 80,100,100
20100      80      IEV1=I
20200      KT=1
20300      CALL CENTR(N,M,K,IEV1,I,XC,X,K1)
20400      DO 90 JJ=1,N
20500      X(I,JJ)=(X(I,JJ)+XC(JJ))/2.0
20600      90      CONTINUE
20700      100     CONTINUE
20800      IF(KT) 110,110,10
20900      110     RETURN
21000      END
21100      C
21200      C
21300      SUBROUTINE CONSX(N,M,K,ITMAX,ALPHA,BETA,GAMMA,DELTA,X,R,F,
21400      1IT,IEV2,NO,G,H,XC,IPRIN,DASH1,ONETH,TWOTH,BRACE1,BRACE2,
21500      1BRACE3,C50,C51,C52,C53,C54,C55,C556,C56,CT1,C7,THETR,DUMMY,XFD,
21600      1C65,SAT)
21700      C*****
21800      C
21900      C      THIS SUBROUTINE IS USED TO FIND MAXIMA/MINIMA OF A
22000      C      CONSTRAINED MULTIVARIABLE FUNCTION BY BOX'S METHOD,IT IS USED
22100      C      TO FIND INITIAL GUESSES TO SOLVE THE EQUATIONS OF FIRST SUBGROUP
22200      C      OF GROUP 1,FOR DETAILS SEE CHAPTER 4 AND REF 32
22300      C
22400      C      DISRIPTION OF PRINCIPAL VARIABLES-----
22500      C      ALPHA-----COMPLEX EXPANSION PARAMETER,
22600      C      (USUALLY 1,3)
22700      C      BETA-----CONVERGENCE CRITERION
22800      C      GAMMA-----CONVERGENCE CRITERION
22900      C      DELTA-----USED TO FIND A FEASIBLE POINT
23000      C      WHEN THE CALCULATED VERTEX OF TH
23100      C      COMPLEX IS FOUND TO BE OUTSIDE
23200      C      ANY CONSTRAINT.
23300      C      N-----NUMBER OF VARIABLES,
23400      C      M-----NUMBER OF CONSTRAINTS,
23500      C      K-----NUMBER OF VERTICES,
23600      C      ITMAX-----MAXIMUM NUMBER OF ITERATIONS
23700      C      ALLOWED FOR OPTIMIZATION,

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23800 C      R-----RANDOM NUMBERS.
23900 C      F-----OBJECTIVE FUNCTION.
24000 C      X(I)-----VECTOR OF VARIABLES
24100 C
24200 C*****
24300      DOUBLE PRECISION ONETH,TWOTH,BRACE1,BRACE3,BRACE2,THETR,DUMMY
24400      DOUBLE PRECISION C50,C51,C52,C53,C54,C55,C556,C56,C7,CT1,C65
24500      DOUBLE PRECISION DASHI,DASHI1,DASHI2,SAI
24600      INTEGER GAMMA
24700      DOUBLE PRECISION X(K,M),F(K),G(M),H(M),XC(N),XFD(N)
24800      DIMENSION R(K,N)
24900      IT=1
25000      IPRIN1=IPRIN
25100      KODE=0
25200      IF(M=N)20,20,10
25300      10 KODE=1
25400      20 CONTINUE
25500      DO 40 II=2,K
25600      DO 30 J=1,N
25700      X(II,J)=0.0
25800      30 CONTINUE
25900      40 CONTINUE
26000      DO 65 II=2,K
26100      DO 50 J=1,N
26200      I=II
26300      CALL CONST(N,M,K,X,G,H,I)
26400      X(II,J)=G(J)+R(II,J)*(H(J)-G(J))
26500      50 CONTINUE
26600      K1=II
26700      CALL CHECK(N,M,K,X,G,H,I,KODE,XC,DELTA,K1)
26800      IF(II=2)51,51,55
26900      51 IF(IPRIN)52,65,52
27000      52 WRITE(NO,018)
27100      18 FORMAT(1H1//2X,"COORDINATES OF INITIAL COMPLEX")
27200      IO=1
27300      WRITE (NO,159) (IO,J,X(IO,J),J=1,N)
27400      159 FORMAT(/3(2X,2HX(,I2,1H,,I2,4H) = ,E13,6))
27500      55 IF(IPRIN) 56,65,56
27600      56 WRITE (NO,159) (II,J,X(II,J),J=1,N)
27700      65 CONTINUE
27800      K1=K
27900      DO 70 IL=1,K
28000      I=IL
28100      CALL FUNC(N,M,K,X,F,I,DASHI,BRACE1,BRACE2,
28200      1BRACE3,C50,C51,C52,C53,C54,C55,C556,C56,CT1,C7,THETR,DUMMY,XFD,
28300      1C65,SAI)
28400      70 CONTINUE
28500      KOUNT=1
28600      IA=0
28700      IF(IPRIN) 72,80,72

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28800 72 WRITE(NO,21)
28900 21 FORMAT(/2X,'VALUES OF THE FUNCTION')
29000 WRITE(NO,22) (J,F(J),J=1,K)
29100 22 FORMAT(/,3(2X,2HF(,I2,4H) = ,E13,6))
29200 80 IEV1=1
29300 DO 100 ICM=2,K
29400 IF(F(IEV1)-F(ICM)) 100,100,90
29500 90 IEV1=ICM
29600 100 CONTINUE
29700 IEV2=1
29800 DO 120 ICM=2,K
29900 IF(F(IEV2)-F(ICM)) 110,110,120
30000 110 IEV2=ICM
30100 120 CONTINUE
30200 IF(F(IEV2)-(F(IEV1)+BETA)) 140,130,130
30300 130 KOUNT=1
30400 GO TO 150
30500 140 KOUNT=KOUNT+1
30600 IF(KOUNT=GAMMA) 150,240,240
30700 150 CALL CENTR(N,M,K,IEV1,I,XC,X,K1)
30800 DO 160 JJ=1,N
30900 X(IEV1,JJ)=(1.0+ALPHA)*XC(JJ)-ALPHA*X(IEV1,JJ)
31000 160 CONTINUE
31100 I=IEV1
31200 CALL CHECK(N,M,K,X,G,H,I,KODE,XC,DELTA,K1)
31300 CALL FUNC(N,M,K,X,F,I,DASHI,BRACE1,BRACE2,
31400 1BRACE3,C50,C51,C52,C53,C54,C55,C556,C56,CT1,C7,THETR,DUMMY,XFD,
31500 1C65,SAI)
31600 170 IEV2=1
31700 DO 190 ICM=2,K
31800 IF(F(IEV2)-F(ICM)) 190,190,180
31900 180 IEV2=ICM
32000 190 CONTINUE
32100 IF(IEV2-IEV1) 220,200,220
32200 200 DO 210 JJ=1,N
32300 X(IEV1,JJ)=(X(IEV1,JJ)+XC(JJ))/2.0
32400 210 CONTINUE
32500 I=IEV1
32600 CALL CHECK(N,M,K,X,G,H,I,KODE,XC,DELTA,K1)
32700 CALL FUNC(N,M,K,X,F,I,DASHI,BRACE1,BRACE2,
32800 1BRACE3,C50,C51,C52,C53,C54,C55,C556,C56,CT1,C7,THETR,DUMMY,XFD,
32900 1C65,SAI)
33000 GO TO 170
33100 220 CONTINUE
33200 IF(IT/50*50.NE.IT) IPRIN=0
33300 MTYPE=1
33400 IF(IT/50*50.NE.IT) MTYPE=0
33500 IF(MTYPE.EQ.0) GO TO 4511
33600 C IF(IT.GT.300) IPRIN=1
33700 TYPE *,IT

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33800 4511 IF(IPRIN) 230,228,230
33900 230 WRITE(NO,23)IT
34000 23 FORMAT(/,2X,"ITERATION NUMBER",I5)
34100 WRITE(NO,24)
34200 24 FORMAT(/,2X,"CO-ORDINATES OF CORRECTED POINT")
34300 WRITE(NO,159) (IEV1,JC,X(IEV1,JC),JC=1,N)
34400 WRITE(NO,21)
34500 WRITE(NO,22)(I,F(I),I=1,K)
34600 WRITE(NO,25)
34700 25 FORMAT(/,2X,"CO-ORDINATES OF THE CENTROID")
34800 WRITE(NO,26)(JC,XC(JC),JC=1,N)
34900 26 FORMAT(/,3(2X,2HX(,I2,6H,C) = ,E13,6))
35000 228 IT=IT+1
35100 IPRIN=IPRIN1
35200 IF(IT=ITMAX) 80,80,240
35300 240 RETURN
35400 END
35500 C
35600 C
35700 SUBROUTINE CONST(N,M,K,X,G,H,I)
35800 DOUBLE PRECISION X(14,16),G(16),H(16)
35900 C*****
36000 C
36100 C THIS SUBROUTINE DEFINES VARIOUS EXPLICIT AND IMPLICIT
36200 C CONSTRAINTS.
36300 C G(I)-----LOWER CONSTRAINTS.
36400 C H(I)-----UPPER CONSTRAINTS.
36500 C I=1,---N-----EXPLICIT CONSTRAINTS.
36600 C I=N+1,---M-----IMPLICIT CONSTRAINTS.
36700 C
36800 C*****
36900 G(1)=0.0
37000 H(1)=4.9
37100 G(2)=0.0
37200 H(2)=1.0
37300 G(3)=0.0
37400 H(3)=1.92
37500
37600 G(4)=1.0E-06
37700 H(4)=1.0E+06
37800 G(5)=1.0E-06
37900 H(5)=1.0E+06
38000 G(6)=.01
38100 H(6)=1.0
38200 G(7)=1.0E-06
38300 H(7)=1.0E+03
38400 G(8)=1.0E-06
38500 H(8)=1.0E+04
38600 G(9)=1.0E-06
38700 H(9)=1.0E+05

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38800      G(10)=1.0E-06
38900      H(10)=1.0E+06
39000      G(11)=1.0E-07
39100      H(11)=1.0E+07
39200      IF(N=M) 1,2,1
39300 1      X(I,12)=X(I,7)*X(I,9)/X(I,8)**2
39400      G(12)=1.0
39500      H(12)=10.0
39600      X(I,13)=X(I,8)/X(I,7)
39700      G(13)=0.5
39800      H(13)=10.0
39900      X(I,14)=X(I,9)/X(I,8)
40000      G(14)=1.0
40100      H(14)=10
40200      X(I,15)=X(I,10)/X(I,9)
40300      G(15)=1.0
40400      H(15)=10.0
40500      X(I,16)=X(I,11)/X(I,10)
40600      G(16)=1.0
40700      H(16)=10.0
40800 2      RETURN
40900      END
41000  C
41100  C
41200      SUBROUTINE FUNC (INN,M,K,XDUMY,F,J,DASHI,BRACE1,BRACE2,
41300      1BRACE3,C50,C51,C52,C53,C54,C55,C556,C56,CT1,C7,THETR,DUMMY,XFD)
41400 C*****
41500  C
41600  C
41700  C      THIS SUBROUTINE RETURNS THE VALUE OF THE OBJECTIVE
41800  C      FUNCTION TO BE OPTIMIZED.THE OBJECTIVE FUNCTION IS MADE FROM
41900  C      R.H.S. OF EQUATIONS OF FIRST SUBGROUP OF GROUP 1.
42000 C*****
42100      DOUBLE PRECISION XDUMY(K,M),F(K),XFD(11),Z(6),FFD(6),FOLD(6)
42200      DOUBLE PRECISION X(25),RUMY(6)
42300      DOUBLE PRECISION ONETH,TWOTH,BRACE1,BRACE3,BRACE2,THETR,DUMMY
42400      DOUBLE PRECISION C50,C51,C52,C53,C54,C55,C556,C56,C7,CT1
42500      DOUBLE PRECISION DASHI,RES,FZERO,FFDZER,FIRST,SECOND,THIRD
42600      DOUBLE PRECISION FOURTH,FIFTH,SIX,SEVENT,SUM
42700      NMAX=4
42800  C
42900  C
43000      ONETH=1.0/3.
43100      TWOTH=2.0/3.
43200  C
43300      W1=8.08E-04
43400      W2=47.69
43500      W3=3.807E-02
43600      W4=2.937E+06
43700      W5=7.0998E-01

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43800      W6=0.63
43900      W7=0.1981
44000      FOLD(1)=8.5111E+02
44100      FOLD(2)=7.3E+06
44200      FOLD(3)=9.432E+10
44300      FOLD(4)=1.622E+15
44400      C
44500      DUMMY1=1.0E+06
44600      DUMMY2=1.0E-02
44700      DUMMY3=1.0E-03
44800      DUMMY4=4.2985E+03
44900      C
45000      C      X ARE THE OUTPUT VARIABLES FROM THE REACTOR
45100      C      XFD ARE THE FEED CHARECTARISTIC OF AFORESAID VARIABLES
45200      DO 450 IK=1,INN
45300      KI=IK
45400      X(KI)=XDUMY(J,KI)
45500      450  CONTINUE
45600      C
45700      N=K-7
45800      FZERO=X(7)
45900      FFDZER=XFD(7)
46000      DO 210 INTEMP=1,NMAX
46100      NTEM=INTEMP
46200      Z(NTEM)=X(NTEM+7)
46300      FFD(NTEM)=XFD(NTEM+7)
46400      210  CONTINUE
46500      1    RES=(XFD(1)-X(1))/THETR
46600      FIRST=RES/W1
46700      2    RES=C56*(XFD(2)-X(2))/THETR-X(2)
46800      SECOND=RES/W2
46900      3    RES=(XFD(3)-X(3))/(THETR*DUMMY2)-C53*X(6)*(FZERO/DUMMY3)*DASHI
47000      THIRD=RES/W3
47100      4    RES=C50*X(1)-X(4)-C51-CONFR(TWOTH,Z,FZERO)*(DUMMY4**TWOTH)/
47200      1DUMMY3-C52*X(5)/DUMMY1
47300      FOURTH=RES/W4
47400      5    RES=C54*X(3)/DUMMY2-(X(5)/DUMMY1)-C55*(Z(1)*DUMMY4/DUMMY3)*
47500      1X(6)-C556*X(4)
47600      FIFTH=RES/W5
47700      6    RES=X(6)-0.615
47800      SIX=RES/W6
47900      RES=C7*X(2)*X(4)/(X(4)+(CONFR(TWOTH,Z,FZERO)*(DUMMY4**TWOTH)/
48000      1DUMMY3))+(FFDZER-FZERO)/(THETR*DUMMY3)
48100      C
48200      C*****
48300      C
48400      C      INSERT THIS LINE INSTEAD OF PREVIOUS LINE FOR SECOND APPROACH
48500      C6    RES=1.0-X(6)+DLOG(X(6))+SAI*(1.0-X(6))+C65*X(7)/
48600      C      1(CONFR(ONETH,Z,FZERO)*(DUMMY4*ONETH))
48700      C

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48800 C*****
48900 C
49000 SEVENT=RES/W7
49100 DO 650 N=1,NMAX
49200 IF(N.GT.1) GO TO 207
49300 RES=C7*X(2)*X(4)/(X(4)+(CONFR(TWOTH,Z,FZERO)*(DUMMY4**TWOTH)/
49400 1DUMMY3))+CT1*DASHI*N*X(6)*FZERO/(DUMMY3*(1-X(6)))+
49500 1(FFD(N)-Z(N))*(DUMMY4**N)/(THETR*DUMMY3)
49600 RUMY(N)=RES/FOLD(N)
49700 GO TO 650
49800 207 RES=C7*X(2)*X(4)/(X(4)+(CONFR(TWOTH,Z,FZERO)*(DUMMY4**TWOTH)/
49900 1DUMMY3))+CT1*DASHI*N*X(6)*Z(N-1)*(DUMMY4**(N-1))/(DUMMY3*
50000 1(1-X(6)))+(FFD(N)-Z(N))*(DUMMY4**N)/(THETR*DUMMY3)
50100 RUMY(N)=RES/FOLD(N)
50200 650 CONTINUE
50300 SUM=0.0
50400 DO 550 IK=1,NMAX
50500 KI=IK
50600 SUM=SUM+ABS(RUMY(KI))
50700 550 CONTINUE
50800 F(J)=SUM+ABS(FIRST)+ABS(SECOND)+ABS(THIRD)+ABS(FOURTH)+
50900 1ABS(FIFTH)+ABS(SIX)+ABS(SEVENT)
51000 F(J)=-F(J)
51100 RETURN
51200 END
51300 C
51400 C
51500 SUBROUTINE QG10(XL,XU,Y,AMV)
51600 C*****
51700 C
51800 C THIS SUBROUTINE RETURNS THE VALUE OF INTEGRAL OF A
51900 C FUNCTION EVALUATED IN THE RANGE XL TO XU.IT IS USED FOR
52000 C CALCULATING THE CUMULATIVE DISTRIBUTION FROM A DIFFERENTIAL
52100 C DISTRIBUTION.
52200 C
52300 C*****
52400 DOUBLE PRECISION Y,XL,XU
52500 DOUBLE PRECISION AMV(0:5)
52600 DOUBLE PRECISION A,B,C
52700 A=0.5*(XU-XL)+XL
52800 B=(XU-XL)
52900 C=0.4869533*B
53000 Y=0.03333567*(DIFF2(AMV,A+C)+DIFF2(AMV,A-C))
53100 C=0.4325317*B
53200 Y=Y+0.07472567*(DIFF2(AMV,A+C)+DIFF2(AMV,A-C))
53300 C=0.3397048*B
53400 Y=Y+0.1095432*(DIFF2(AMV,A+C)+DIFF2(AMV,A-C))
53500 C=0.2166977*B
53600 Y=Y+0.1346334*(DIFF2(AMV,A+C)+DIFF2(AMV,A-C))
53700 C=0.07443717*B

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53800      Y=B*(Y+0.1477621*(DIFF2(AMV,A+C)+DIFF2(AMV,A-C)))
53900      RETURN
54000      END
54100      C
54200      C
54300      SUBROUTINE DISTR(XL1,PART1,AMV,IWHICH,XCORD,YCORD,ZCORD,XCUM)
54400      C*****
54500      C
54600      C          THIS SUBROUTINE GENERATES THE DISTRIBUTION FUNCTION FROM
54700      C          ITS MOMENT. INITIALLY THE FUNCTION IS EVALUATED AT STEPS OF
54800      C          PART1 STARTING WITH THE LOWER LIMIT OF XL1. THE USEFUL RANGE OF
54900      C          DISTRIBUTION IS THEN CALCULATED. THE USEFUL RANGE IS THEN
55000      C          DIVIDED INTO 50 INTERVALS AND THE DISTRIBUTION FUNCTION IS
55100      C          CALCULATED AT THESE 51 POINTS. XCORD(I) IS THE SET OF X
55200      C          CO-ORDINATES AT WHICH THE FUNCTION CALCULATED. YCORD(I) IS THE
55300      C          SET OF FUNCTION VALUES AT THESE POINTS ON NUMBER BASIS.
55400      C          ZCORD(I) IS THE SET OF FUNCTION VALUES ON WT. BASIS. XCUM
55500      C          REPRESENTS THE SET OF NORMALISED CUMULATIVE DISTRIBUTION
55600      C          FUNCTION CALCULATED AT THE AFORESAID POINTS
55700      C          IWHICH=0---NORMALISED CUMULATIVE DISTRIBUTION IS NOT CALCULATED
55800      C
55900      C
56000      C*****
56100      DOUBLE PRECISION XCORD(0:50),YCORD(0:50),ZCORD(0:50),XCUM(0:50)
56200      DOUBLE PRECISION STAR(50),AMV(0:5)
56300      DOUBLE PRECISION XL,XL1,XU1,XU,PART1,PART2,Y,YMAX
56400      DOUBLE PRECISION  RATIO,AREA,TOTAL
56500      XL=XL1
56600      C          TO FIND USEFUL RANGE OF DISTRIBUTION.
56700      YMAX=0.0
56800      1      Y=DIFF2(AMV,XL)
56900      IF(Y.GT.YMAX) GO TO 15
57000      RATIO=Y/YMAX
57100      IF(RATIO.GT.0.0001)GO TO 16
57200      GO TO 2
57300      15      YMAX=Y
57400      16      XL=XL+PART1
57500      GO TO 1
57600      C          UPPER USEFUL LIMIT FOR THE DISTRIBUTION HAS BEEN OBTAINED
57700      2      XU1=XL
57800      PART2=(XU1-XL1)/50.0
57900      XL=XL1
58000      C          TO FIND DIFFERENTIAL AND CUMULATIVE DISTRIBUTION
58100      XCORD(0)=XL1
58200      YCORD(0)=DIFF2(AMV,XL1)
58300      ZCORD(0)=XCORD(0)*YCORD(0)
58400      XCUM(0)=0.0
58500      AREA=0.0
58600      DO 3 KI=1,50
58700      XU=XL+PART2

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58800      XCORD(KI)=XU
58900      YCORD(KI)=DIFF2(AMV,XU)
59000      ZCORD(KI)=XCORD(KI)*YCORD(KI)
59100      IF(IWHICH.EQ.0) GO TO 6
59200      CALL QG10(XL,XU,Y,AMV)
59300      STAR(KI)=Y
59400      AREA=AREA+STAR(KI)
59500      6      XL=XU
59600      3      CONTINUE
59700      IF(IWHICH.EQ.0) GO TO 5
59800      C      TO CALCULATE NORMALISED CUMULATIVE DISTRIBUTION
59900      TOTAL=AREA
60000      AREA=0.0
60100      DO 4 KI=1,50
60200      AREA=AREA+STAR(KI)
60300      XCUM(KI)=AREA/TOTAL
60400      4      CONTINUE
60500      5      RETURN
60600      END
60700      C
60800      C
60900      SUBROUTINE FUNXX(X,RES,K,DASHI,
61000      1ONETH,TWOTH,BRACE1,BRACE2,BRACE3,C50,C51,C52,C53,C54,C55,C556,
61100      1C56,C7,CT1,C65,DUMMY,THETR,SAI,XFD,X66)
61200      C*****
61300      C      THIS SUBROUTINE IS SIMILAR TO FUNX.FOR SECOND APPROACH
61400      C      CHANGE THE NAME OF SUBROUTINE TO FUNX.
61500      C*****
61600      DOUBLE PRECISION X(13),XFD(13),F(6),FFD(6),FOLD(6)
61700      DOUBLE PRECISION RES,DASHI,ONETH,TWOTH
61800      DOUBLE PRECISION BRACE1,BRACE2,BRACE3,DUMMY,THETR,SAI
61900      DOUBLE PRECISION C50,C51,C52,C53,C54,C55,C556,C56,C7,CT1,C65
62000      DOUBLE PRECISION FZERO,FFDZER,X66
62100      C      COMMON ONETH,TWOTH,BRACE1,BRACE2,BRACE3,C50,C51,C52,C53,C54,C55,
62200      C      1C556,C56,CT1,C7,THETR,DUMMY
62300      NMAX=4
62400      C
62500      C
62600      C
62700      W1=8.08E-04
62800      W2=47.69
62900      W3=3.807E-02
63000      W4=2.937E+06
63100      W5=7.0998E-01
63200      W6=0.63
63300      W7=0.1981
63400      FOLD(1)=8.5111E+02
63500      FOLD(2)=7.3E+06
63600      FOLD(3)=9.432E+10
63700      FOLD(4)=1.622E+15

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63800 C
63900 DUMMY1=1.0E+06
64000 DUMMY2=1.0E-02
64100 DUMMY3=1.0E-03
64200 DUMMY4=4.2985E+03
64300 C
64400 C X ARE THE OUTPUT VARIABLES FROM THE REACTOR
64500 C XFD ARE THE FEED CHARECTARISTIC OF AFORESAID VARIABLES
64600 N=K-7
64700 FZERO=X(6)
64800 FFDZER=XFD(6)
64900 DO 210 INTEMP=1,NMAX
65000 NTEM=INTEMP
65100 F(NTEM)=X(NTEM+6)
65200 FFD(NTEM)=XFD(NTEM+6)
65300 210 CONTINUE
65400 IF(K.GT.6) GO TO 205
65500 GO TO (1,2,3,4,5,6),K
65600 1 CONH=CONFR(TWOTH,F,FZERO)
65700 RES=X(11)-CONH
65800 RETURN
65900 2 RES=(XFD(1)-X(1))/THETR
66000 RES=RES/W1
66100 RETURN
66200 3 RES=C56*(XFD(2)-X(2))/THETR-X(2)
66300 RES=RES/W2
66400 RETURN
66500 4 RES=(XFD(3)-X(3))/(THETR*DUMMY2)-C53*X66*(FZERO/DUMMY3)*DASHI
66600 RES=RES/W3
66700 RETURN
66800 5 RES=C50*X(1)-X(4)-C51-X(11)*(DUMMY4**TWOTH)/
66900 1DUMMY3-C52*X(5)
67000 RES=RES/W4
67100 RETURN
67200 6 RES=C54*X(3)/DUMMY2-(X(5))-C55*(F(1)*DUMMY4/DUMMY3)*
67300 1X66-C556*X(4)
67400 RES=RES/W5
67500 RETURN
67600 205 IF (K.GT.7) GO TO 206
67700 RES=C7*X(2)*X(4)/(X(4)+(X(11)*(DUMMY4**TWOTH)/
67800 1DUMMY3))+(FFDZER-FZERO)/(THETR*DUMMY3)
67900 CCC RES=RES/W7
68000 RETURN
68100 206 IF(K.GT.8)GO TO 207
68200 RES=C7*X(2)*X(4)/(X(4)+(X(11)*(DUMMY4**TWOTH)/
68300 1DUMMY3))+CT1*DASHI*N*X66*FZERO/(DUMMY3*(1-X66))+
68400 1(FFD(N)-F(N))*(DUMMY4**N)/(THETR*DUMMY3)
68500 CCC RES=RES/FOLD(N)
68600 RETURN
68700 207 RES=C7*X(2)*X(4)/(X(4)+(X(11)*(DUMMY4**TWOTH)/

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68800      1DUMMY3))+CT1*DASHI*N*X66*F(N-1)*(DUMMY4**(N-1))/(DUMMY3*
68900      1(1-X66)))+(FFD(N)-F(N))*(DUMMY4**N)/(THETR*DUMMY3)
69000  CCC  RES=RES/FOLD(N)
69100      RETURN
69200      END
69300  C
69400  C
69500      SUBROUTINE GOLD(X01,X02,C,XMAX,D,X7,SAI,C65,DUMMY4,ONETH)
69600  C*****
69700  C
69800  C      THIS SUBROUTINE SOLVES EQN.4,21 BY GOLDEN SECTION METHOD
69900  C      FOR OPTIMIZATION.
70000  C*****
70100      DOUBLE PRECISION X01,X02,C,XMAX,D,X7,SAI,C65,C1,ONETH
70200      DOUBLE PRECISION X3,X4,TEMPX2,Y3,Y4,X1,X2
70300      C1=C/10.0
70400      IF(ABS(X01-X02).LT.C) GO TO 100
70500      X1=X01
70600      X2= X02
70700      X3=X1+0.3820*(X2-X1)
70800      X4=X2-0.3820*(X2-X1)
70900      CALL OBJECT(Y3,X3,D,X7,SAI,C65,DUMMY4,ONETH)
71000      CALL OBJECT(Y4,X4,D,X7,SAI,C65,DUMMY4,ONETH)
71100  50    IF(Y3.GE.Y4) GO TO 30
71200      X1=X3
71300      IF(ABS(X1-X2).LT.C) GO TO 100
71400      TEMPX2=X1+0.3820*(X2-X1)
71500      IF(ABS(TEMPX2-X4).GT. C1) GO TO 40
71600      Y3=Y4
71700      GO TO 45
71800  40    X3=TEMPX2
71900      CALL OBJECT(Y3,X3,D,X7,SAI,C65,DUMMY4,ONETH)
72000  45    X4=X2-0.3820*(X2-X1)
72100      CALL OBJECT(Y4,X4,D,X7,SAI,C65,DUMMY4,ONETH)
72200      GO TO 50
72300
72400  30    X2=X4
72500      IF(ABS(X1-X2).LT.C) GO TO 100
72600      TEMPX2=X2-0.3820*(X2-X1)
72700      IF(ABS(TEMPX2-X3).GT.C1) GO TO 60
72800      Y4=Y3
72900      GO TO 65
73000  60    X4= TEMPX2
73100      CALL OBJECT(Y4,X4,D,X7,SAI,C65,DUMMY4,ONETH)
73200  65    X3=X1+0.3820 *(X2-X1)
73300      CALL OBJECT(Y3,X3,D,X7,SAI,C65,DUMMY4,ONETH)
73400      GO TO 50
73500  100   XMAX=(X1+X2)/2.0
73600      CALL OBJECT(YMAX,XMAX,D,X7,SAI,C65,DUMMY4,ONETH)
73700      YMAXX=YMAX

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73800      RETURN
73900      END
74000      C
74100      C
74200      C
74300      C
74400      SUBROUTINE OBJECT(YOPT,XOPT,D,X7,SAI,C65,DUMMY4,ONETH)
74500      C*****
74600      C
74700      C      THIS SUBROUTINE REPRESENTS THE OBJECTIVE FUNCTION TO BE
74800      C      OPTIMIZED FOR THE SOLUTION OF EQN. 4.21.
74900      C
75000      C*****
75100      C
75200      DOUBLE PRECISION YOPT,XOPT,D,X7,SAI,C65,ONETH
75300      YOPT=1.0-XOPT+DLOG(XOPT)+SAI*(1.0-XOPT)*(1.0-XOPT)+C65*X7/
75400      1(D*(DUMMY4**ONETH))
75500      YOPT=-ABS(YOPT)
75600      RETURN
75700      END

```

DATA FOR THE PROBLEM

DATA FOR CONSTRAINED OPTIMIZATION_____

NOPT = 11 M = 16 K = 14 ITMAX = 2000 IPRIN = 0
 ALPHA = 1.300000 BETA = 0.000100 DELTA = 0.001000
 GAMMA = 5 DASHI = 0.900000

INITIAL GUESSES_____

XSTART(1) = 0.40000000E+01 XSTART(2) = 0.99000000E+00
 XSTART(3) = 0.16300000E+01 XSTART(4) = 0.99000000E+00
 XSTART(5) = 0.99000000E+00 XSTART(6) = 0.64000000E+00
 XSTART(7) = 0.99000000E+00 XSTART(8) = 0.14000000E+01
 XSTART(9) = 0.30000000E+01 XSTART(10) = 0.60000000E+01
 XSTART(11) = 0.24000000E+02 XSTART(

INPUT PARAMETERS FOR LINEAR EQUATIONS AND THE CONVERGENCE CRITERION

MAXIT = 200 NUMSIG = 6 IPT = 1
 NUMEO = 6

EPS = 0.10000000E-07 DELNON = 0.10000000E-06 PREC = 0.50000000E-05
 ZETA = 0.10000000E-08

RATE CONSTANTS_____

KP = 0.26500000E+06 KD = 0.23300000E-04 KZERO = 0.00000000E+00
 KIP = 0.00000000E+00 KTD = 0.00000000E+00 KTC = 0.72000000E+10
 KFT = 0.00000000E+00 KFM = 0.21200000E+02

MONOMER-POLYMER CHARECTARISTICS____

MW = 0.10400000E+03 DP = 0.10526000E+01 DM = 0.87800000E+00
 VMOL = 0.11845000E+03 SAI = 0.25600000E+00

REACTOR CHARECTARISTICS____

THETA = 0.90000000E+03 TEMP = 0.34300000E+03 VW = 0.64940000E+00

OTHER CONSTANTS____

A1OW = 0.15400000E-04 AMZERO = 0.16611000E-12 VD = 0.41888000E-11
 VM = 0.65449800E-19 CE = 0.36720000E+10 AMM = 0.63000000E-02
 FEF = 0.10000000E+01 SWC = 0.89000000E-05 YETA = 0.75000000E+01

FEED CONDITIONS____

RADICAL NUMBER DISTRIBUTION_____

FFD(1) = 0.000000 FFD(2) = 0.000000
 FFD(3) = 0.000000 FFD(4) = 0.000000
 FFD(5) = 0.000000 FFD(6) = 0.000000
 FFD(7) = 0.000000 FFD(8) = 0.000000

LIVE POLYMER CHAINS-----

FDLIV(0, 1) = 0.0000	FDLIV(0, 2) = 0.0000	FDLIV(0, 3) = 0.0000
FDLIV(0, 4) = 0.0000	FDLIV(0, 5) = 0.0000	FDLIV(0, 6) = 0.0000
FDLIV(0, 7) = 0.0000	FDLIV(0, 8) = 0.0000	FDLIV(0, 9) = 0.0000
FDLIV(0,10) = 0.0000	FDLIV(
FDLIV(1, 1) = 0.0000	FDLIV(1, 2) = 0.0000	FDLIV(1, 3) = 0.0000
FDLIV(1, 4) = 0.0000	FDLIV(1, 5) = 0.0000	FDLIV(1, 6) = 0.0000
FDLIV(1, 7) = 0.0000	FDLIV(1, 8) = 0.0000	FDLIV(1, 9) = 0.0000
FDLIV(1,10) = 0.0000	FDLIV(
FDLIV(2, 1) = 0.0000	FDLIV(2, 2) = 0.0000	FDLIV(2, 3) = 0.0000
FDLIV(2, 4) = 0.0000	FDLIV(2, 5) = 0.0000	FDLIV(2, 6) = 0.0000
FDLIV(2, 7) = 0.0000	FDLIV(2, 8) = 0.0000	FDLIV(2, 9) = 0.0000
FDLIV(2,10) = 0.0000	FDLIV(
FDLIV(3, 1) = 0.0000	FDLIV(3, 2) = 0.0000	FDLIV(3, 3) = 0.0000
FDLIV(3, 4) = 0.0000	FDLIV(3, 5) = 0.0000	FDLIV(3, 6) = 0.0000
FDLIV(3, 7) = 0.0000	FDLIV(3, 8) = 0.0000	FDLIV(3, 9) = 0.0000
FDLIV(3,10) = 0.0000	FDLIV(
FDLIV(4, 1) = 0.0000	FDLIV(4, 2) = 0.0000	FDLIV(4, 3) = 0.0000
FDLIV(4, 4) = 0.0000	FDLIV(4, 5) = 0.0000	FDLIV(4, 6) = 0.0000
FDLIV(4, 7) = 0.0000	FDLIV(4, 8) = 0.0000	FDLIV(4, 9) = 0.0000
FDLIV(4,10) = 0.0000	FDLIV(
FDLIV(5, 1) = 0.0000	FDLIV(5, 2) = 0.0000	FDLIV(5, 3) = 0.0000
FDLIV(5, 4) = 0.0000	FDLIV(5, 5) = 0.0000	FDLIV(5, 6) = 0.0000
FDLIV(5, 7) = 0.0000	FDLIV(5, 8) = 0.0000	FDLIV(5, 9) = 0.0000
FDLIV(5,10) = 0.0000	FDLIV(

DEAD POLYMER CHAINS-----

FDDED(0, 1) = 0.0000	FDDED(0, 2) = 0.0000	FDDED(0, 3) = 0.0000
FDDED(0, 4) = 0.0000	FDDED(0, 5) = 0.0000	FDDED(0, 6) = 0.0000
FDDED(0, 7) = 0.0000	FDDED(0, 8) = 0.0000	FDDED(0, 9) = 0.0000
FDDED(0,10) = 0.0000	FDDED(
FDDED(1, 1) = 0.0000	FDDED(1, 2) = 0.0000	FDDED(1, 3) = 0.0000
FDDED(1, 4) = 0.0000	FDDED(1, 5) = 0.0000	FDDED(1, 6) = 0.0000
FDDED(1, 7) = 0.0000	FDDED(1, 8) = 0.0000	FDDED(1, 9) = 0.0000
FDDED(1,10) = 0.0000	FDDED(
FDDED(2, 1) = 0.0000	FDDED(2, 2) = 0.0000	FDDED(2, 3) = 0.0000
FDDED(2, 4) = 0.0000	FDDED(2, 5) = 0.0000	FDDED(2, 6) = 0.0000
FDDED(2, 7) = 0.0000	FDDED(2, 8) = 0.0000	FDDED(2, 9) = 0.0000
FDDED(2,10) = 0.0000	FDDED(
FDDED(3, 1) = 0.0000	FDDED(3, 2) = 0.0000	FDDED(3, 3) = 0.0000
FDDED(3, 4) = 0.0000	FDDED(3, 5) = 0.0000	FDDED(3, 6) = 0.0000
FDDED(3, 7) = 0.0000	FDDED(3, 8) = 0.0000	FDDED(3, 9) = 0.0000
FDDED(3,10) = 0.0000	FDDED(
FDDED(4, 1) = 0.0000	FDDED(4, 2) = 0.0000	FDDED(4, 3) = 0.0000
FDDED(4, 4) = 0.0000	FDDED(4, 5) = 0.0000	FDDED(4, 6) = 0.0000
FDDED(4, 7) = 0.0000	FDDED(4, 8) = 0.0000	FDDED(4, 9) = 0.0000
FDDED(4,10) = 0.0000	FDDED(
FDDED(5, 1) = 0.0000	FDDED(5, 2) = 0.0000	FDDED(5, 3) = 0.0000

FDDED(5, 4) = 0.0000 FDDDED(5, 5) = 0.0000 FDDDED(5, 6) = 0.0000
FDDED(5, 7) = 0.0000 FDDDED(5, 8) = 0.0000 FDDDED(5, 9) = 0.0000
FDDED(5, 10) = 0.0000 FDDDED(

FEED COND. FOR VARIABLES OF INITIAL SET

1	0.4080E+01	2	0.1000E+01	3	0.1920E+01
4	0.0000E+00	5	0.0000E+00	6	0.0000E+00
7	0.0000E+00	8	0.0000E+00	9	0.0000E+00
10	0.0000E+00	11	0.0000E+00		

OUTPUT FROM THE PROGRAM_____

(1)	NUMBER OF PARTICLES (/CC EMULSION)_____	0.11466412E+16
(2)	AVERAGE VOLUME OF PARTICLES (CC)_____	0.26767643E-15
(3)	AVERAGE DIA ; VOLUME BASIS ("A")_____	0.79956124E+03
(4)	AVERAGE DIA ; NUMBER BASIS ("A")_____	0.71403866E+03
(5)	CONVERSION OBTAINED_____	0.40438988E+00
(6)	RATE (MOLES/(LIT.HOUR))_____	0.47828000E+01
(7)	NUMBER AVERAGE MOLECULAR WEIGHT_____	0.39254813E+06
(8)	WEIGHT AVERAGE MOLECULAR WEIGHT_____	0.92157680E+06
(9)	POLYDISPERSITY INDEX_____	0.23476785E+01

DETAILS OF THE DISTRIBUTIONS FOLLOW-----

LIVE POLYMER CHAIN LENGTH DISTRIBUTION

NO	MOL. WT.	NUMB.BASIS	WT.BASIS(NORMAL)
0	0.1000E+01	0.1134E-03	0.6037E-07
1	0.7010E+03	0.1116E-03	0.4164E-04
2	0.1401E+04	0.9412E-04	0.7020E-04
3	0.2101E+04	0.7868E-04	0.8801E-04
4	0.2801E+04	0.6548E-04	0.9765E-04
5	0.3501E+04	0.5434E-04	0.1013E-03
6	0.4201E+04	0.4501E-04	0.1007E-03
7	0.4901E+04	0.3722E-04	0.9712E-04
8	0.5601E+04	0.3074E-04	0.9168E-04
9	0.6301E+04	0.2537E-04	0.8511E-04
10	0.7001E+04	0.2092E-04	0.7799E-04
11	0.7701E+04	0.1725E-04	0.7071E-04
12	0.8401E+04	0.1421E-04	0.6356E-04
13	0.9101E+04	0.1171E-04	0.5673E-04
14	0.9801E+04	0.9646E-05	0.5033E-04
15	0.1050E+05	0.7947E-05	0.4443E-04
16	0.1120E+05	0.6549E-05	0.3906E-04
17	0.1190E+05	0.5399E-05	0.3421E-04
18	0.1260E+05	0.4452E-05	0.2987E-04
19	0.1330E+05	0.3673E-05	0.2601E-04
20	0.1400E+05	0.3032E-05	0.2260E-04
21	0.1470E+05	0.2504E-05	0.1960E-04
22	0.1540E+05	0.2069E-05	0.1697E-04
23	0.1610E+05	0.1711E-05	0.1467E-04
24	0.1680E+05	0.1416E-05	0.1266E-04
25	0.1750E+05	0.1172E-05	0.1092E-04
26	0.1820E+05	0.9712E-06	0.9412E-05
27	0.1890E+05	0.8052E-06	0.8103E-05
28	0.1960E+05	0.6680E-06	0.6971E-05
29	0.2030E+05	0.5545E-06	0.5993E-05
30	0.2100E+05	0.4605E-06	0.5148E-05
31	0.2170E+05	0.3826E-06	0.4420E-05
32	0.2240E+05	0.3180E-06	0.3793E-05
33	0.2310E+05	0.2644E-06	0.3252E-05
34	0.2380E+05	0.2199E-06	0.2787E-05
35	0.2450E+05	0.1829E-06	0.2386E-05
36	0.2520E+05	0.1522E-06	0.2042E-05
37	0.2590E+05	0.1266E-06	0.1746E-05
38	0.2660E+05	0.1053E-06	0.1491E-05
39	0.2730E+05	0.8758E-07	0.1273E-05
40	0.2800E+05	0.7281E-07	0.1085E-05
41	0.2870E+05	0.6050E-07	0.9245E-06
42	0.2940E+05	0.5025E-07	0.7865E-06
43	0.3010E+05	0.4170E-07	0.6683E-06
44	0.3080E+05	0.3458E-07	0.5671E-06
45	0.3150E+05	0.2865E-07	0.4805E-06
46	0.3220E+05	0.2371E-07	0.4064E-06
47	0.3290E+05	0.1959E-07	0.3432E-06
48	0.3360E+05	0.1617E-07	0.2893E-06
49	0.3430E+05	0.1332E-07	0.2433E-06
50	0.3500E+05	0.1096E-07	0.2042E-06

DEAD POLYMER CHAIN LENGTH DISTRIBUTION

NO	MOL.WT.	NUMB.BASIS	WT.BASIS(NORMAL)
0	0.1000E+01	0.4579E+00	0.4217E-06
1	0.5890E+03	0.7485E-01	0.4060E-04
2	0.1177E+04	0.5342E-01	0.5790E-04
3	0.1765E+04	0.4136E-01	0.6723E-04
4	0.2353E+04	0.3326E-01	0.7206E-04
5	0.2941E+04	0.2739E-01	0.7417E-04
6	0.3529E+04	0.2295E-01	0.7460E-04
7	0.4117E+04	0.1951E-01	0.7396E-04
8	0.4705E+04	0.1677E-01	0.7267E-04
9	0.5293E+04	0.1456E-01	0.7095E-04
10	0.5881E+04	0.1273E-01	0.6896E-04
11	0.6469E+04	0.1121E-01	0.6680E-04
12	0.7057E+04	0.9928E-02	0.6452E-04
13	0.7645E+04	0.8828E-02	0.6215E-04
14	0.8233E+04	0.7877E-02	0.5972E-04
15	0.8821E+04	0.7046E-02	0.5723E-04
16	0.9409E+04	0.6315E-02	0.5471E-04
17	0.9997E+04	0.5666E-02	0.5216E-04
18	0.1059E+05	0.5087E-02	0.4959E-04
19	0.1117E+05	0.4569E-02	0.4701E-04
20	0.1176E+05	0.4102E-02	0.4443E-04
21	0.1235E+05	0.3682E-02	0.4187E-04
22	0.1294E+05	0.3301E-02	0.3933E-04
23	0.1352E+05	0.2957E-02	0.3682E-04
24	0.1411E+05	0.2645E-02	0.3437E-04
25	0.1470E+05	0.2362E-02	0.3198E-04
26	0.1529E+05	0.2106E-02	0.2965E-04
27	0.1588E+05	0.1874E-02	0.2740E-04
28	0.1647E+05	0.1664E-02	0.2523E-04
29	0.1705E+05	0.1475E-02	0.2316E-04
30	0.1764E+05	0.1304E-02	0.2118E-04
31	0.1823E+05	0.1150E-02	0.1930E-04
32	0.1882E+05	0.1011E-02	0.1752E-04
33	0.1941E+05	0.8870E-03	0.1585E-04
34	0.1999E+05	0.7757E-03	0.1428E-04
35	0.2058E+05	0.6762E-03	0.1282E-04
36	0.2117E+05	0.5875E-03	0.1145E-04
37	0.2176E+05	0.5086E-03	0.1019E-04
38	0.2234E+05	0.4385E-03	0.9024E-05
39	0.2293E+05	0.3766E-03	0.7952E-05
40	0.2352E+05	0.3218E-03	0.6971E-05
41	0.2411E+05	0.2737E-03	0.6076E-05
42	0.2470E+05	0.2314E-03	0.5263E-05
43	0.2529E+05	0.1944E-03	0.4527E-05
44	0.2587E+05	0.1622E-03	0.3865E-05
45	0.2646E+05	0.1342E-03	0.3270E-05
46	0.2705E+05	0.1100E-03	0.2739E-05
47	0.2764E+05	0.8908E-04	0.2267E-05
48	0.2823E+05	0.7116E-04	0.1850E-05
49	0.2881E+05	0.5586E-04	0.1482E-05
50	0.2940E+05	0.4287E-04	0.1161E-05

PARTICLE SIZE DISTRIBUTION [CUMULATIVE]

NO	DIA("A)	FRACTION
0	0.5000E+02	0.0000E+00
1	0.4553E+03	0.1683E+00
2	0.5735E+03	0.3083E+00
3	0.6564E+03	0.4248E+00
4	0.7225E+03	0.5216E+00
5	0.7782E+03	0.6022E+00
6	0.8270E+03	0.6692E+00
7	0.8706E+03	0.7249E+00
8	0.9102E+03	0.7713E+00
9	0.9466E+03	0.8098E+00
10	0.9805E+03	0.8419E+00
11	0.1012E+04	0.8685E+00
12	0.1042E+04	0.8907E+00
13	0.1070E+04	0.9091E+00
14	0.1097E+04	0.9244E+00
15	0.1122E+04	0.9372E+00
16	0.1147E+04	0.9478E+00
17	0.1170E+04	0.9566E+00
18	0.1193E+04	0.9639E+00
19	0.1214E+04	0.9700E+00
20	0.1235E+04	0.9751E+00
21	0.1256E+04	0.9793E+00
22	0.1275E+04	0.9828E+00
23	0.1294E+04	0.9857E+00
24	0.1313E+04	0.9881E+00
25	0.1331E+04	0.9901E+00
26	0.1348E+04	0.9918E+00
27	0.1365E+04	0.9932E+00
28	0.1382E+04	0.9944E+00
29	0.1398E+04	0.9953E+00
30	0.1414E+04	0.9961E+00
31	0.1430E+04	0.9968E+00
32	0.1445E+04	0.9974E+00
33	0.1460E+04	0.9978E+00
34	0.1474E+04	0.9982E+00
35	0.1489E+04	0.9985E+00
36	0.1503E+04	0.9988E+00
37	0.1516E+04	0.9990E+00
38	0.1530E+04	0.9992E+00
39	0.1543E+04	0.9993E+00
40	0.1556E+04	0.9995E+00
41	0.1569E+04	0.9996E+00
42	0.1582E+04	0.9997E+00
43	0.1594E+04	0.9997E+00
44	0.1607E+04	0.9998E+00
45	0.1619E+04	0.9998E+00
46	0.1631E+04	0.9999E+00
47	0.1642E+04	0.9999E+00
48	0.1654E+04	0.1000E+01
49	0.1665E+04	0.1000E+01
50	0.1676E+04	0.1000E+01

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